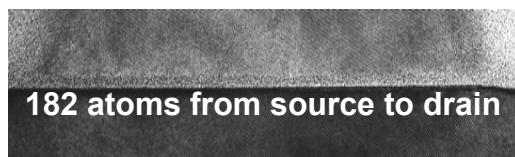
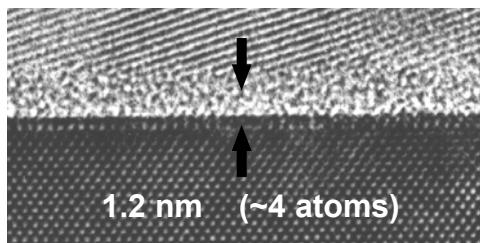
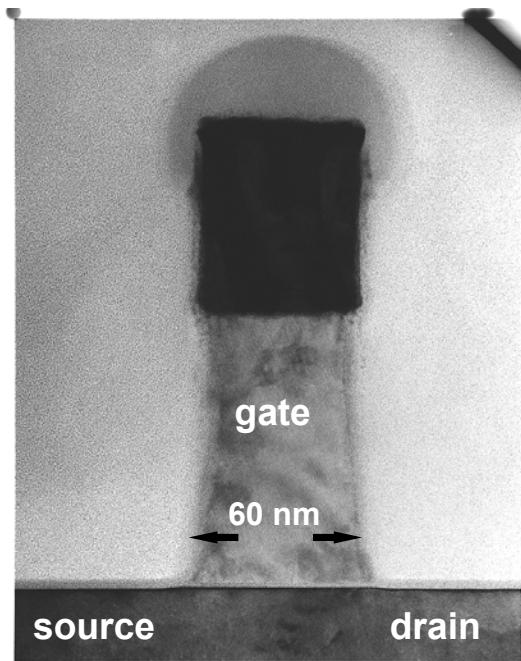


Marrying Reaction Chemistry to Surfaces

Colin Nuckolls
Department of Chemistry
and
The Nanoscience Center
Columbia University

NNI-Washington
June 17, 2005

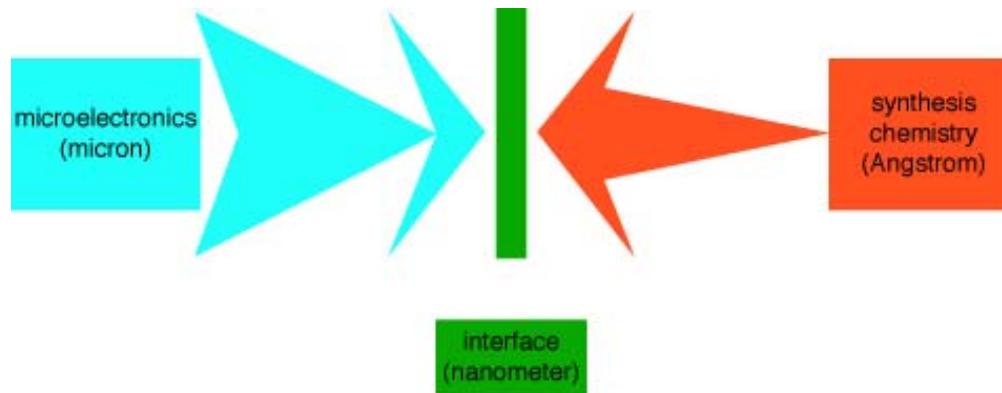
Electronics: Smaller, faster, and cheaper



A real Field Effect Transistor

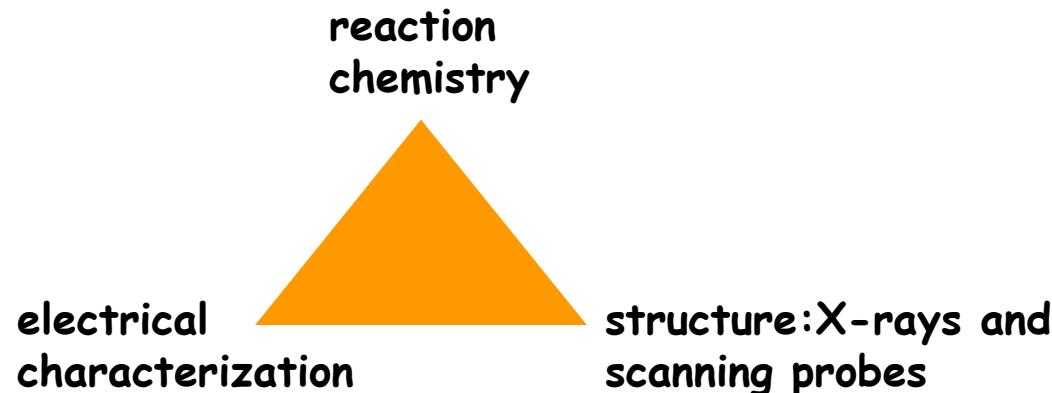
Next Generation of Devices will be a Network of Interfaces

The Nexus of Fabrication and Synthesis is Interfaces

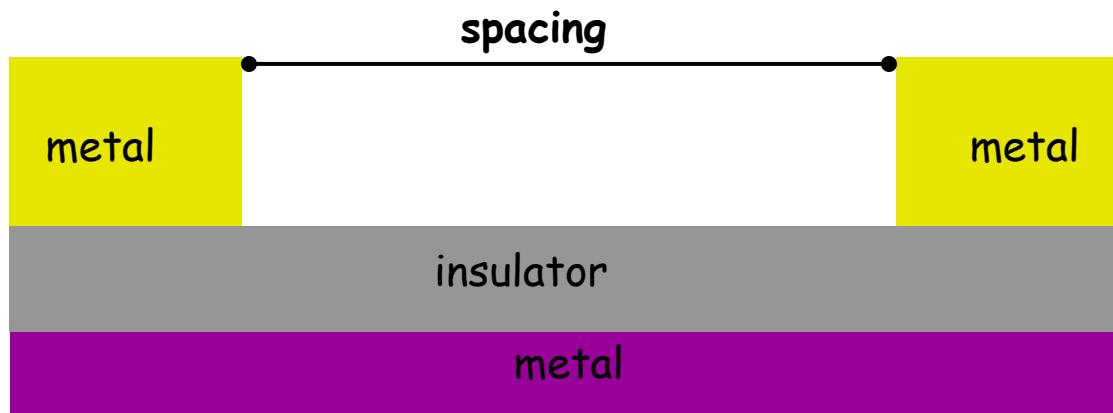


What does reaction chemistry bring to electronics?

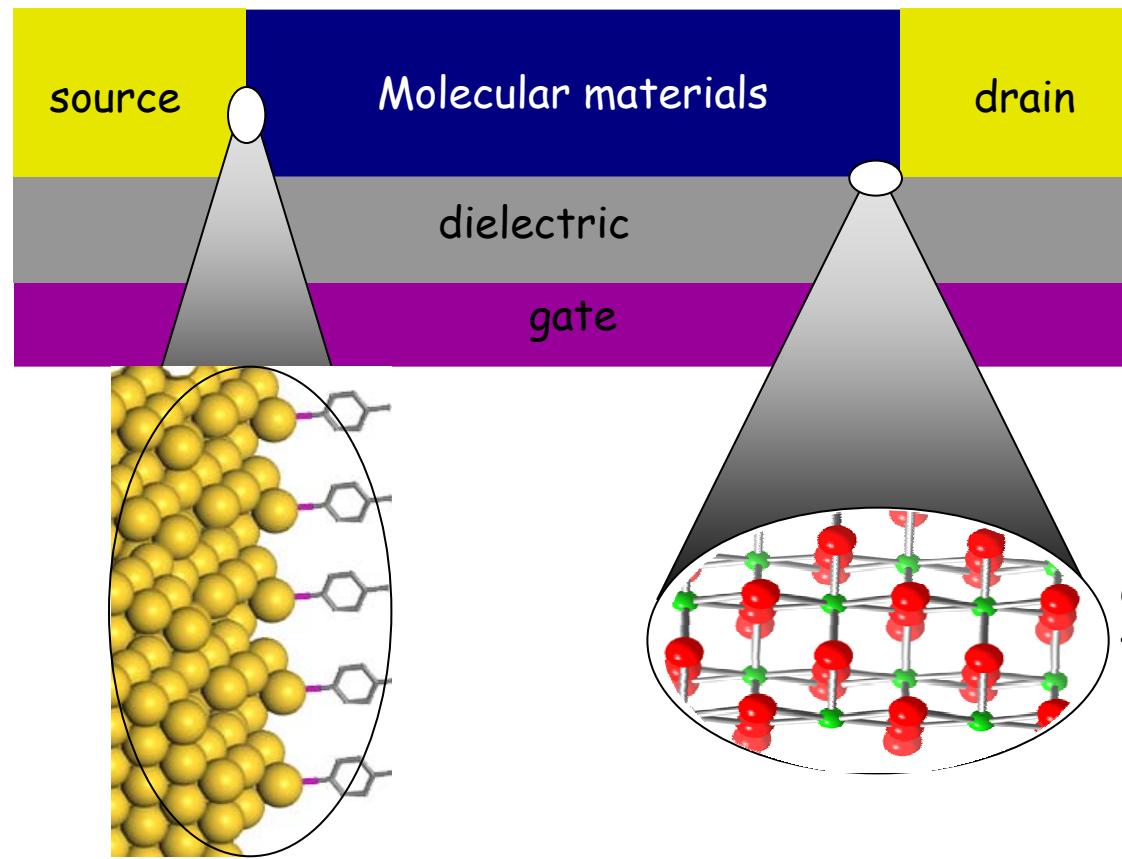
- Self-assembly: programmed complexity
- Recognition
- Programmed reactivity (growth and in-plane):
- Functionality



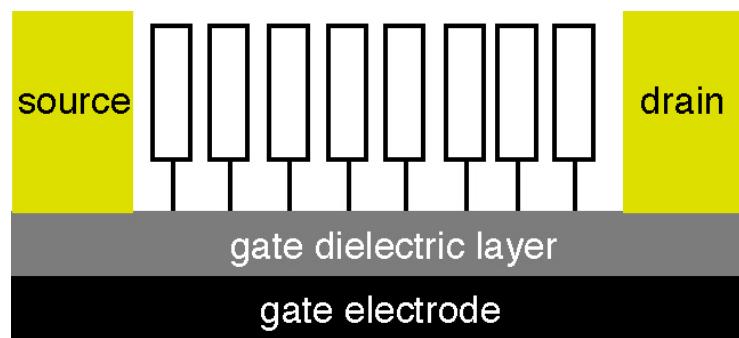
Integrating Reaction Chemistry with Devices



A diverse set of materials available: metals, semiconductors, and oxides
A range of sizes: atoms to centimeters
A sensitive reporter for assembly: *nothing more sensitive*
A sensitive reporter for chemical reactions
Technology: thin film, molecular, and hybrid devices

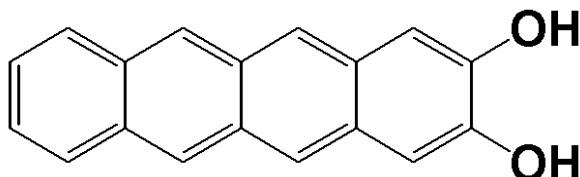


Strategy



Covalently Attached Monolayers

Self-Assembly on High K-Dielectrics

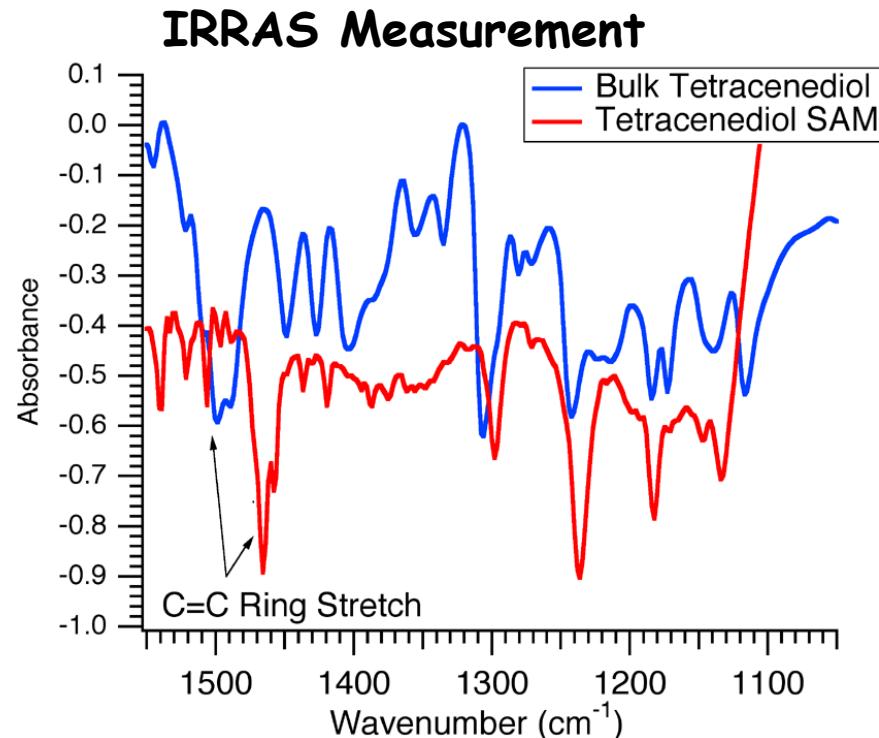


THF solution with
 Al_2O_3 Surface

Water Contact Angle:
Advancing = 76°
Receding = 63°

XPS

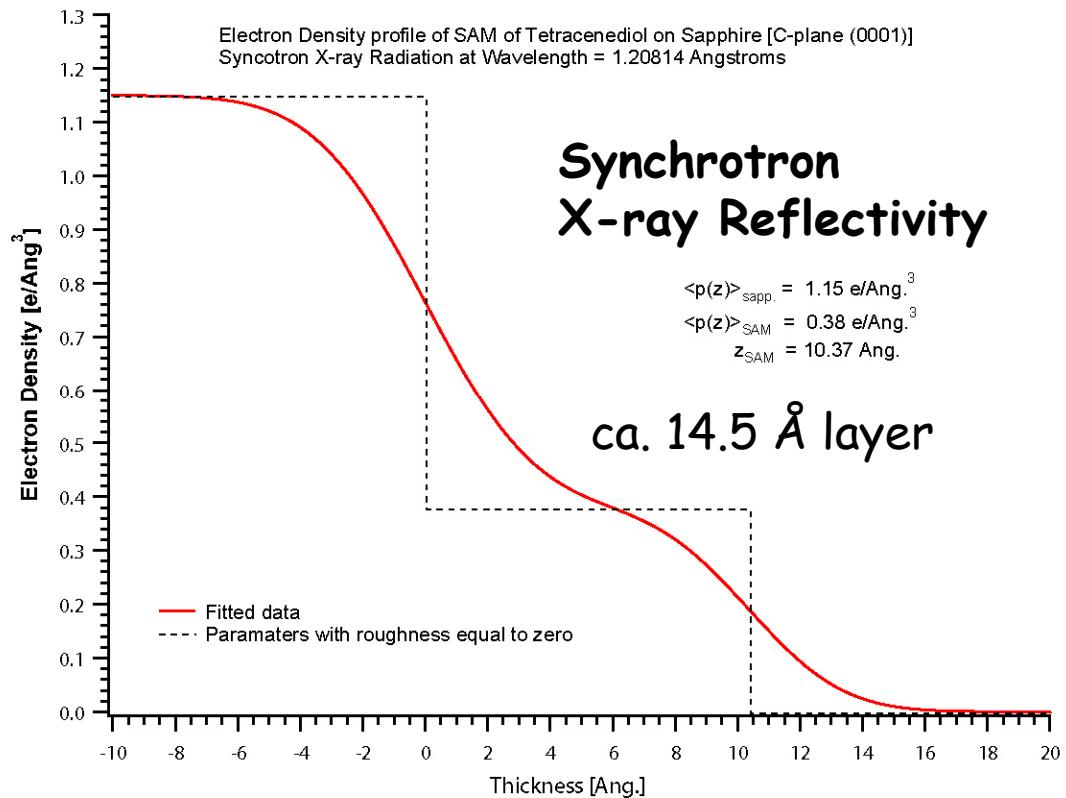
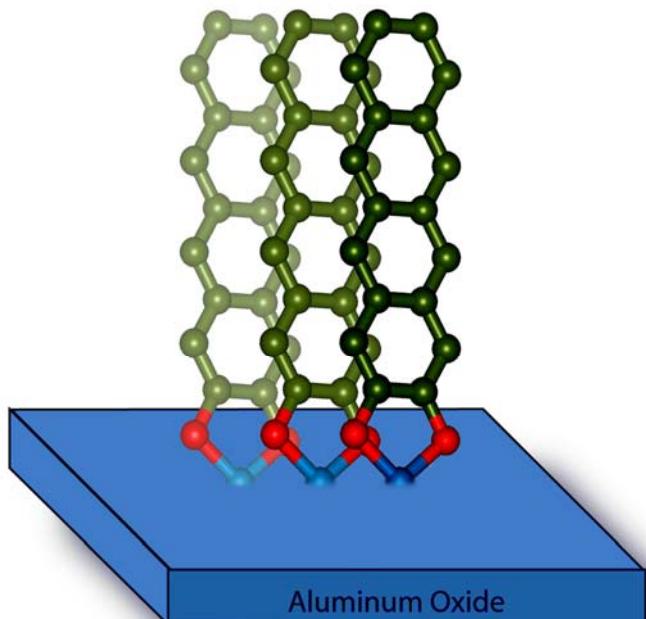
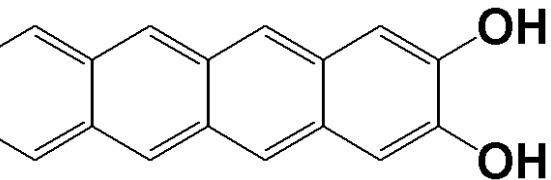
	15°	35°	70°
O	31.4	36.4	41.2
C	47.1	36.3	26.2
Al	21.4	27.2	32.5
C/Al	2.20	1.34	0.81



Ellipsometry:
Measured Thickness = 14.8 \AA
Calculated: 14 \AA

Also: HfO_2 , ZrO_2 , and Y_2O_3

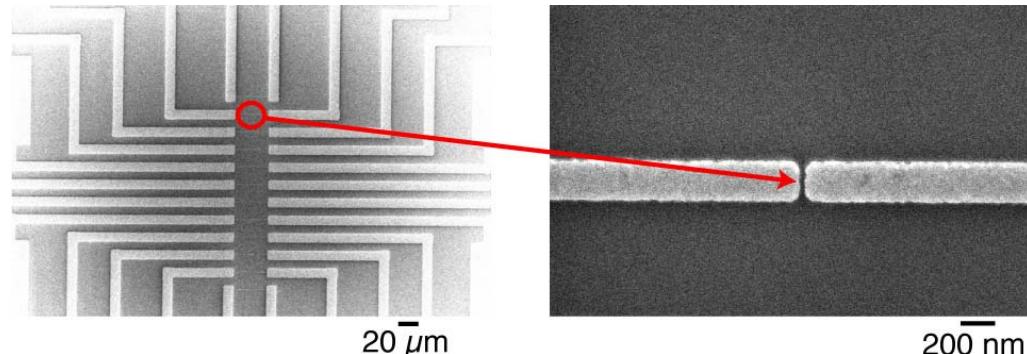
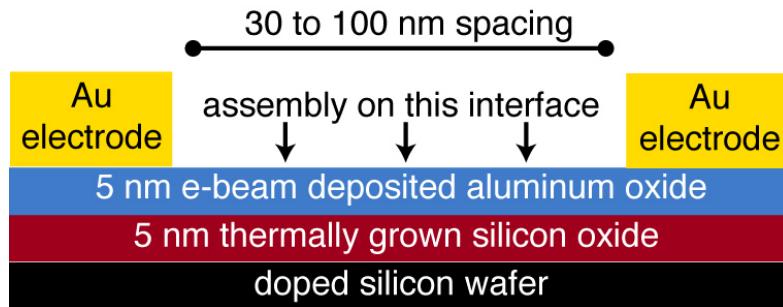
Monolayers on sapphire crystals



electron density
0.38 $\text{e}^-/\text{\AA}^3$ for monolayer
0.39 $\text{e}^-/\text{\AA}^3$ for tetracene crystal

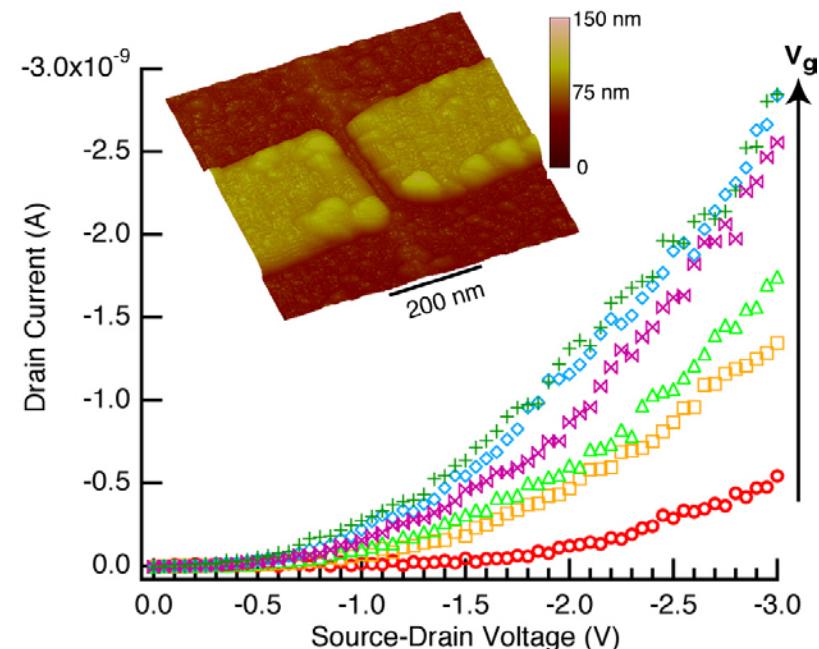
Columbia: George Tulevski, Qian Miao, BNL: Masafumi Fukuto, Ben Ocko,
Ron Pindak, IBM: Cherie Kagan

Monolayer devices



500 sets on an 8" wafer

0.5 V steps



40 nm device ~40,000 molecules

George Tulevski, Qian Miao, Cherie Kagan (IBM)

Yield: 42/80 devices \leq 60 nm source-drain
Much lower $>$ 60 nm
Grains of Al_2O_3 are ca. 40 nm in diameter
Control experiments: catechol and naphthalene

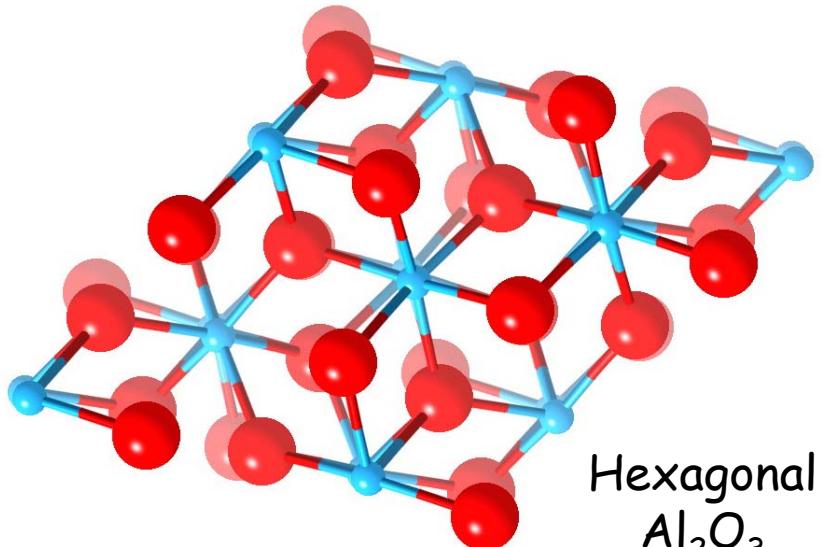
The next step: tune the three interfaces



Source/Drain:
Metal
Derivatized

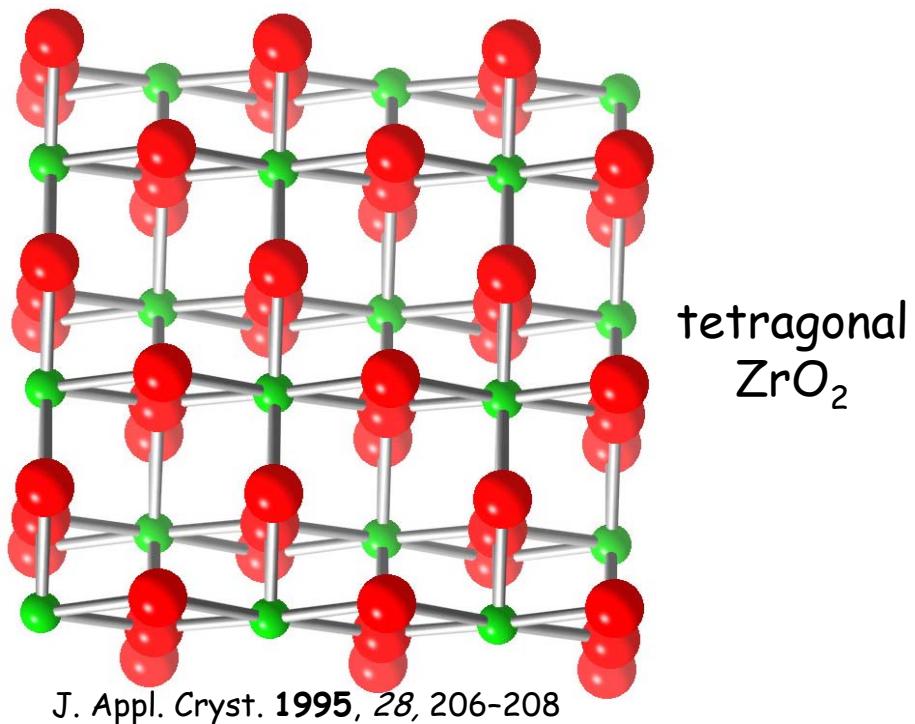
Gate dielectric:
Aluminum oxide
Hafnium oxide
Zirconium oxide
Crystalline

Viewing down the z-axis of Al_2O_3 and ZrO_2



Hexagonal
 Al_2O_3

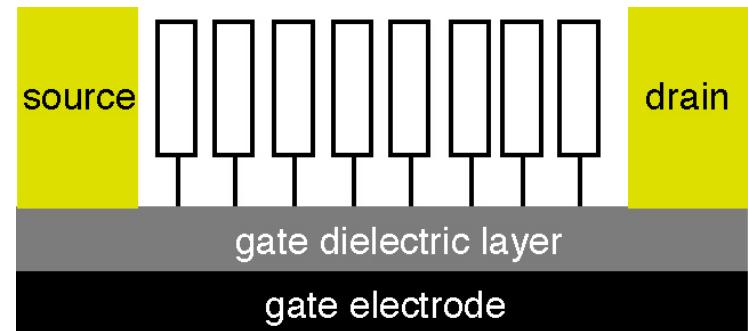
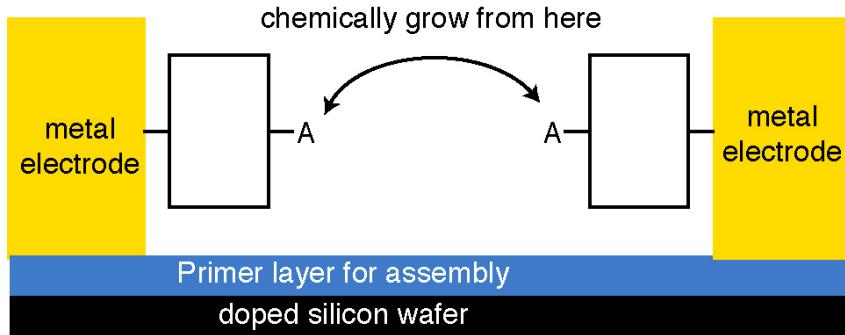
J. Appl. Phys. 1978, 49, 5823.



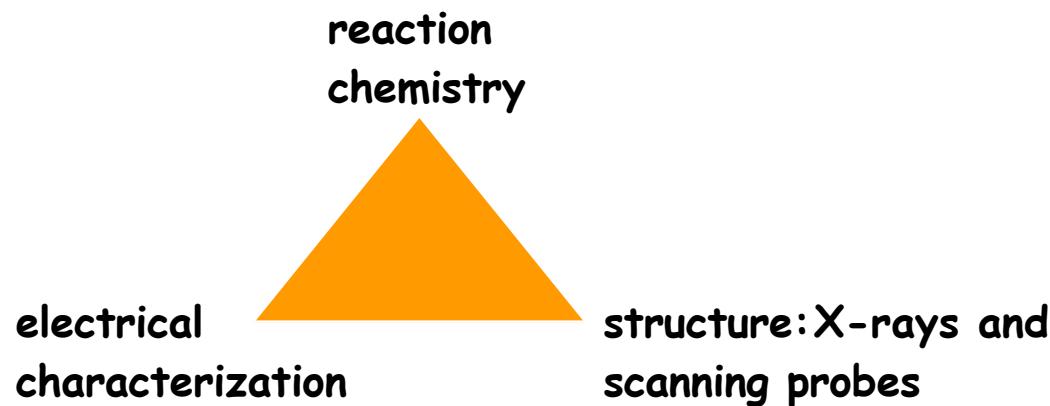
tetragonal
 ZrO_2

J. Appl. Cryst. 1995, 28, 206-208

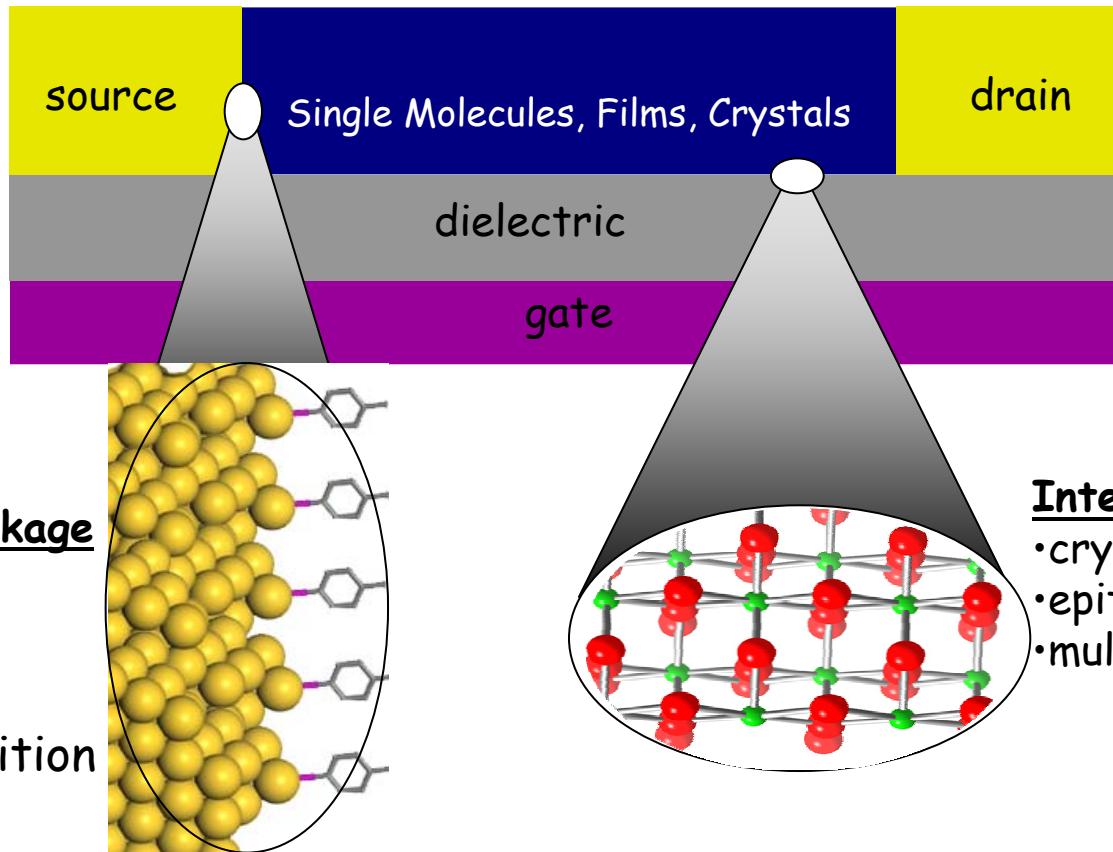
Marry the two approaches



In situ chemical reactions
Synthesis of interfaces



Interfaces



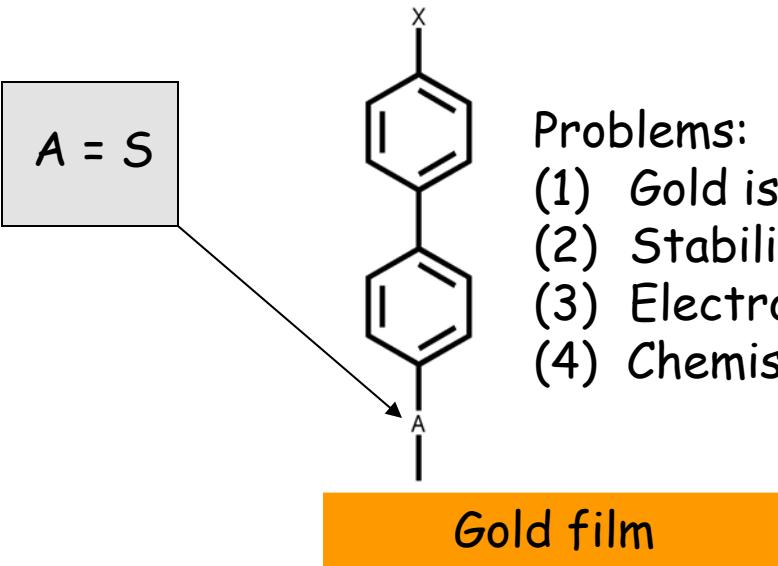
Metal-Molecule Linkage

- CONTACT!
- Order
- Functionality:
growth, recognition

Interfacial Assembly

- crystalline vs. glassy
- epitaxial growth
- multicomponent films

Are there alternatives to thiol/gold?



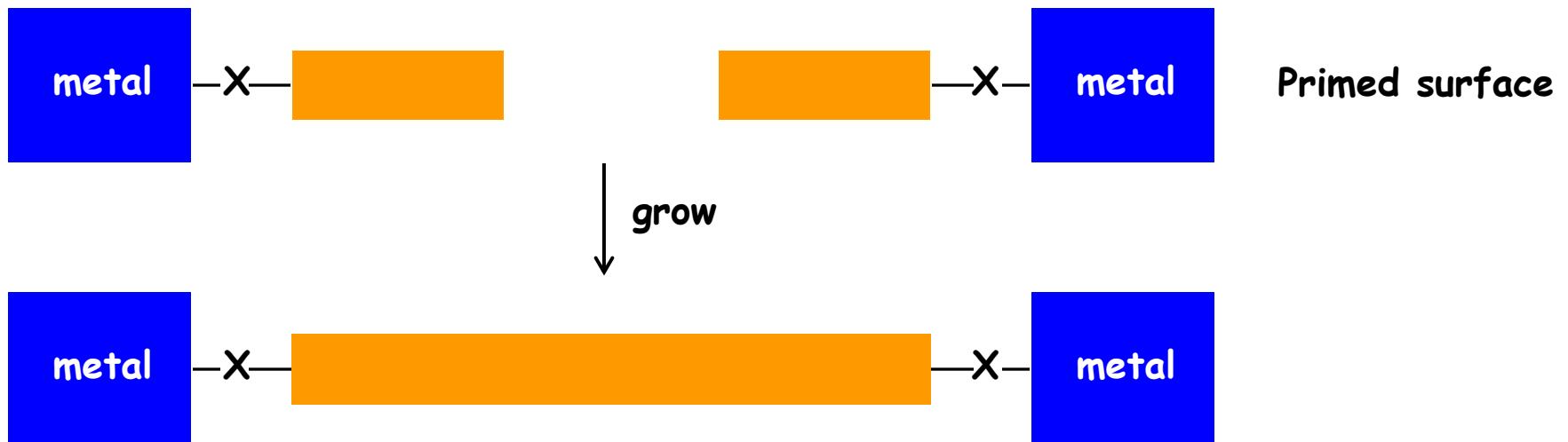
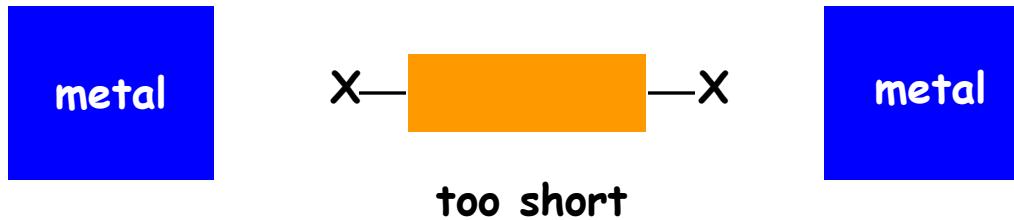
Problems:

- (1) Gold is soft and mobile
- (2) Stability
- (3) Electronics: an insulating link
- (4) Chemistry is limited with gold

What do we want?

Strong
Chemically Stable
Transparent
Catalytic

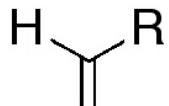
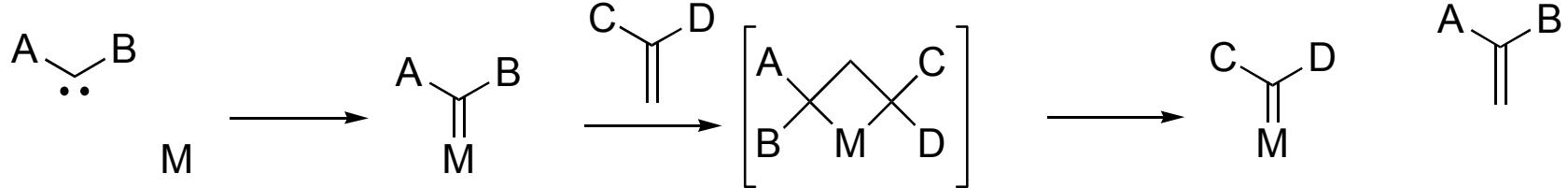
We need to be able to vary the length Catalytic junctions--growth



What do we want?

Strong
Chemically Stable
Transparent
Catalytic

Metal-Carbon Multiple Bonds

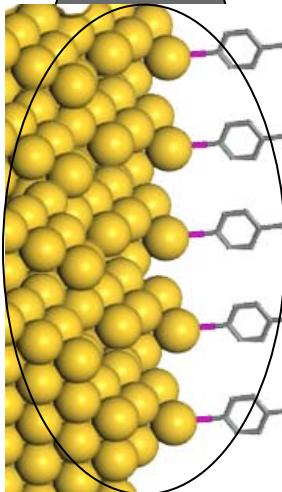
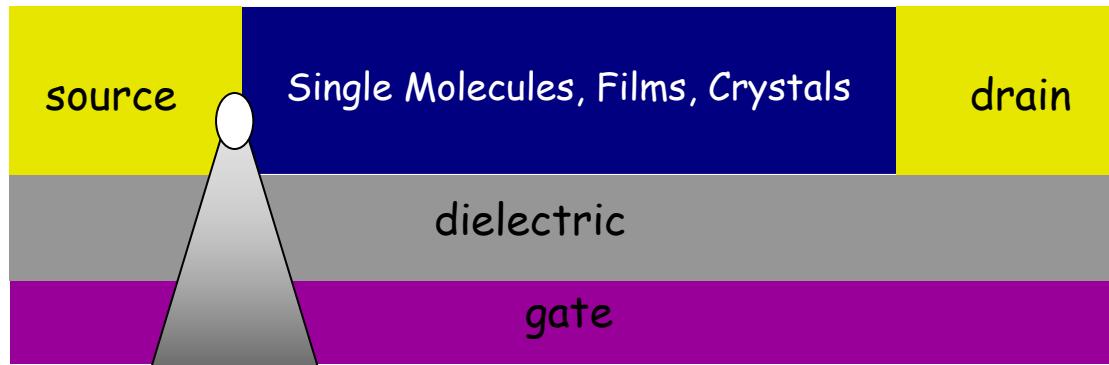


Metal film

Questions:

- Stability?
- Which metals are best?
- Which carbene substituents ?
- What precursor?

Tools that are needed in the toolkit



Metal-Molecule Linkage

- CONTACT!
- Order
- Functionality:
growth, recognition

Need better tools to interrogate these!

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Acknowledgements

Collaborators

- Dr. Mike Steigerwald, Dr. Mark Hybertsen, Prof. Louis Brus (Columbia)
- Hayn Park and Prof. Tony Heinz (Columbia)
- Gina Florio, Michelle Pasamba, and Prof. George Flynn (Columbia)
- Dr. Jochen Ulrich, Prof. Horst Stormer (Columbia)
- Dr. Cherie Kagan (IBM)
- Dr. Art Ramirez, Dr. Christian Kloc, Dr. Theo Siegrist, Dr. Robert Willett (Columbia/Lucent)
- Dr. Graciela Blanchet (DuPont)

Funding:

- Alfred P. Sloan Foundation
- Camille and Henry Dreyfus Foundation
- Dupont Young Investigator
- Beckman Young Investigator
- J. D. Watson Investigator NYSTAR
- National Science Foundation
 - Nanoscale Exploratory Research Grant
 - CAREER Award
- Nanoscience Center at Columbia University
- MRSEC at Columbia University
- U. S. Department of Energy Nanoscience Initiative
- American Chemical Society, Petroleum Research Fund