

#### X-RAY ABSORPTION SPECTROSCOPY (XAS)



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## OUTLINE

- Background X-ray absorption spectroscopy (XAS)
- X-ray absorption near edge spectra (XANES) process
- XANES examples
- Extended X-ray absorption fine structure (EXAFS) fundamentals
- EXAFS examples







## **1920 PUBLISHED ABSORPTION EDGES**

The K-Characteristic Absorption Frequencies for the chemical elements magnesium to chromium

By Hugo Fricke



3 https://journals.aps.org/pr/pdf/10.1103/PhysRev.16.202 Argonne



## **1971 UNDERSTANDING OF EXAFS**

New Technique for Investigating Nanocrystalline Structures: Fourier Analysis of the Extended X-ray-Absorption Fine Structure

By Dale Sayers, Edward Stern and Farrel Lytle





4

## ACRONYMS

- General terms:
  - XAS: X-ray absorption spectroscopy
  - XAFS: X-ray absorption fine structure
- Specific terms:
  - XANES: x-ray absorption near edge structure

5

 EXAFS: Extended x-ray absorption fine structure



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6

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#### X-ray Absorption Near-Edge Structure



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7

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#### Extended X-ray Absorption Fine Structure



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## **XANES: ELECTRONIC TRANSITIONS**

- XANES directly probes the angular momentum of the unoccupied electronic states: these may be bound or unbound, discrete or broad, atomic or molecular.
- Dipole selection rules apply:  $\Delta I = \pm 1$
- Primary transition will be:
  - $s \rightarrow p$  for K (1s core electron) and L<sub>1</sub> (2s core electron initial state) edges
  - $p \rightarrow d$  for L<sub>2</sub> (2 $p_{\frac{1}{2}}$ ) and L<sub>3</sub> (2 $p_{\frac{3}{2}}$ ) edges
- But....final state usually not atomic-like and may have mixing (hybridization) with other orbitals. This is often the interesting part of the XANES!

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## **METAL K-EDGE XANES**



- Absorption edge: dipole  $1s \rightarrow 4p$  transition ( $\Delta l = \pm 1$ )
- Pre-edge: mixing of 3d-4p opens  $1s \rightarrow 3d$  transition

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#### OXIDATION STATE OF MN OXIDES Mn K-edge XANES



- Many edges of many elements show significant edge shifts (binding energy shifts) with oxidation state.
- Early observation was by Berengren for phosphorus in 1920\*!
   \*See "A history of X-ray absorption fine structure", R. Stumm von Bordwehr, Ann. Phys. Fr. 14 (1989) 377-466



#### PT XANES DURING IN SITU REDUCTION Direct in situ measurement of Pt reduction



- Pt L<sub>3</sub>-edge p->d transition
- Pt electrons: [Xe]4f<sup>14</sup>5d<sup>6</sup>s<sup>1</sup>
- · Pt white line intensity decreases as temperature increases due to 5d electrons filling
- Rate of decrease is fastest at ~270°C.

   **ENERGY** Assessment of Every laboratory is a sub-sub-structure of Every laboratory.



#### FAST-CHARGING LITHIUM-ION BATTERIES Mechanism from XANES and DFT

Article

## A disordered rock salt anode for fast-charging lithium-ion batteries

Haodong Llu <sup>138</sup> , Zhuoying Zhu <sup>113</sup> , Qizhang Yan <sup>1</sup> , Sicen Yu <sup>1</sup> , Xin He <sup>2</sup> , Yan Chen <sup>3</sup> , Rui Zhang <sup>4</sup> ,
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Khalil Amine <sup>6</sup> , Tianpin Wu <sup>11</sup> , Jun Lu <sup>6</sup> <sup>62</sup> , Huolin L. Xin <sup>42</sup> , Shyue Ping Ong <sup>112</sup> <sup>62</sup> & Ping Liu <sup>112</sup>

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The team includes 26 authors with expertise in electrode chemistry, materials synthesis, neutron diffraction, in-situ XRD

12 ICP-OES, STEM, XAS, SEM, XPS, and DFT.



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# BEER'S LAW: ABSORPTION OF X-RAYS BY MATTER



 $I_t = I_0 e^{-\mu x}$ 

- μ x: absorption length of a material
- One absorption length,  $I_t = 37\% I_0$
- Two absorption lengths,  $I_t = 13\% I_0$







#### MEASUREMENT OF X-RAY ABSORPTION COEFFICIENT



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14



#### **DEFINITION OF EXAFS** X(E)



$$\chi(E) = \frac{\mu(E) - \mu_0(E)}{\Delta \mu(E)}$$

Measured absorption coefficient in transmission or fluorescence

$$\mu = \log \frac{I_0}{I_t} \qquad \mu \propto \frac{I_f}{I_0}$$

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#### **DEFINITION OF EXAFS**



$$\chi(E) = \frac{\mu(E) - \mu_0(E)}{\Delta \mu(E)}$$

Smoothly varying background function, representing the absorption without fine structure

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17



### **DEFINITION OF EXAFS**



$$\chi(E) = \frac{\mu(E) - \mu_0(E)}{\Delta \mu(E)}$$

$$k^2 = 2 m_e(E-E_0)/\hbar$$

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## FERMI'S GOLDEN RULE

## $\mu(E) \propto |\langle i|H|f\rangle|^2$

 $\langle i |$  Initial State: atom with core electron

- H Interaction term: incident x-ray
- $|f\rangle$  Final State: atom with core hole, photo-electron
- Transition between two quantum states
- Initial state is well localized at the absorbing atom
- Final state is not, but can be written in terms of two parts

$$|f\rangle = |f_0\rangle + |\Delta f\rangle$$

adsorbing neighboring

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$$\chi(k) \propto \psi_{scatt}(k, r = 0)$$

$$\propto \frac{e^{ikR}}{kR} [F(k)e^{-i\delta(k)}] \frac{e^{ikR}}{kR}$$

$$\chi(k) \propto \frac{F(k)}{k^2 R^2} \sin(2kR + \delta(k))$$

Atomic Phase shift has two parts. One part from the absorbing atom and another part from the scattering atom.

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#### LIFETIME OF PHOTOELECTRON AND CORE HOLE



- X-ray absorption fine structure requires a core hole and a coherent photoelectron.
- The mean free path  $\lambda$  of the photoelectron limits the contributions to EXAFS to the first few neighbors

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## PASSIVE ELECTRON REDUCTION FACTOR

$$S_0^2 = \left| \left\langle \phi_f^{N-1} \middle| \phi_i^{N-1} \right\rangle \right|^2$$

- The initial and final states include all the passive electrons of the absorbing atom.
- If the electrons do not react to the creation of the core hole, then there is no difference in these passive electrons and the initial and final states are identical giving S<sub>0</sub><sup>2</sup>= 1, indicating a screened potential.
- If the electrons do react (they only have  $10^{-15}$  seconds to do so) then the final state will be slightly different from the initial state. The final state will have the electrons pulled in slightly due to the core hole and the value for  $S_0^2$  can be as small as ~ 0.7, indicating an unscreened potential.
- Value for  $S_0^2$  depends only on the adsorbing atom and can be transferred from a standard material such as a foil.
- $S_0^2$  often contains experimental affects such as energy resolution so it is best to measure a foil under the same experimental conditions as the unknown sample.

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## MANY SINGLE SCATTERING PATHS



Summing over scatting events

$$\chi(k) = \sum_j \chi_j(k)$$

Summing over shells of atoms

$$\chi(k) = \sum_i N_i \chi_i(k)$$

- EXAFS signal is the sum of many individual scattering events j.
- It is convenient to group scattering events from shells of atoms, i, of the same type at the same radial distance from the adsorbing atom. Then we need to multiply the scattering contribution by the number of atoms in each shell (N<sub>i</sub>).

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## **MEAN SQUARE DISPLACEMENT**



- Summing over shells introduces an error because not all of the atoms in a shell are at exactly the same radial distance from the adsorbing atom.
- The scattering process takes place in 10<sup>-15</sup> seconds. Atomic vibrations occur in 10<sup>-12</sup> to 10<sup>-13</sup> seconds.
- Neighboring atoms are frozen at some position about their equilibrium position.

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# THE EXAFS EQUATION $\chi(k) = \sum_{i} \chi_{i}(k)$ with $\chi(k) = \sum_{j} \frac{N_{j} S_{0}^{2} F_{j}(k) e^{-2k^{2} \sigma_{j}^{2}} e^{-\frac{2R}{\lambda}}}{k R_{j}^{2}} \sin[2kR_{j} + \delta_{j}(k)]$ $R_{i} = R_{0} + \Delta R$ $k^{2} = 2 m_{e}(E-E_{0})/\hbar$



Theoretically calculated values

 $F_i(k)$  effective scattering amplitude

- $\phi_i(k)$  effective scattering phase shift
- $\lambda(k)$  mean free path

#### Starting values

R<sub>0</sub> initial path length

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#### Parameters determined from a fit to data

- N<sub>i</sub> degeneracy of path
- $S_0^2$  passive electron reduction factor
- $\sigma_i^2$  mean squared displacement of half-path length
- E<sub>0</sub> energy shift
- $\Delta R$  change in half-path length





#### PT EXAFS OF REFERENCE MATERIALS

Pt EXAFS: unique information about the average local atomic environment of Pt



PtO <sub>2</sub>			
Neighbor	Number	Distance (Å)	
Pt-O	6	2.07	
Pt-Pt	6	3.10	
K <sub>2</sub> PtCl <sub>6</sub>			
Neighbor	Number	Distance (Å)	
Pt-Cl	6	2.32	
Pt-K	4	4.22	
Pt metal			
Neighbor	Number	Distance (Å)	
Pt-Pt	12	2.77	
Pt-Pt	6	3.92	

Pt-O, Pt-Cl, and Pt-Pt signals are unique and are readily distinguished.

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### PLATINUM EXAFS: COMPARISONS WITH FOIL

3 nanometer 959 atoms Pt-Pt1 CN 10.4





Figure 1. Magnitude (A) and real part of Fourier transform of 300C-reduced (symbols) and Pt foil (solid)

1.0 nanometer 43 atoms Pt-Pt1 CN 5.4



0.6 nanometer 13 atoms Pt-Pt1 CN 4.8



6 atoms Pt-Pt1 CN 4.0





4 atoms



https://iopscience.iop.org/article/10.1088/1742-6596/430/1/012061/pdf

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29



#### **EXAFS STUDY OF NB<sub>3</sub>SN SUPERCONDUCTORS** Heald S. et al., Scientific Reports 2018; Tarantini C, et al. Superconductor Science and Technology 2019

30

#### Challenge

- Nb<sub>3</sub>Sn proposed for future accelerator upgrades, but needs improved properties
- Doping can offer improvement, but optimization needs better understanding EXAFS
- Determine dopant lattice location.
- When combined with other results offered key insights into the role of dopants

#### Result

- Ti, Ta, and Hf dopants studied
- Determined Ta dopant increased antisite disorder with beneficial results
- Hf formed HfO<sub>2</sub> nanoparticle pinning sites
- Combined Ta and Hf doping offers promising route to meeting the needs of future accelerators.

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### HIGH TEMPERATURE SHOCKWAVE STABILIZED SINGLE-ATOM CATALYSTS

#### **Fourier transform comparisons**

nature nanotechnology

ARTICLES https://doi.org/10.1038/s41565-019-0518-7



Yonggang Yao<sup>18</sup>, Zhennan Huang<sup>2,8</sup>, Pengfei Xie<sup>3,8</sup>, Lianping Wu<sup>4,8</sup>, Lu Ma<sup>5,8</sup>, Tangyuan Li<sup>1,8</sup>, Zhenqian Pang<sup>4</sup>, Miaolun Jiao<sup>1</sup>, Zhiqiang Liang<sup>1</sup>, Jinlong Gao<sup>1</sup>, Yang He<sup>6</sup>, Dylan Jacob Kline<sup>7</sup>, Michael R. Zachariah<sup>7</sup>, Chongmin Wang<sup>6</sup>, Jun Lu<sup>5</sup>, Tianpin Wu<sup>5</sup>\*, Teng Li<sup>4</sup>\*, Chao Wang<sup>3</sup>\*, Reza Shahbazian-Yassar<sup>2</sup>\* and Liangbing Hu<sup>11</sup>

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31

## STRUCTURE OF LUMINESCENT PROTEIN-STABILIZED GOLD CLUSTERS

#### **EXAFS with DFT interpretation**



https://doi.org/10.1039/C7SC05086K

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