



Oct 10th, 2025

Liquids Reflectometer (BL-4B) and some science examples

Hanyu Wang
Center for Nanophase Materials Sciences,
Oak Ridge National Laboratory

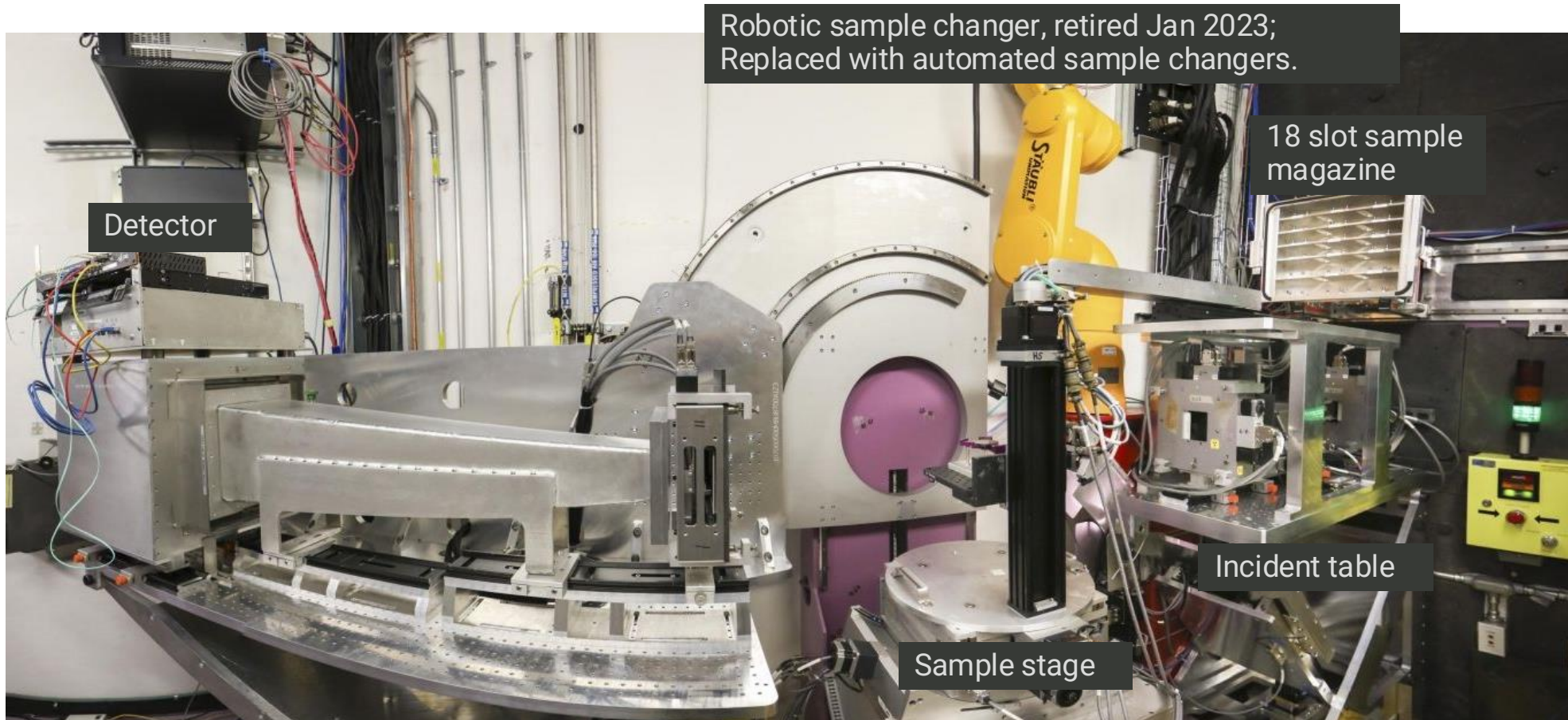


U.S. DEPARTMENT
of **ENERGY**

ORNL IS MANAGED BY UT-BATTELLE LLC
FOR THE US DEPARTMENT OF ENERGY



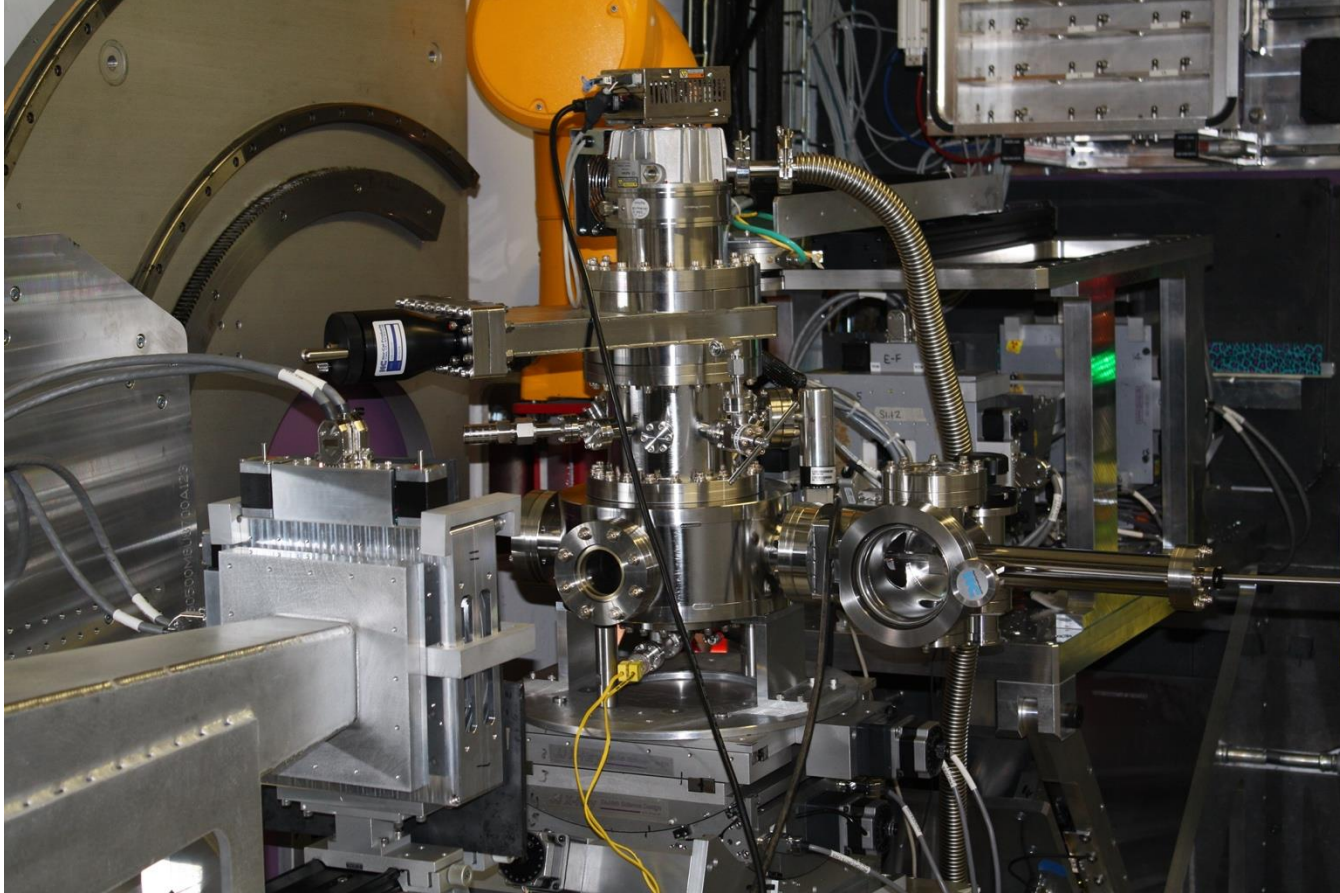
Liquids Reflectometer, BL-4B at SNS, ORNL



LR probes surface and interfacial structures of thin films on length scales of 0.5 nm to 350 nm.

- Change in thickness, scattering length density and roughness
- Air/solid, liquid/solid, air/liquid
- Operates at 60 Hz (3.4 Å) and 30 Hz (6.8 Å), tentatively 20 Hz

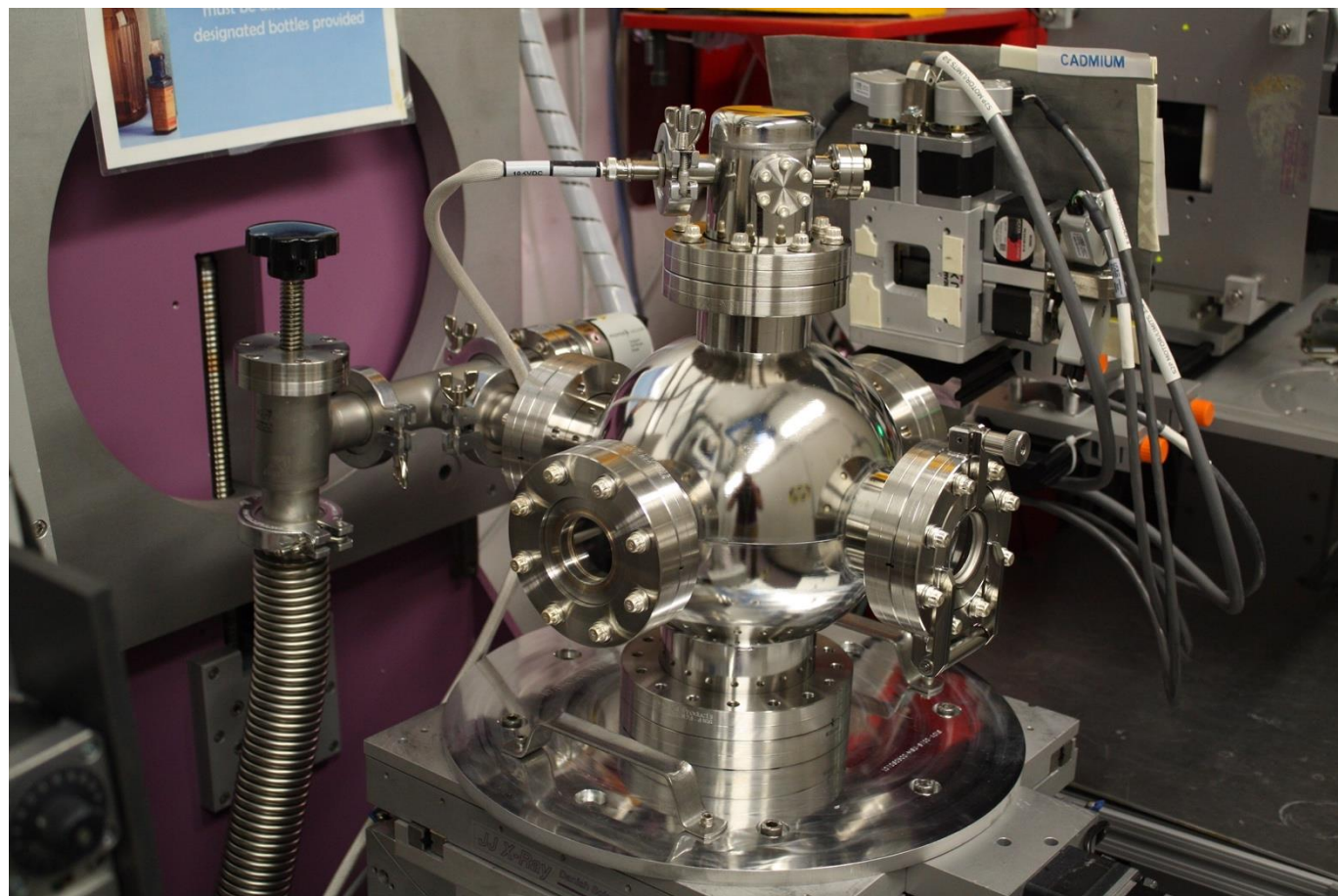
Multi-environment UHV chamber



Gas absorption studies

- Base pressures $< 10^{-9}$ Torr
- Temperature control from RT to 600 °C
- Load lock system from sample loading
- Precision gas doping
- Can be upgraded to incorporate UV-Vis or IR spectroscopy

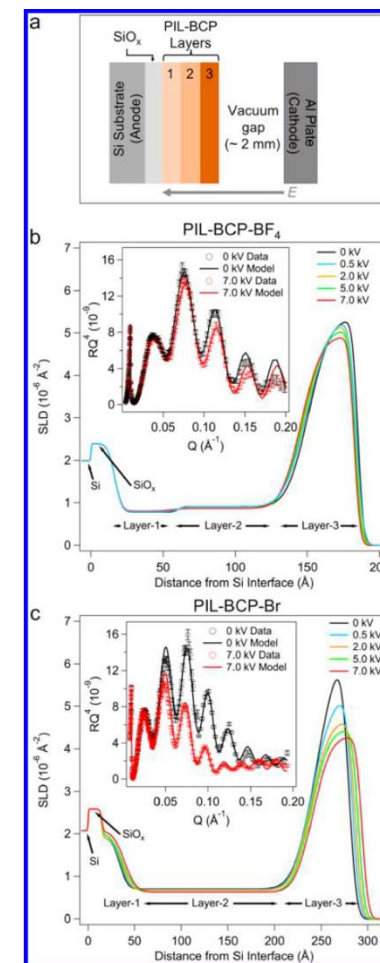
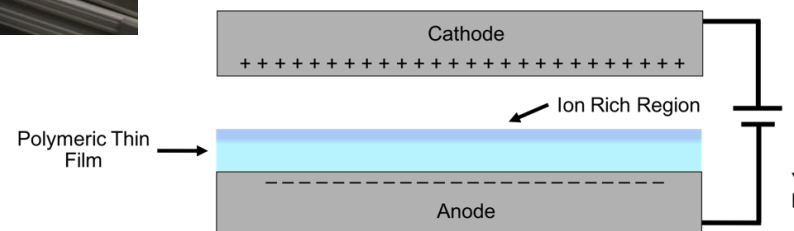
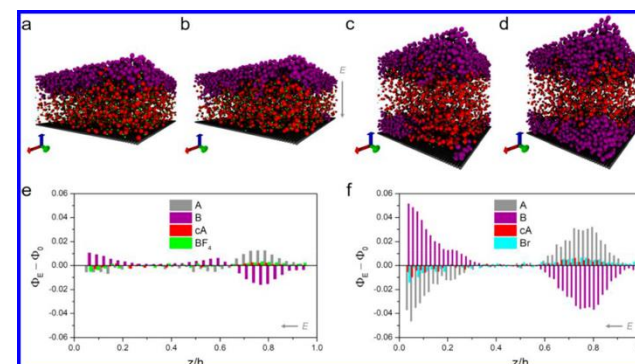
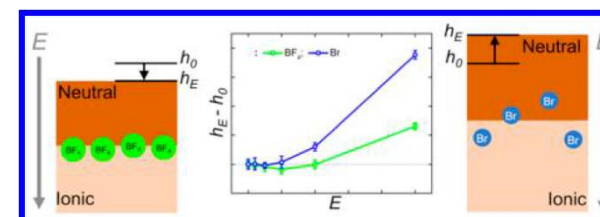
High-field chamber



- Applied electric fields to 7 kV over 7mm gap
- High Vacuum $\sim 10^{-7}$ Torr
- Temperature - RT – 200 °C

Nanoscale Resolution of Electric-field Induced Motion in Ionic Diblock Copolymer Thin Films

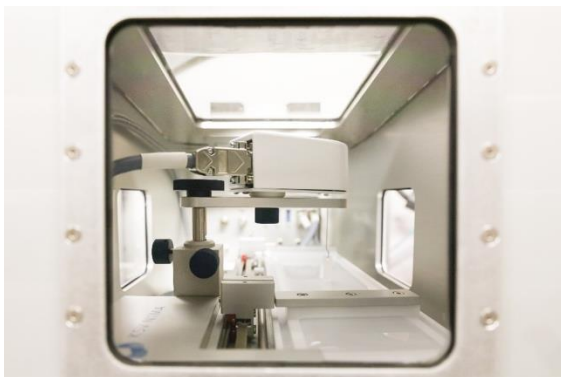
Jason W. Dugger,^{†,‡,§} Wei Li,^{†,‡,§} Mingtao Chen,[‡] Timothy E. Long,^{†,§} Rebecca J. L. Welbourn,^{§,¶} Maximilian W.A. Skoda,[§] James F. Browning,^{*,||} Rajeev Kumar,^{*,†,||} and Bradley S. Lokitz,^{*,†,||}



Langmuir Film Studies



- Materials Science
- Polymer Science
- Biomembrane Science

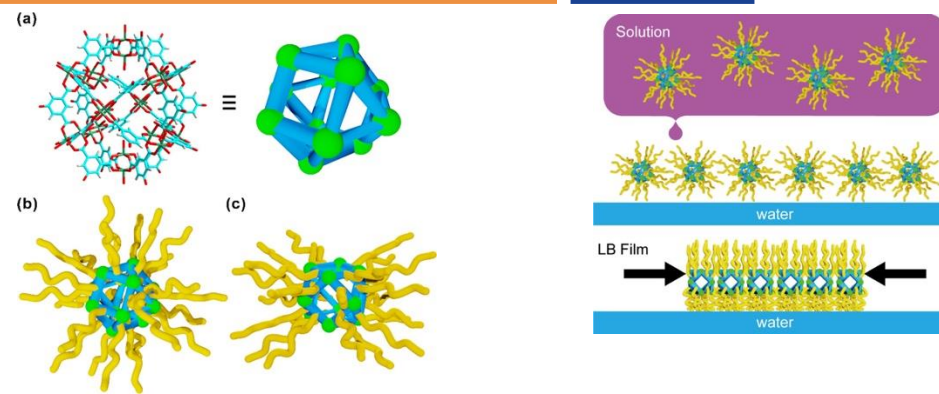


Solvated and Deformed Hairy Metal–Organic Polyhedron

Mu Li, Mingxin Zhang, Yuyan Lai, Yuan Liu, Candice Halbert, James F. Browning,* Dong Liu,* and Panchao Yin*

✓ Cite This: *J. Phys. Chem. C* 2020, 124, 15656–15662

 Read Online

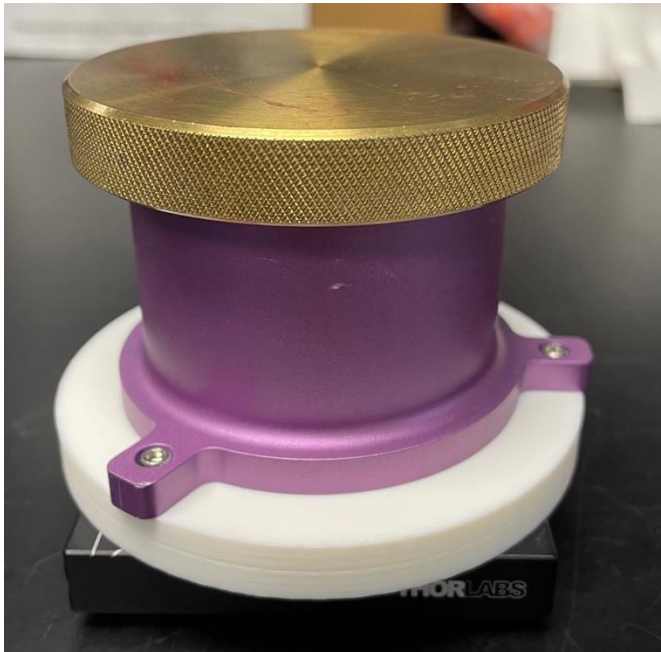


Probe COVID-19 virus infection dynamics in real time in 2020

Critical questions:

- ❖ To protect the host:
 - What factors affect the insertion of fusion peptide (FP) into the host membrane?
 - Membrane compositions, e.g., charge density, mechanical properties
 - Effective concentration of FP
 - Can we impede membrane fusion?
 - Block the insertion of FP
 - Block the folding of HR1 and HR2

Humidity can (humidity chamber)

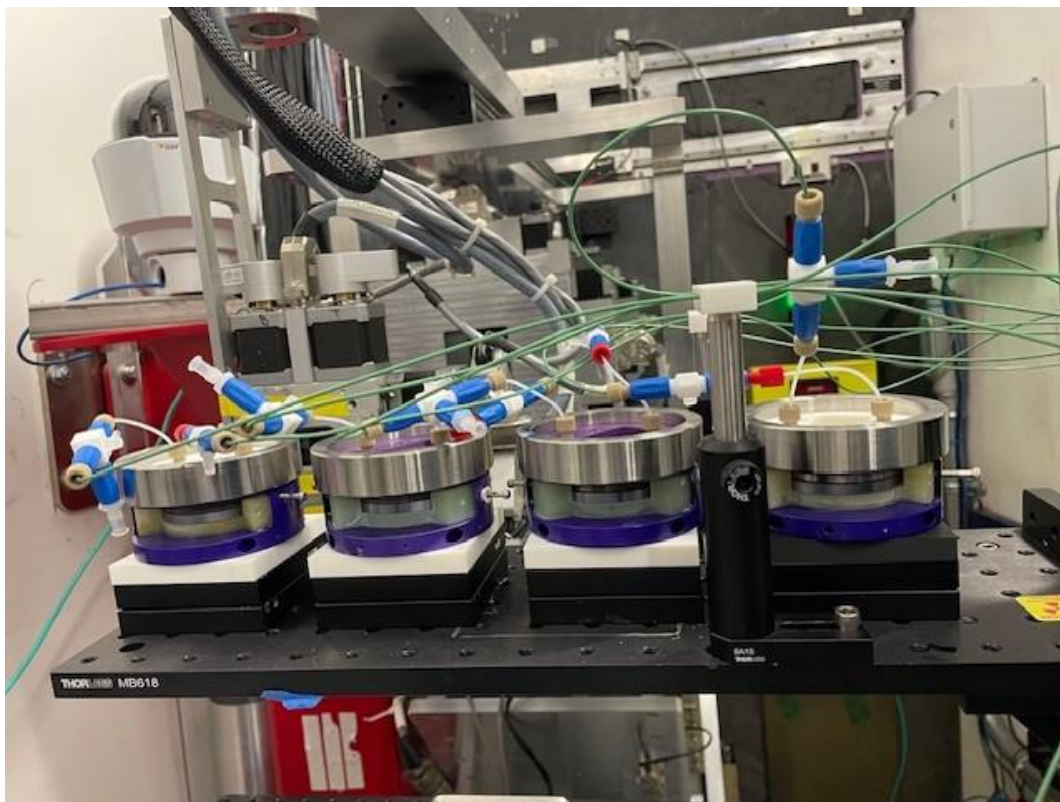


Humidity can

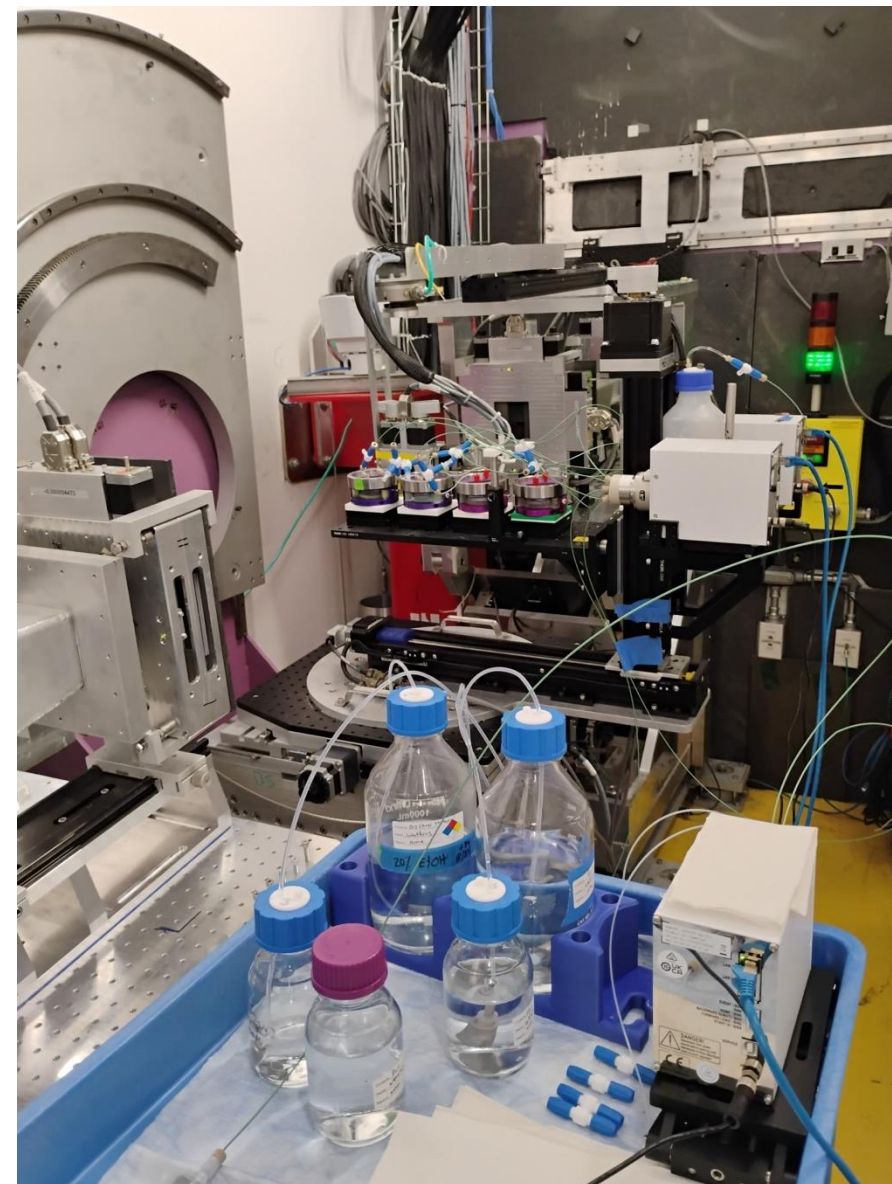
Humidity study:

- Humidity can
 - Fixed single point relative humidity study
 - 3D printed sample holder can accommodate both reflect-up and reflect-down measurement
- Humidity chamber
 - Precise and controlled humidity for top and bottom separately

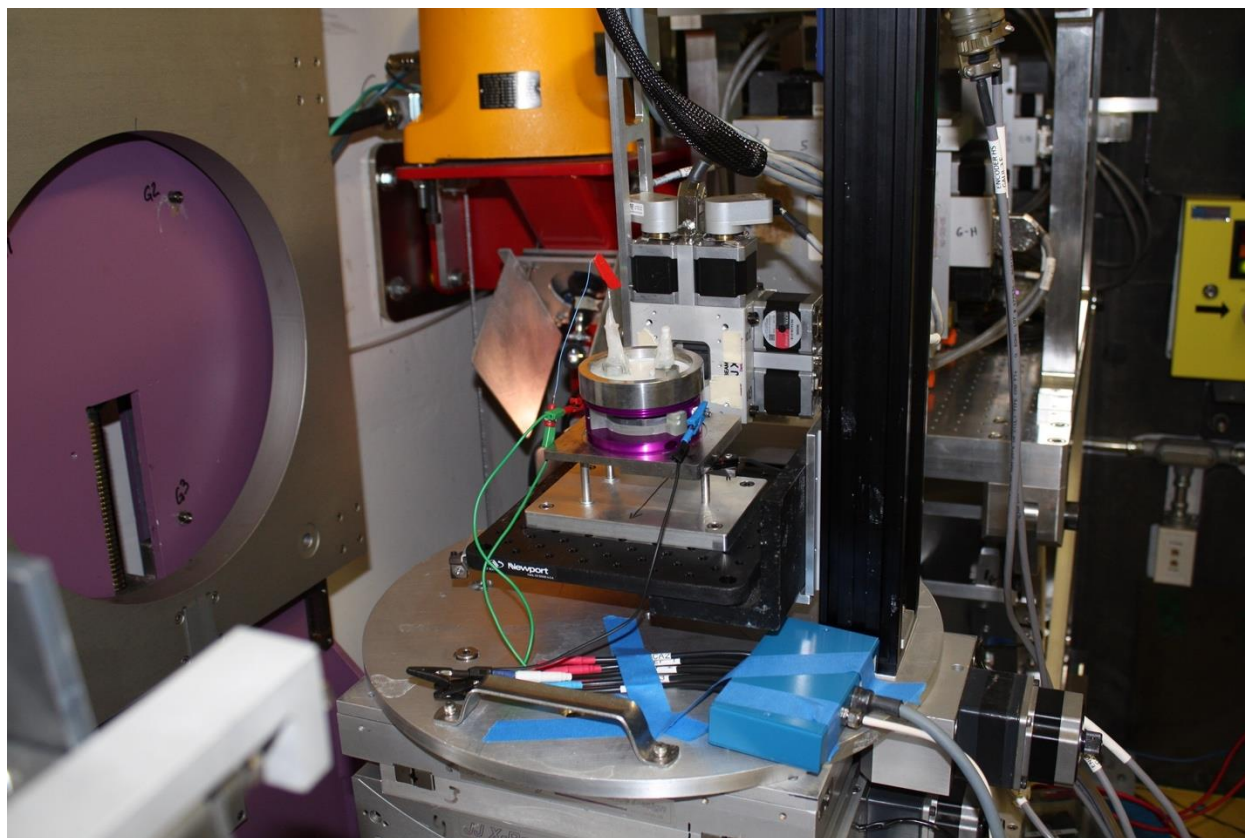
Liquid cell



- Solid-liquid interface
- Mainly use Si substrates (2-inch diameter \times 5 mm thickness)
- Automatic control for up to 4 cells (liquid exchange via HPLC pump)



Electrochemical Cell



In situ electrochemistry

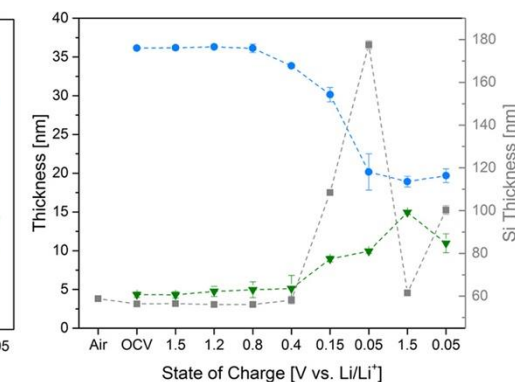
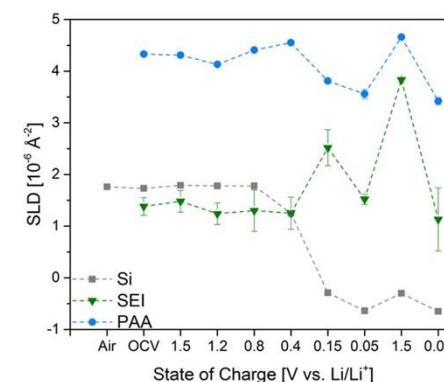
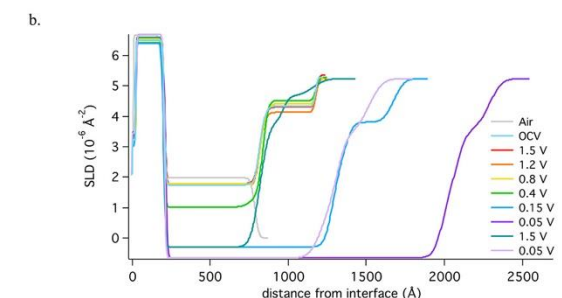
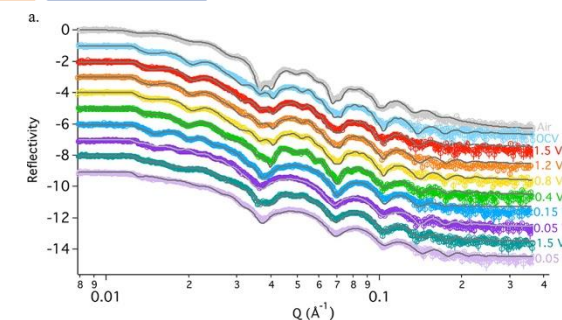
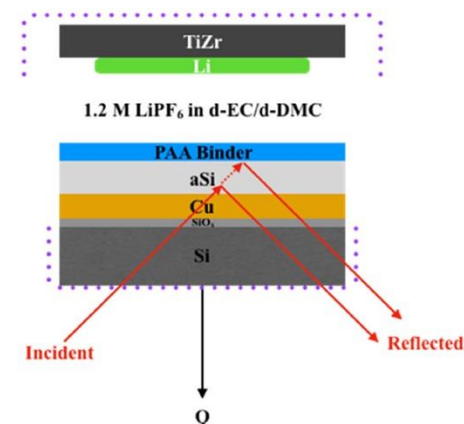
- Ion transport and interphase formation in energy storage materials
- Corrosion phenomena
- Redox in chemistry and biology

The Study of the Binder Poly(acrylic acid) and Its Role in Concomitant Solid–Electrolyte Interphase Formation on Si Anodes

Katie L. Browning,* Robert L. Sacchi,* Mathieu Doucet, James F. Browning, Joshua R. Kim, and Gabriel M. Veith*

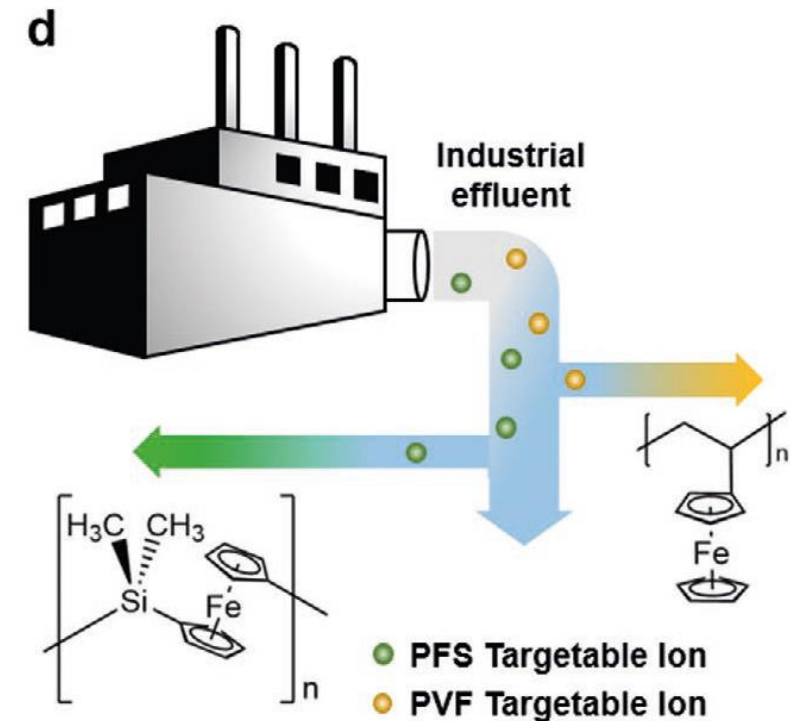
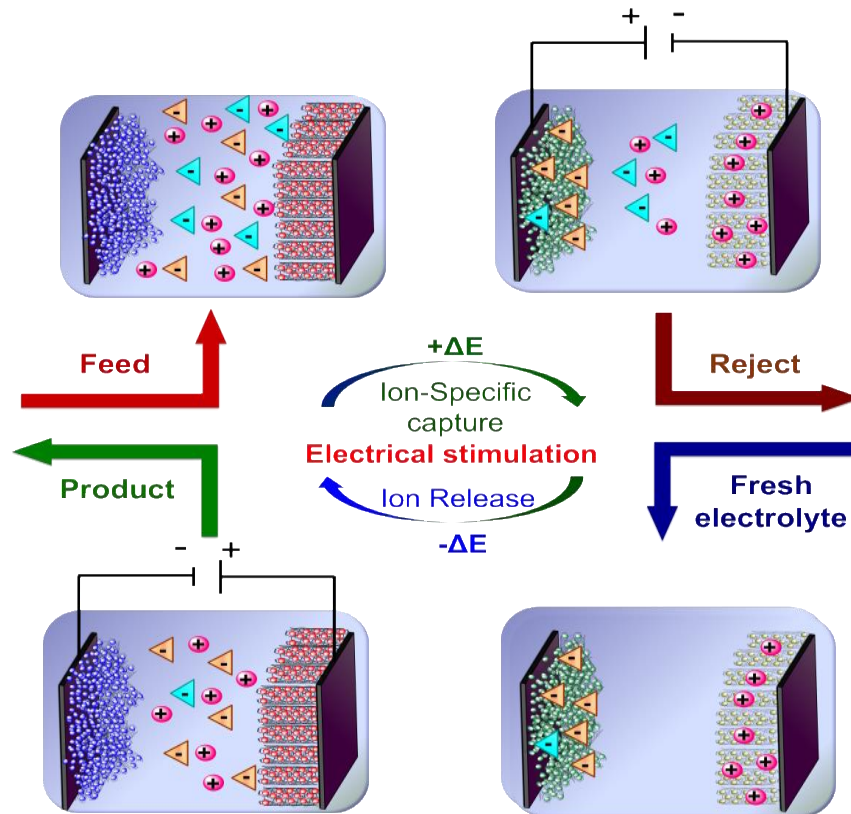
Cite This: *ACS Appl. Mater. Interfaces* 2020, 12, 10018–10030

Read Online

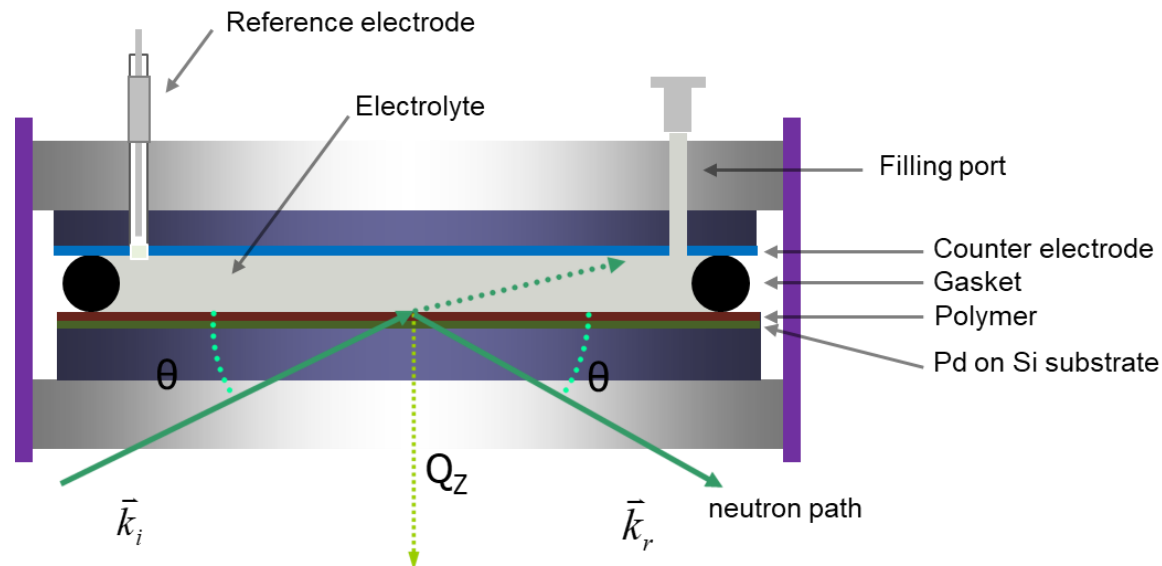


Science case 2: Understanding ion selectivity with redox-polymers

- Fundamental understanding of contributing mechanisms, and control of interactions to achieve ion-selectivity.
- An important platform by tailoring advanced redox-polymer for targeting heavy metal recovery and remediation.

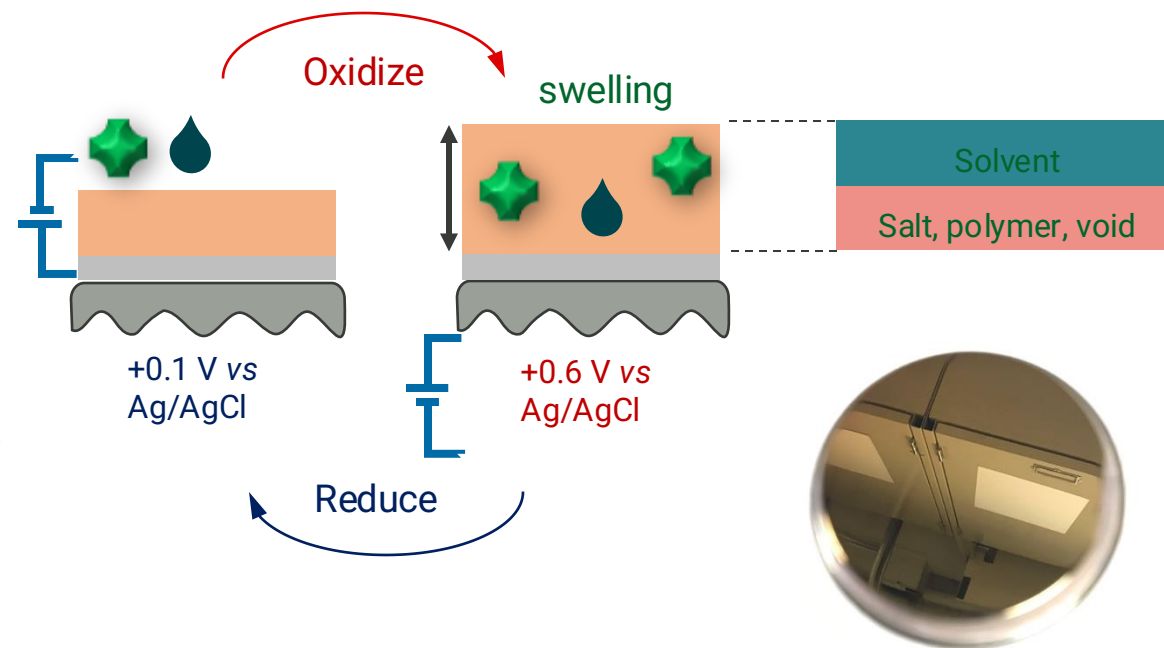


Experiment preparation



Schematic diagram of the EC cell and NR geometry

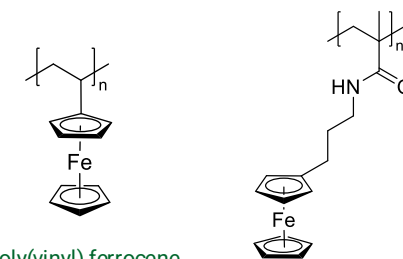
Working electrode: redox-active polymer
 Counter electrode: TiZr
 Reference electrode: Ag/AgCl
 Electrolyte: Target ions in H₂O or D₂O or mix of H₂O and D₂O



Sample on a 2-inch diameter Si wafer

NR: To probe the spatial distribution, ion migration and film swelling as a function of applied potentials on polymer thin film.

Salt: Cr, Mo, Re, V, etc.

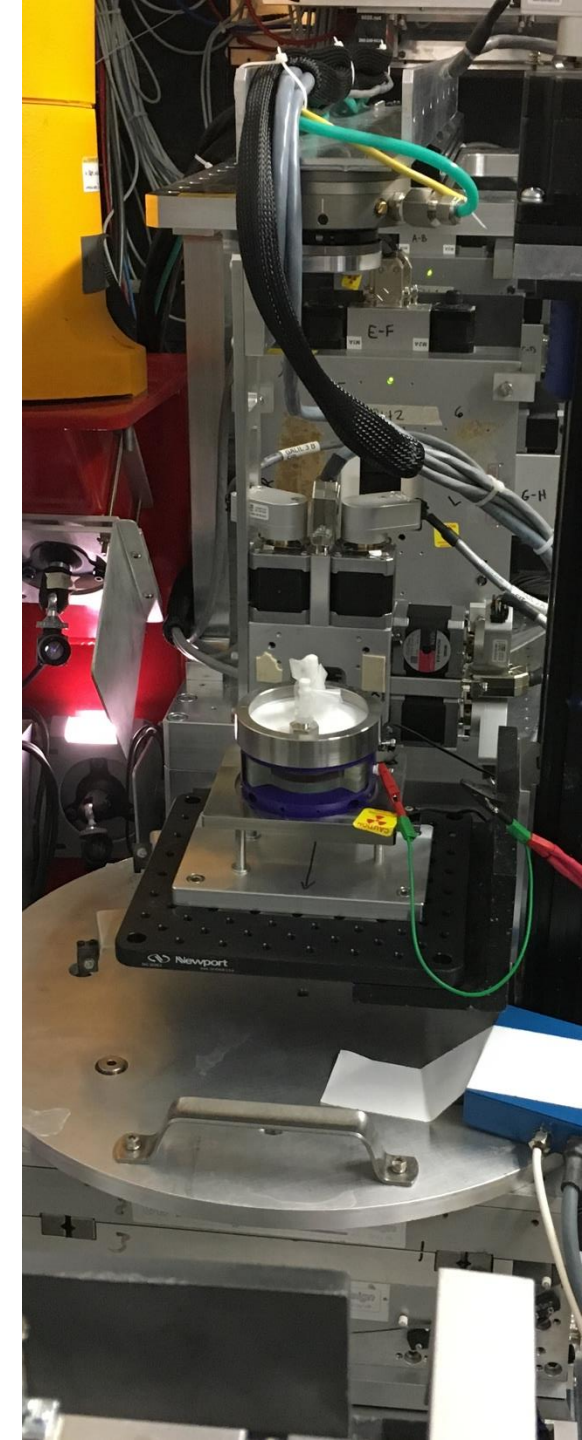
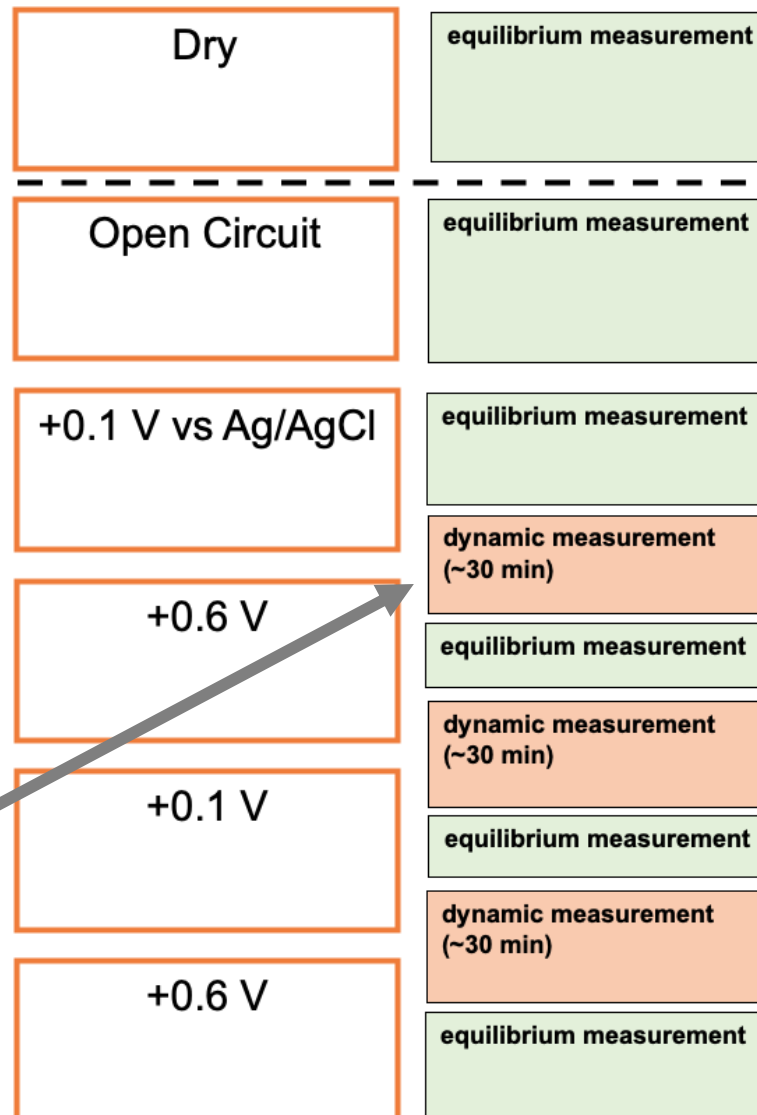


Pendant group polymer

Experiment plan

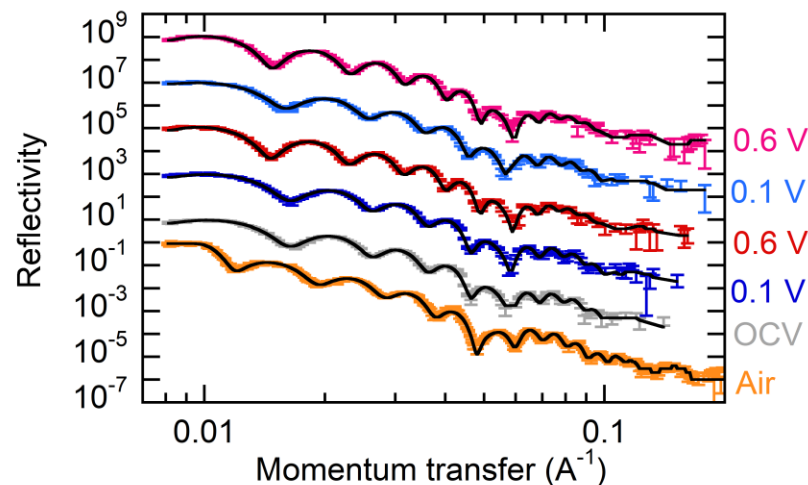
- Measure ion selectivity and hydration by comparing combinations of polymer, target ion, and solvent ($\text{H}_2\text{O}/\text{D}_2\text{O}$).
- Apply a series of potential for each system.
- Because of the way we wanted to measure, no automated control between EC and NR possible... very labor intensive.

Suggested time-resolved measurements. This was done during the relaxation period between each applied potential...

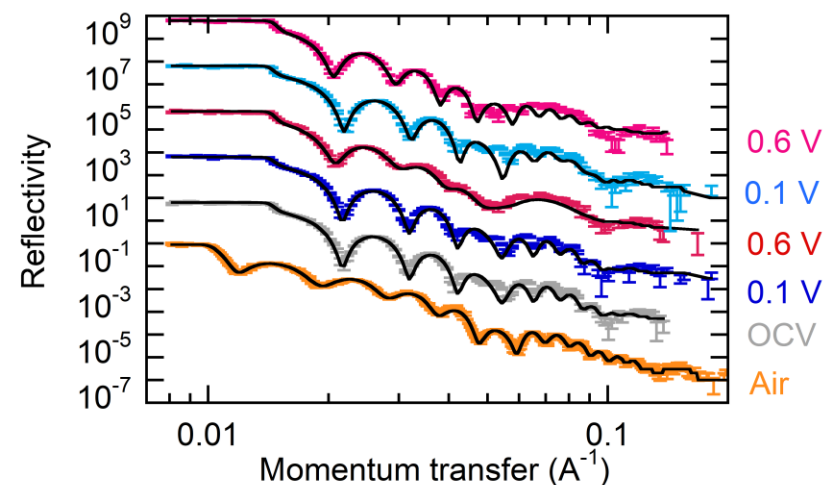


NR and SLD profiles (a very small part of data...)

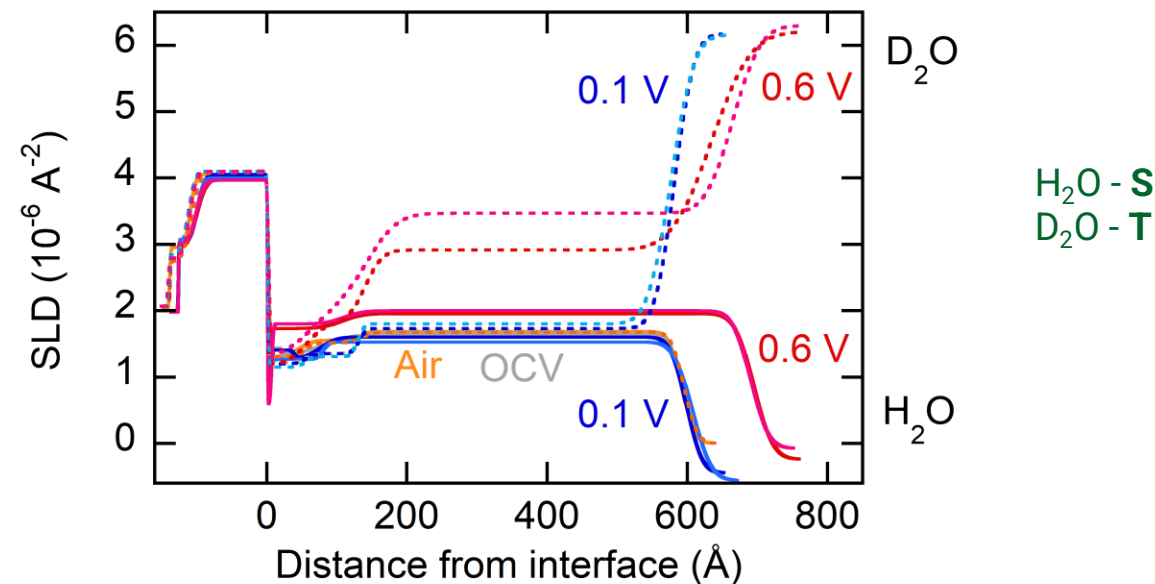
$\text{H}_2\text{O} - \text{S}$



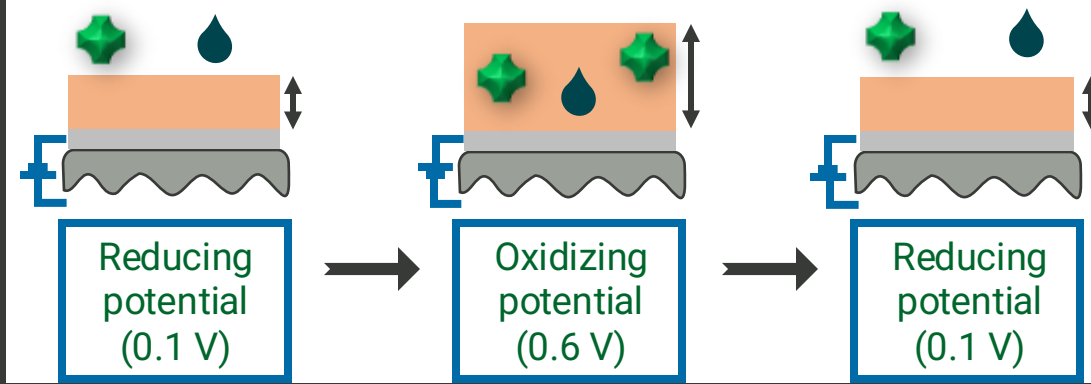
$\text{D}_2\text{O} - \text{T}$



100% ReO_4^-



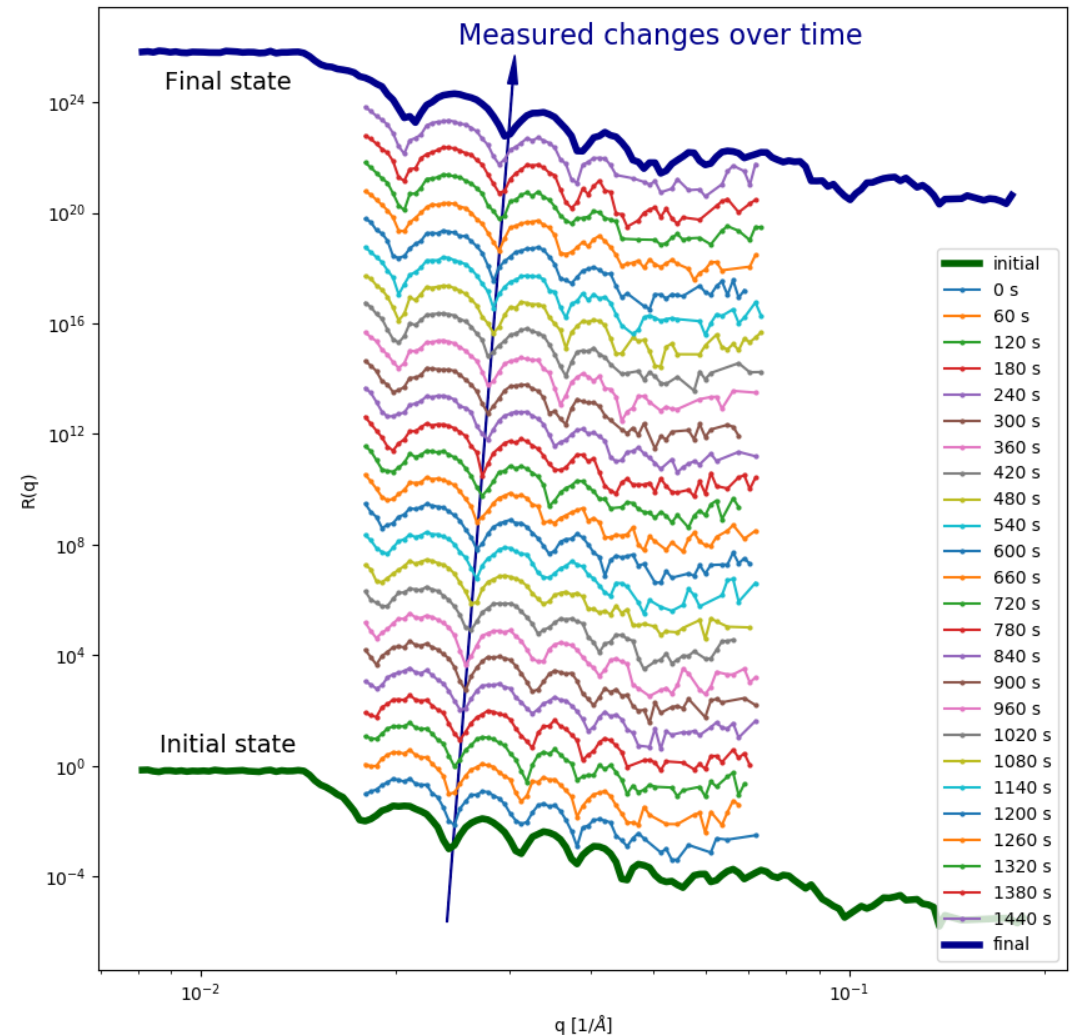
In situ neutron reflectometry: study solvation and swelling



NR and SLD profile monovalent behavior for PFPMAm.

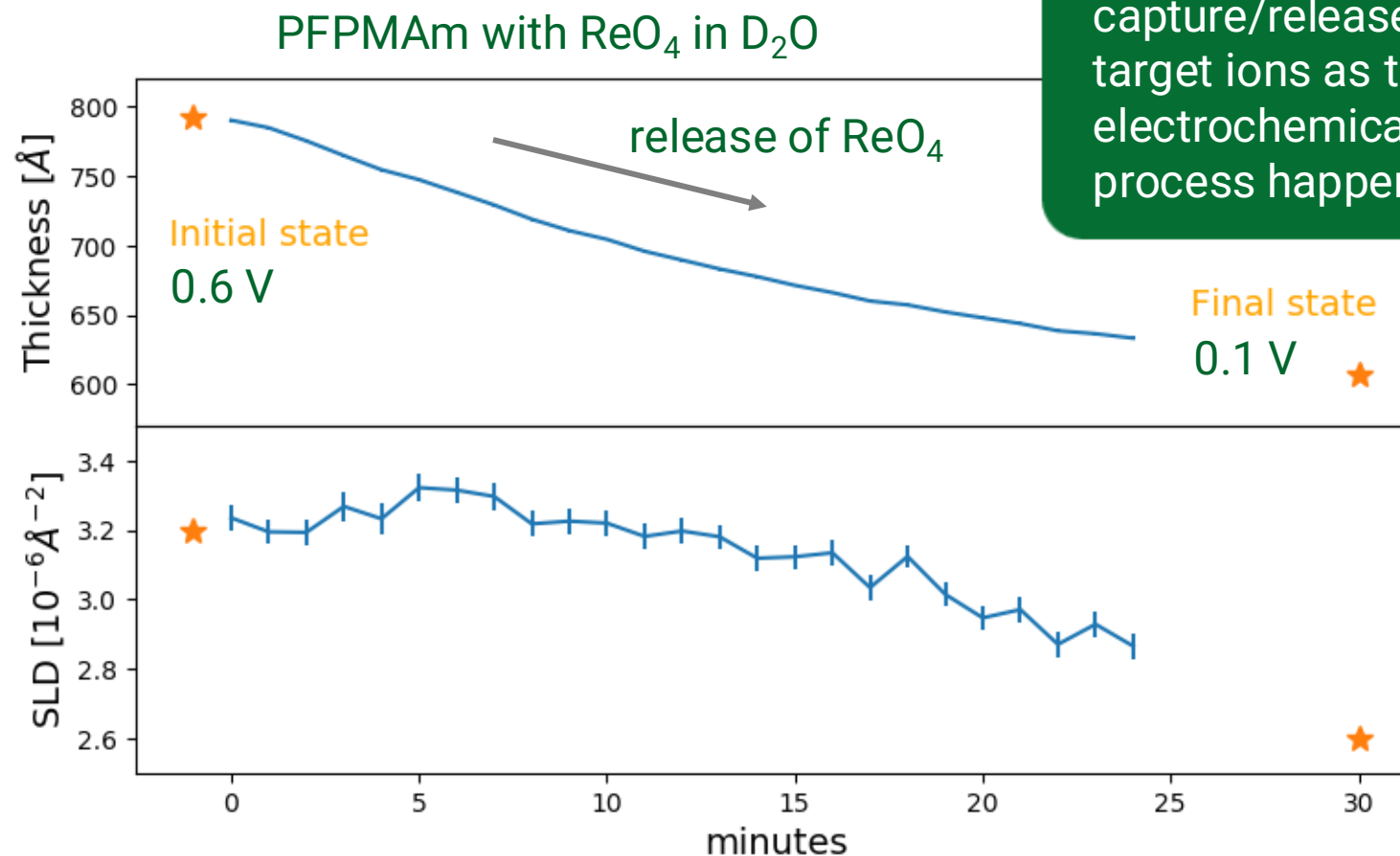
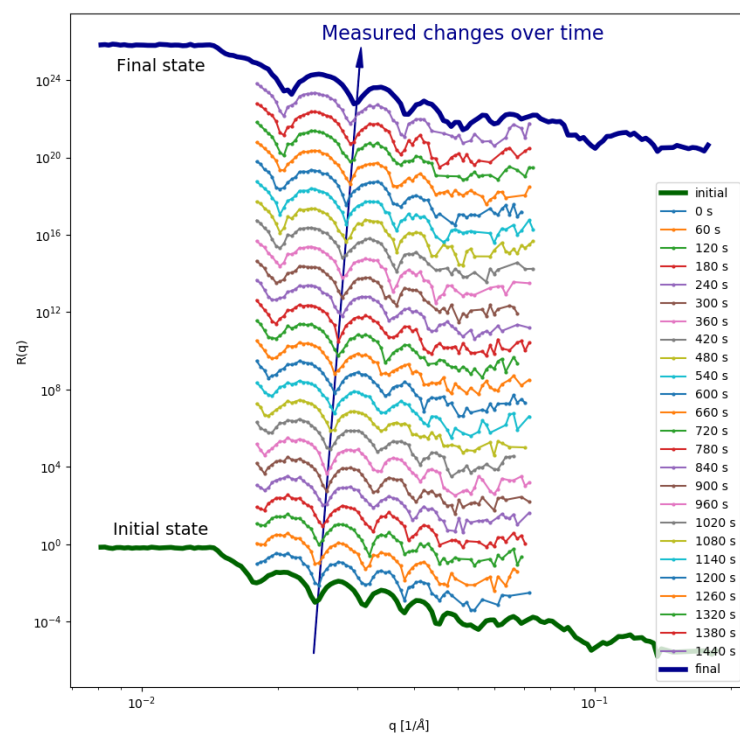
New type of measurements: time-resolved reflectometry

- It complements our standard measurements.
- Time scale available at 4B is well matched to EC processes.
- Our standard measurement is made of multiple runs, e.g., 7 runs at 60 Hz, or 3 runs at 30 Hz.
- Operating at 30 Hz and judiciously picking θ and λ band, could measure a range of $0.017 < Q < 0.05 \text{ 1/\AA}$.
- Initial and final states measured for complete Q range (~45 minutes).
- Data sliced in 1-minute intervals.



Need for new analysis approach: Bayesian fitting loop

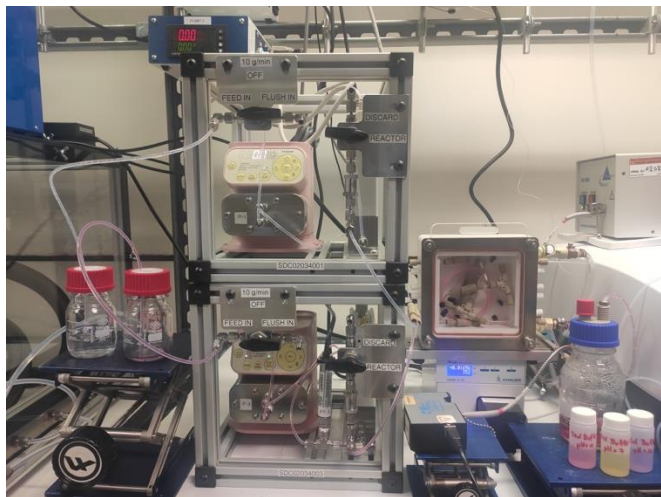
- That's a lot of data to fit: **2,600 time-resolved reflectivity curves...**
- We developed a way to use the initial state as prior information and perform iterative fits, with each fit result becomes the prior for the next.



We can see the capture/release of target ions as the electrochemical process happens!

Capabilities at Macro group CNMS

Autonomous Continuous Flow Reactor Synthesis for Atom Precision with Scalability (AutoFlowS)



Snapdragon

Large Scale synthesis

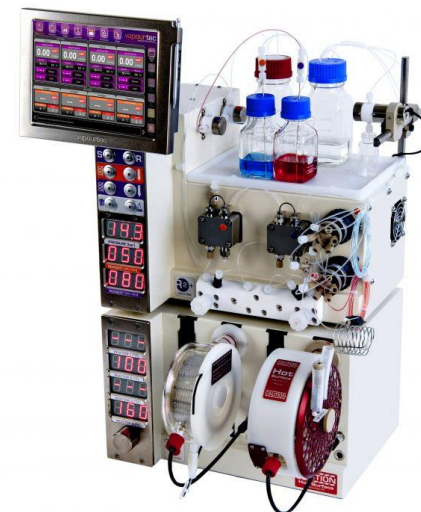
Multiple reactors in one flow

Can connect to a large amount of instrumentation



Phoenix by ThalesNano

- High pressure, high temperature
- Two compartments for temperature control
- Gas module



Vapourtec

- Small scale synthesis
- Organic synthesis
- Deuteration

Various analytical techniques

Raman
UV-Vis
QCM
GPC/HPLC
SPR
ESR
RI



3D printing lab

Fused deposition modeling (FDM)

Ultimaker 2



Ultimaker 3

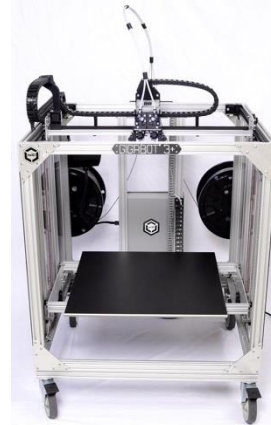


Single print nozzle

double print nozzle

Suitable materials: Filament. Constant diameter during extrusion.

Gigabot

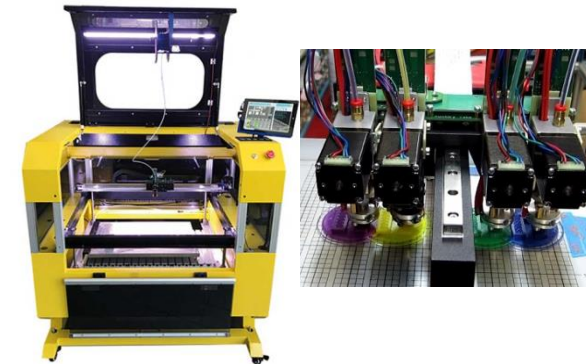


2 print heads. Big size print.



Common materials: ABS, PLA, PC

Hyrel 3D



Up to 5 print heads

Stereolithography (SLA)

Formlab 3B



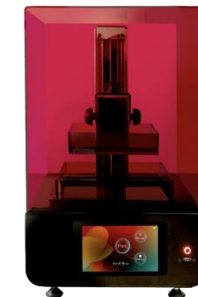
Formlab 3



Form-cure



Visible light source
Photocentric



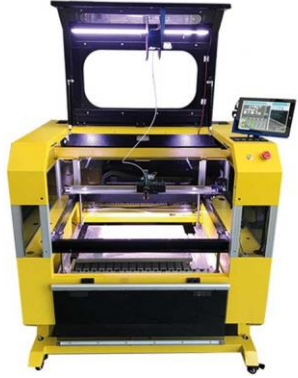
- Post curing of SLA printed object
- UV light chamber,
 - Heated chamber (up to 70 C)
 - Selected time

Suitable material: Photo-polymerization at a fast rate. Good flowability.

Typical example: Acrylate based materials

Direct Ink Writing (DIW)

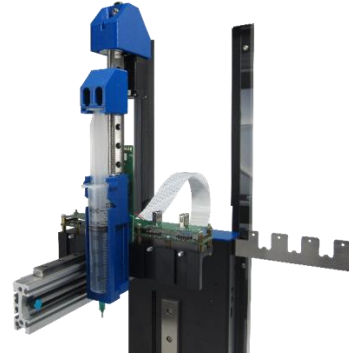
Hyrel 3D



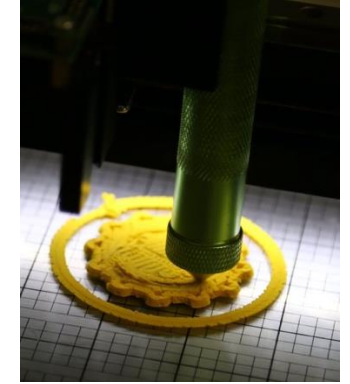
Suitable materials:

- Fluids with thixotropic rheology

Syringe head:
DIW printing



Heated “syringe”
Up to 200 C



Common materials:

Metal, ceramic, thermoset, thermoplastic, solution, emulsion, hydrogel...

Selective Laser Sintering

Sinterit Lisa



Suitable material:
Powder based material

Common materials:
Nylon 12

Injection molding machine

Model 150A injection machine



APX-PIM injection machine



Center for Nanophase Materials Sciences

[Home](#) [About Our Team](#) [About CNMS](#) [Research](#) [Research Sections and Groups](#) [User Information](#) [Media and Events](#)

The Center for Nanophase Materials Sciences (CNMS) at Oak Ridge National Laboratory (ORNL) offers the national and international user community access to staff expertise and state-of-the-art equipment for a broad range of nanoscience research, including nanomaterials synthesis, nanofabrication, imaging/microscopy/characterization, and theory/modeling/simulation.

Quick Resource Links



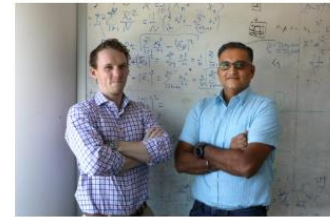
Staff Expertise



Capabilities and Instruments



Research Impact



Become a CNMS User



Spallation Neutron Source (SNS) and
Center for Nanophase Materials
Sciences (CNMS)
at Oak Ridge National Laboratory
(ORNL)

Welcome to submit user proposals to CNMS!

<https://www.ornl.gov/facility/cnms>
<https://neutrons.ornl.gov/>