

Introduction to Small Angle X-ray Scattering for Nanomaterials

Tao Li (tli4@niu.edu; taoli@anl.gov)

Chemistry Department of Northern Illinois University

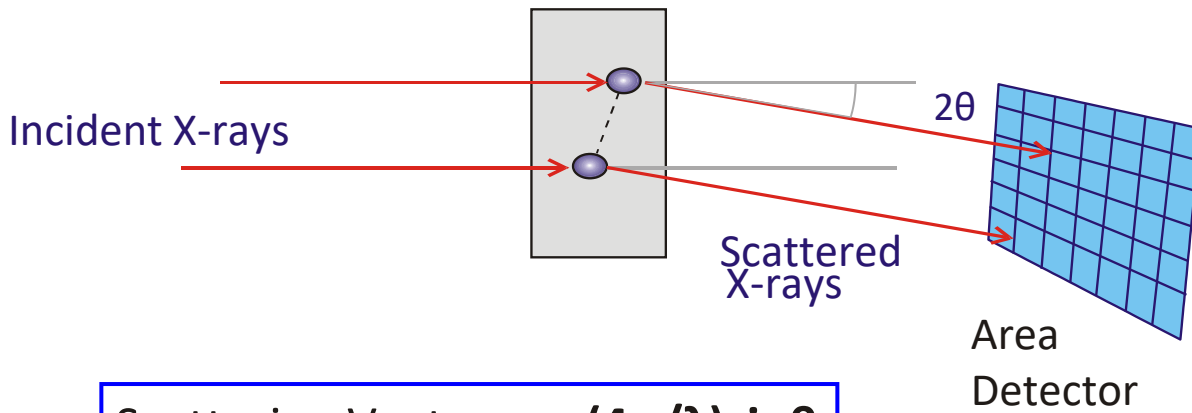
Advanced Photon Source of Argonne National Laboratory (joint)

Thanks to Xiaobing Zuo, Byongdu Lee, Pete Jemian, Jan Ilavsky, Randall Winans for some slides.

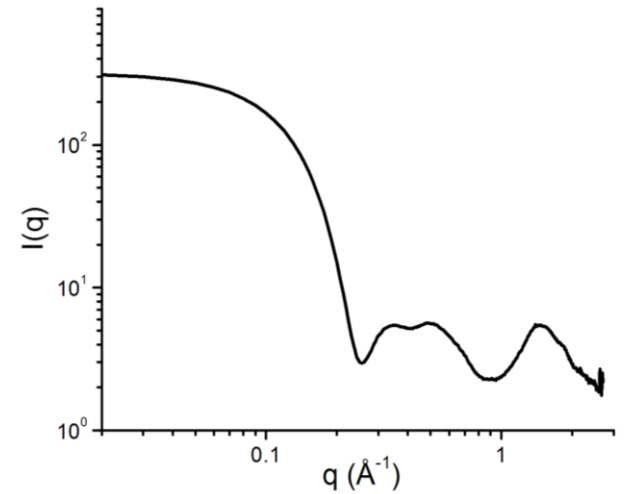
Outline

- **Introduction**
- **Experimental Setups**
- **Fundamentals of X-ray Scattering**
- **Theory and Applications of Small Angle X-ray Scattering**
- **Applications**

What Does Small Angle Scattering Measure?

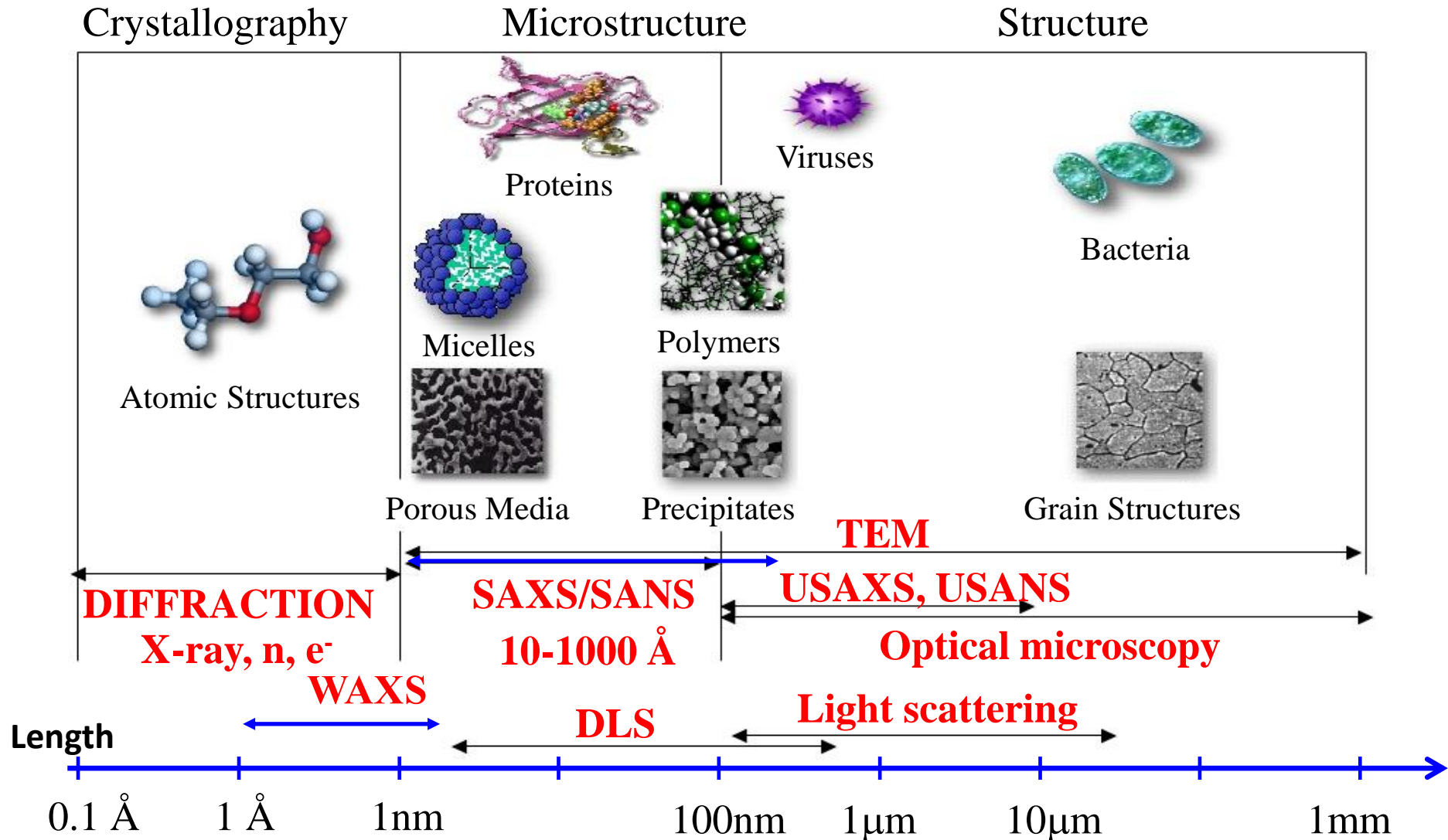


$$\text{Scattering Vector: } \mathbf{q} = (4\pi/\lambda)\sin\theta$$



- Measuring the size range from 1 to 100 nm
- Size and distribution, shape, particle distance, compositions, volume fractions

Length Scales Probed by SAS and Other Characterization Techniques



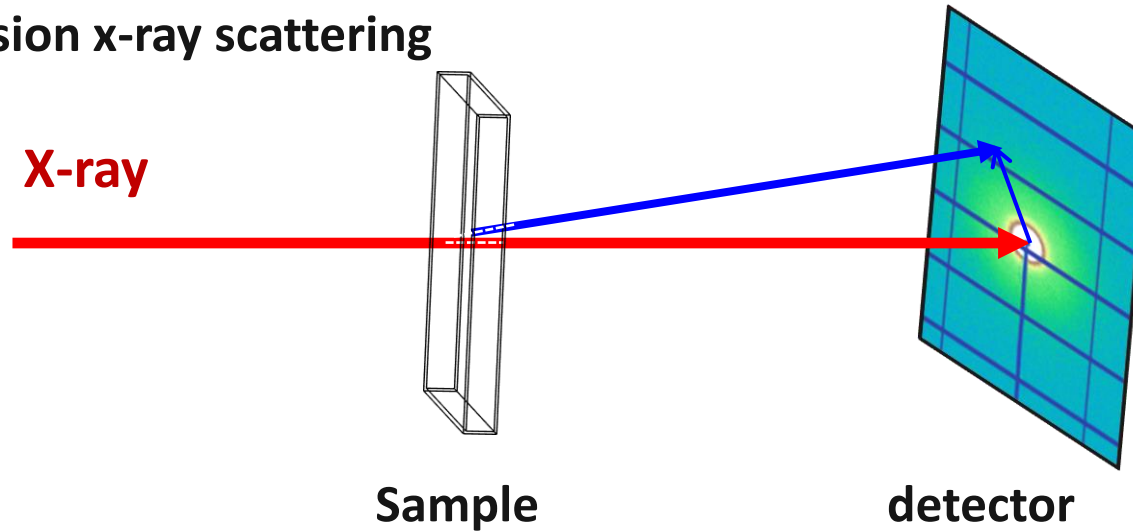
1. X-ray Scattering Setup

- **Storage-Ring synchrotron**
- **X-ray scattering setup**

X-ray Scattering Setup Configuration

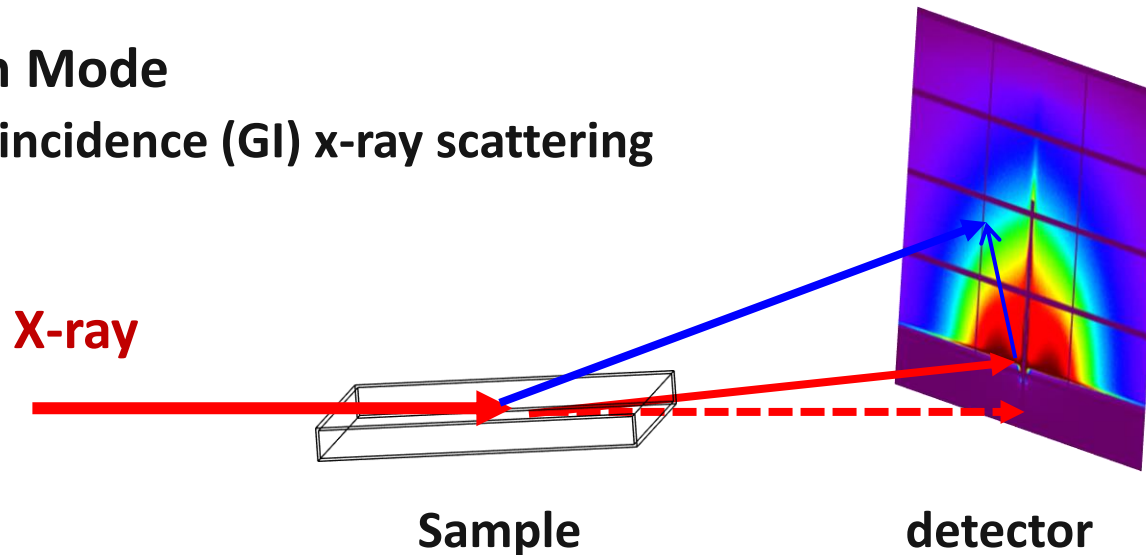
Transmission Mode

Transmission x-ray scattering



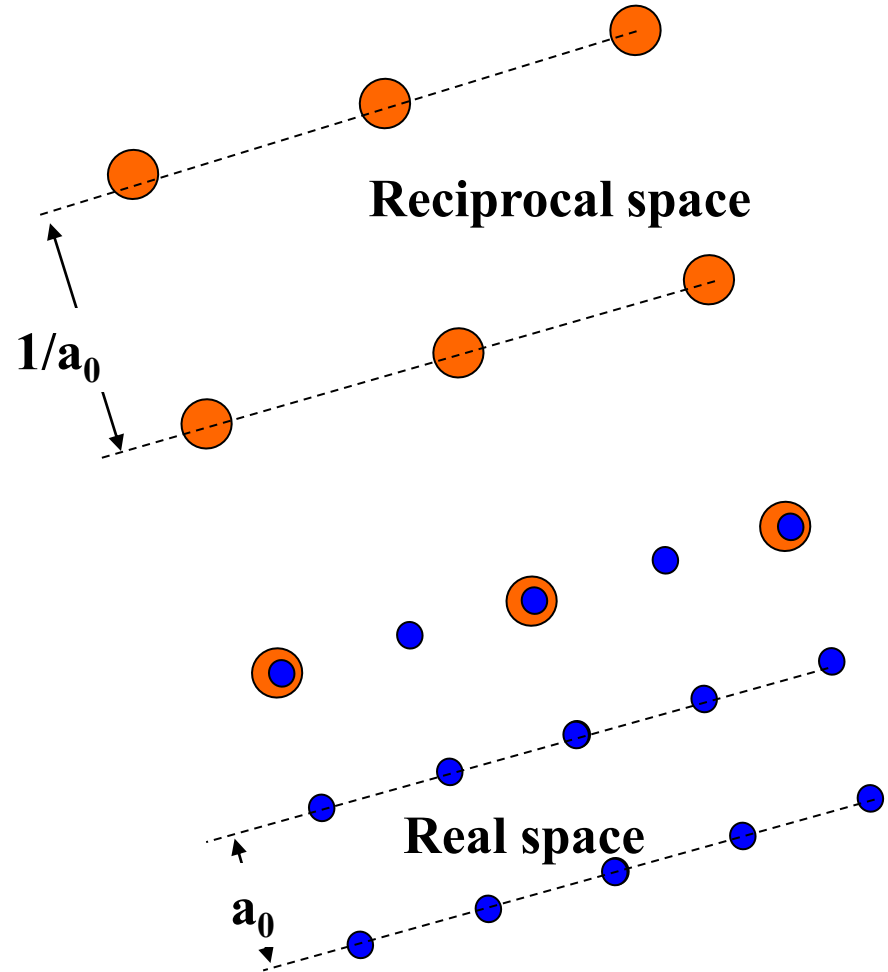
Reflection Mode

Grazing incidence (GI) x-ray scattering

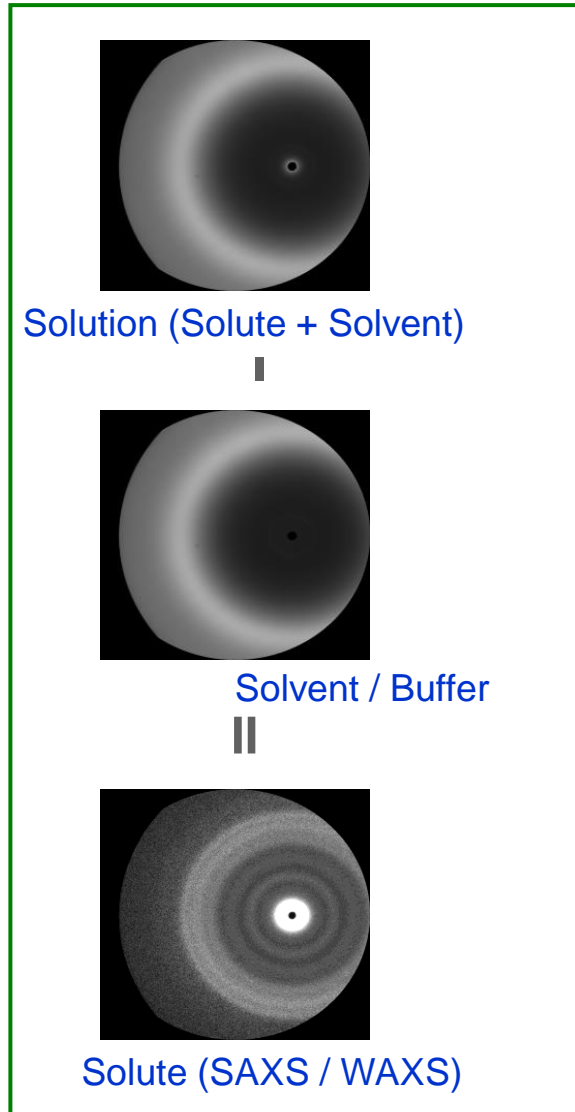


Real space vs Reciprocal space

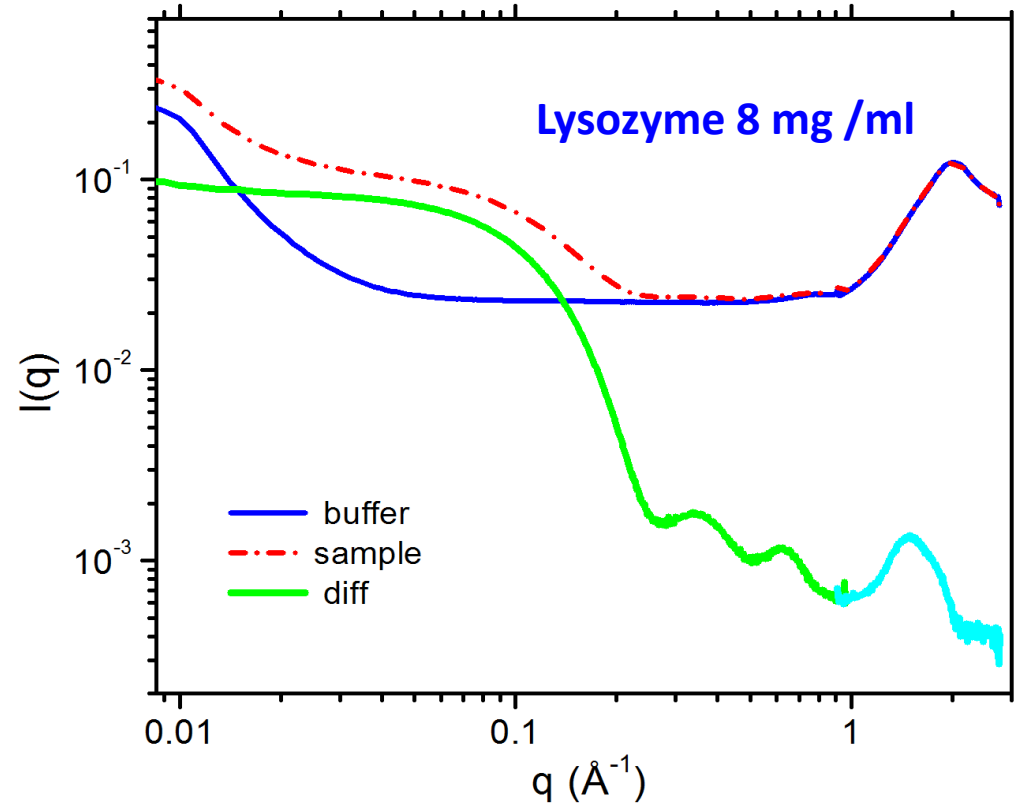
- Properties
- Unit in length $^{-1}$
- Length in reciprocal space is $1/\text{length}$ in real space
- Volume of reciprocal space is $1/\text{volume}$ in real space



(Solution) X-ray Scattering Data



1D SAXS profiles



When do you need synchrotron x-ray source?

1. High flux: high background exp
2. Small beam size: small amount of sample, etc
3. Fast data collection: in-situ, fast kinetics

Lab source vs SR

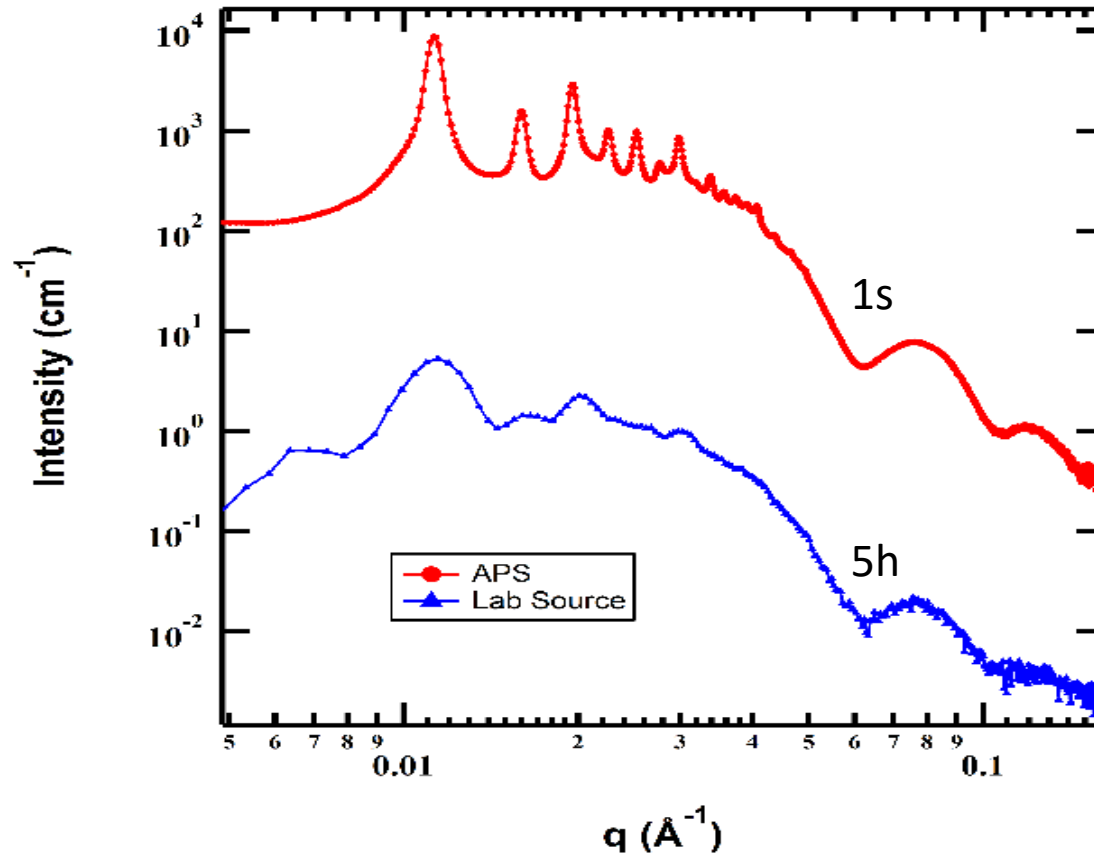
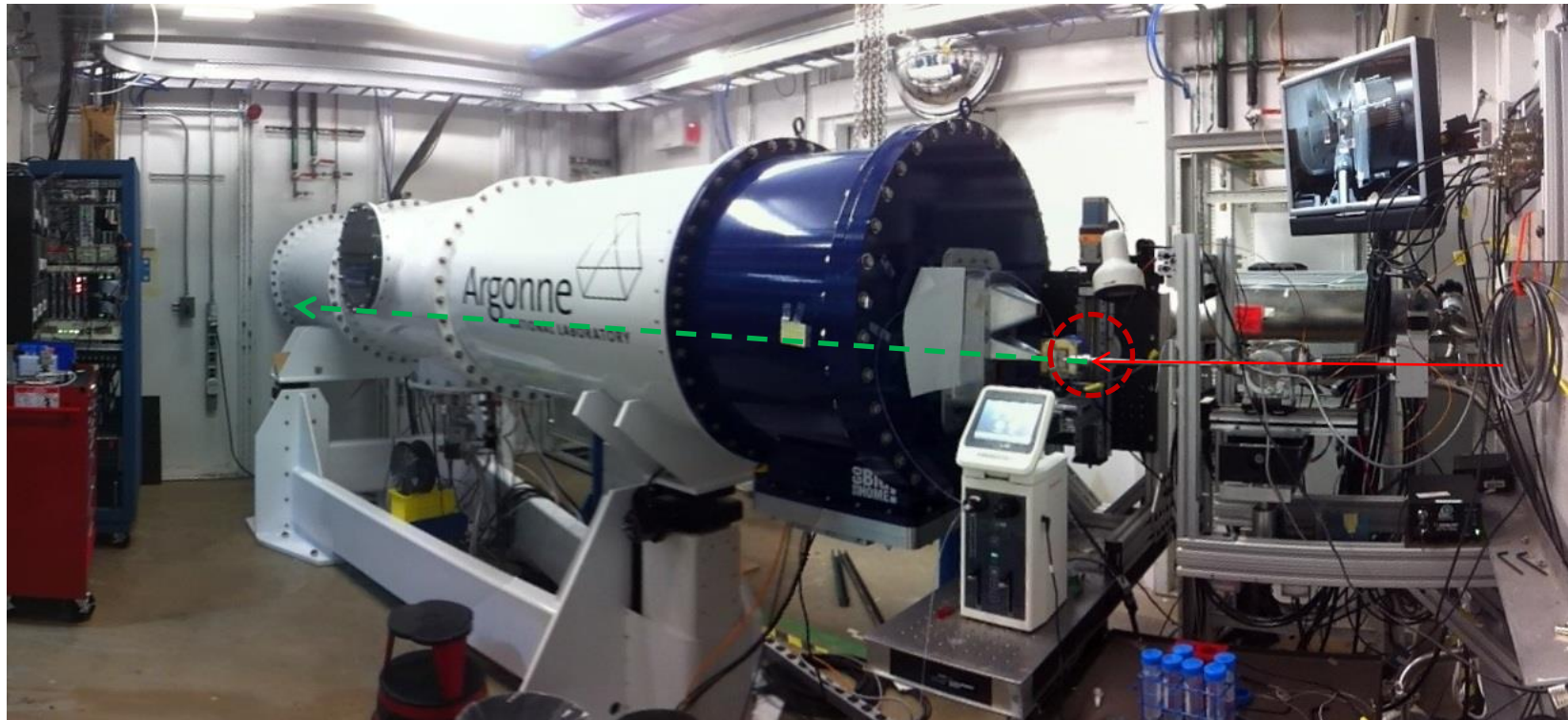


Figure 3. SAXS data of BCC SL of 15nm gold spherical particles linked with DNA. The same sample were measured with APS and Lab sources.

New SAXS/WAXS Setup at Beamline 12ID-B of Argonne



- From 1 to 150 nm
- High flux, one measurement less than 0.1 s
- In situ SAXS study : high temperature (up to 1500°C) and high pressure (up to 20,000 psi)
- Element specific information from ASAXS
- Can be combined with many other techniques such as IR and other spectroscopy

Fundamentals of X-ray Scattering

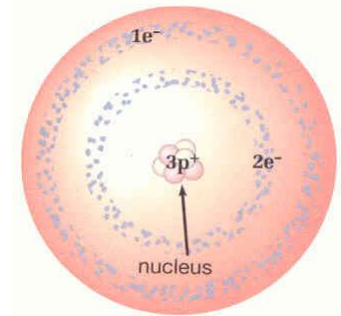
- **X-ray scattering and interference**
- **Form factor: size & shape**
- **Size polydispersity**
- **Structure factor**

Atomic Form Factor

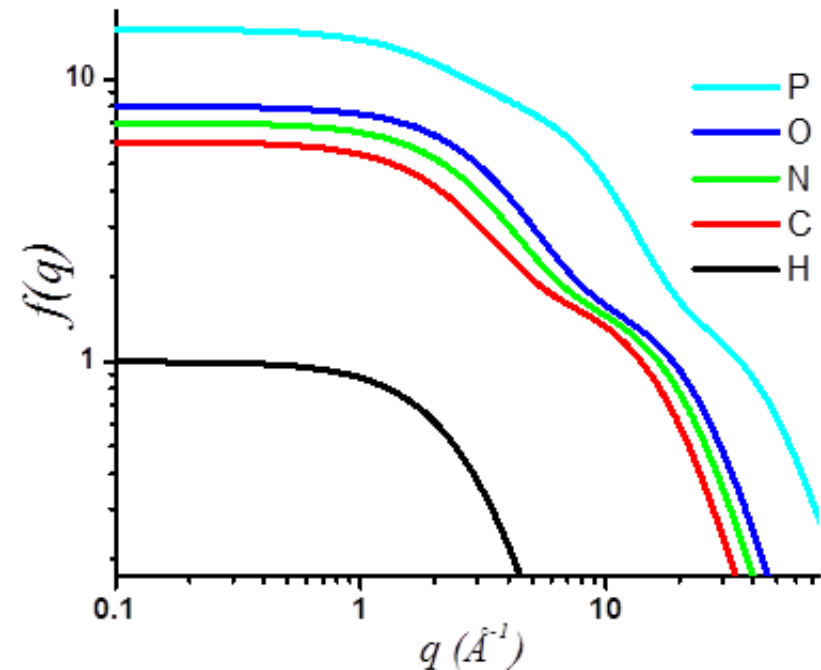
Electron cloud in atoms has radial density distribution $\rho(r)$

$$f(q) = 4\pi \int \rho(r) r^2 \frac{\sin(qr)}{qr} dr$$

- Atomic form factors are fundamental parameters for X-ray techniques.
- $f(0)=Z$: the total electron of the atom, scattering length.
- Atoms with higher Z will scatter stronger.



Atomic electron cloud

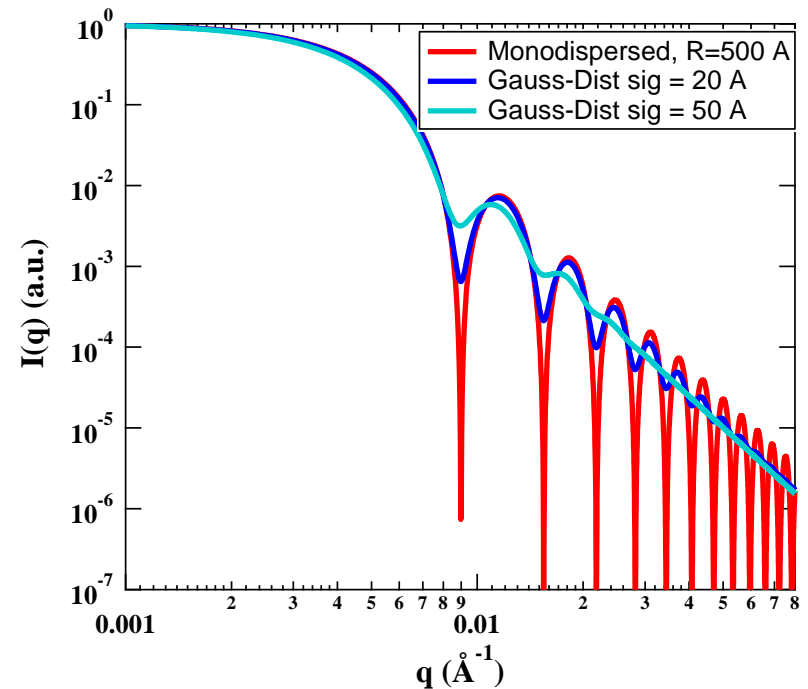
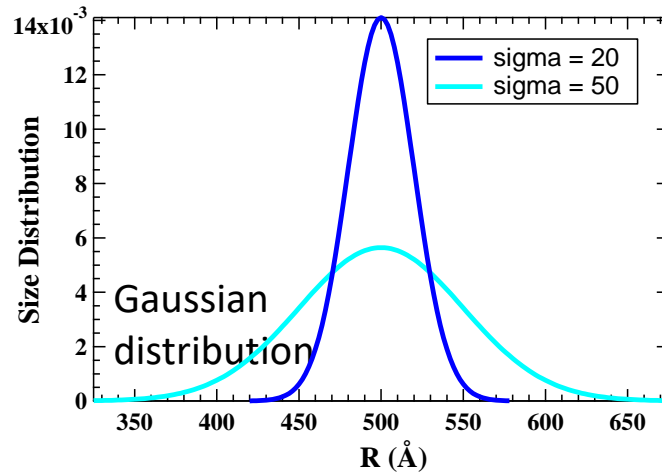
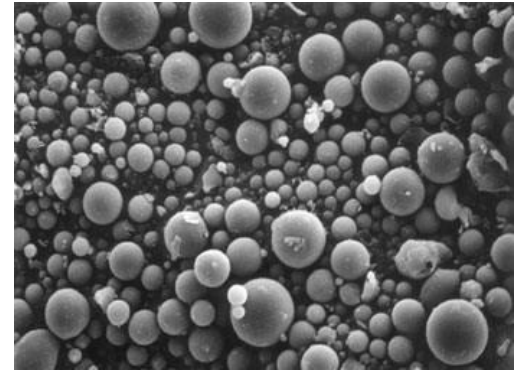


Data taken from **International Tables for Crystallography**, Vol. C, Table 6.1.1.1

Polydispersity: Size Distribution

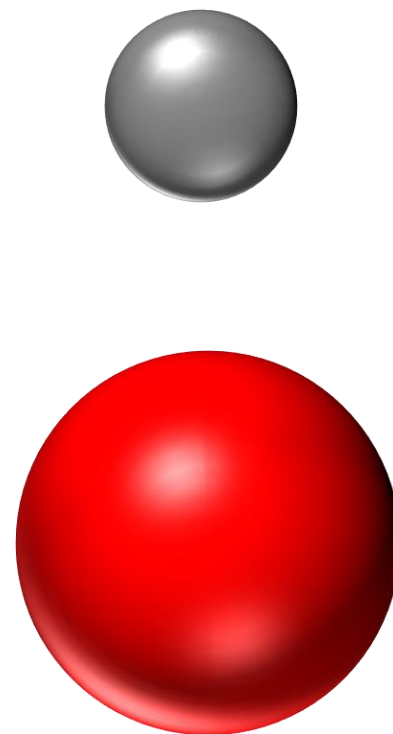
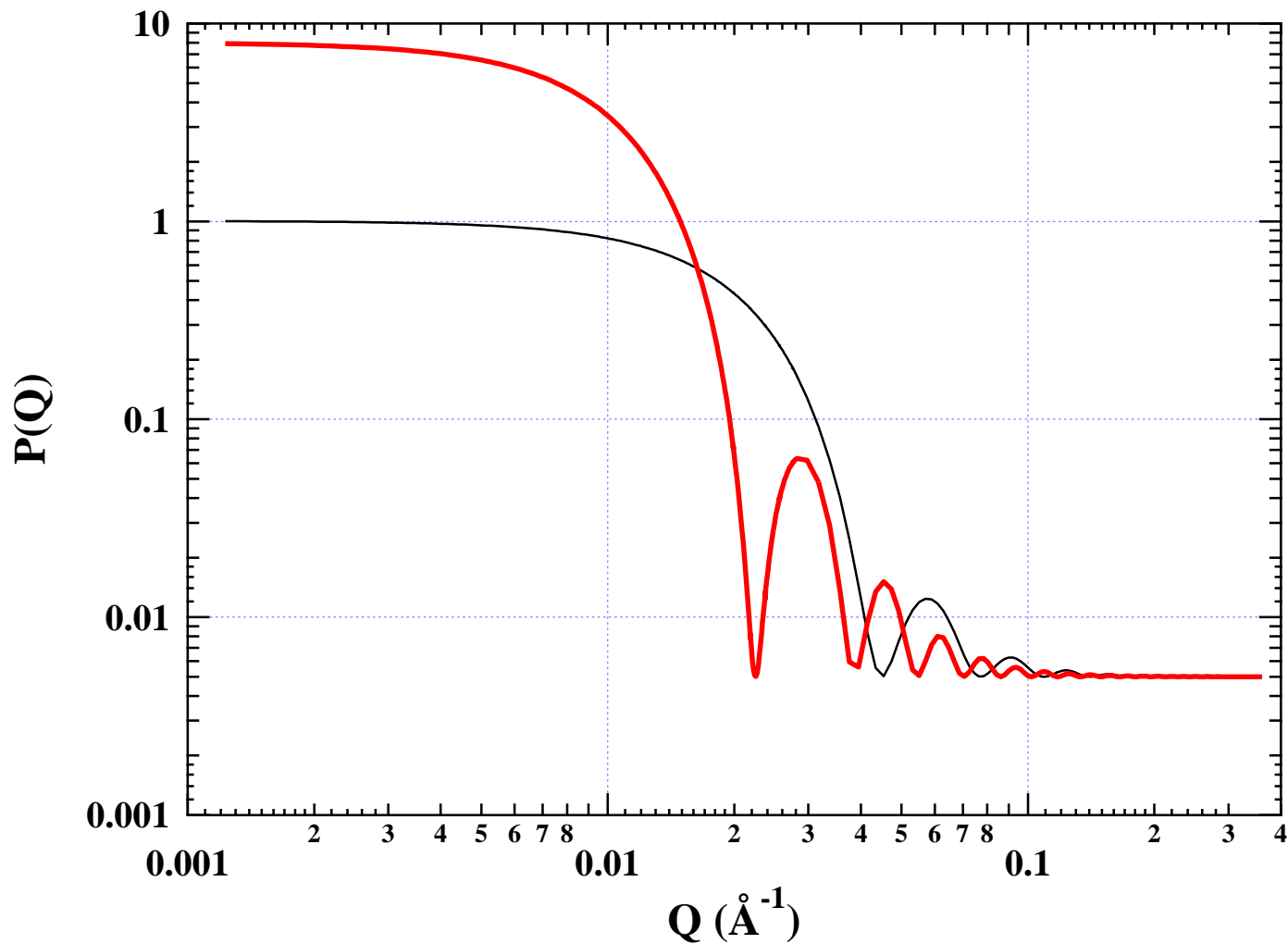
Scattering for an ensemble with different sizes:

Sphere radii with Gaussian distribution

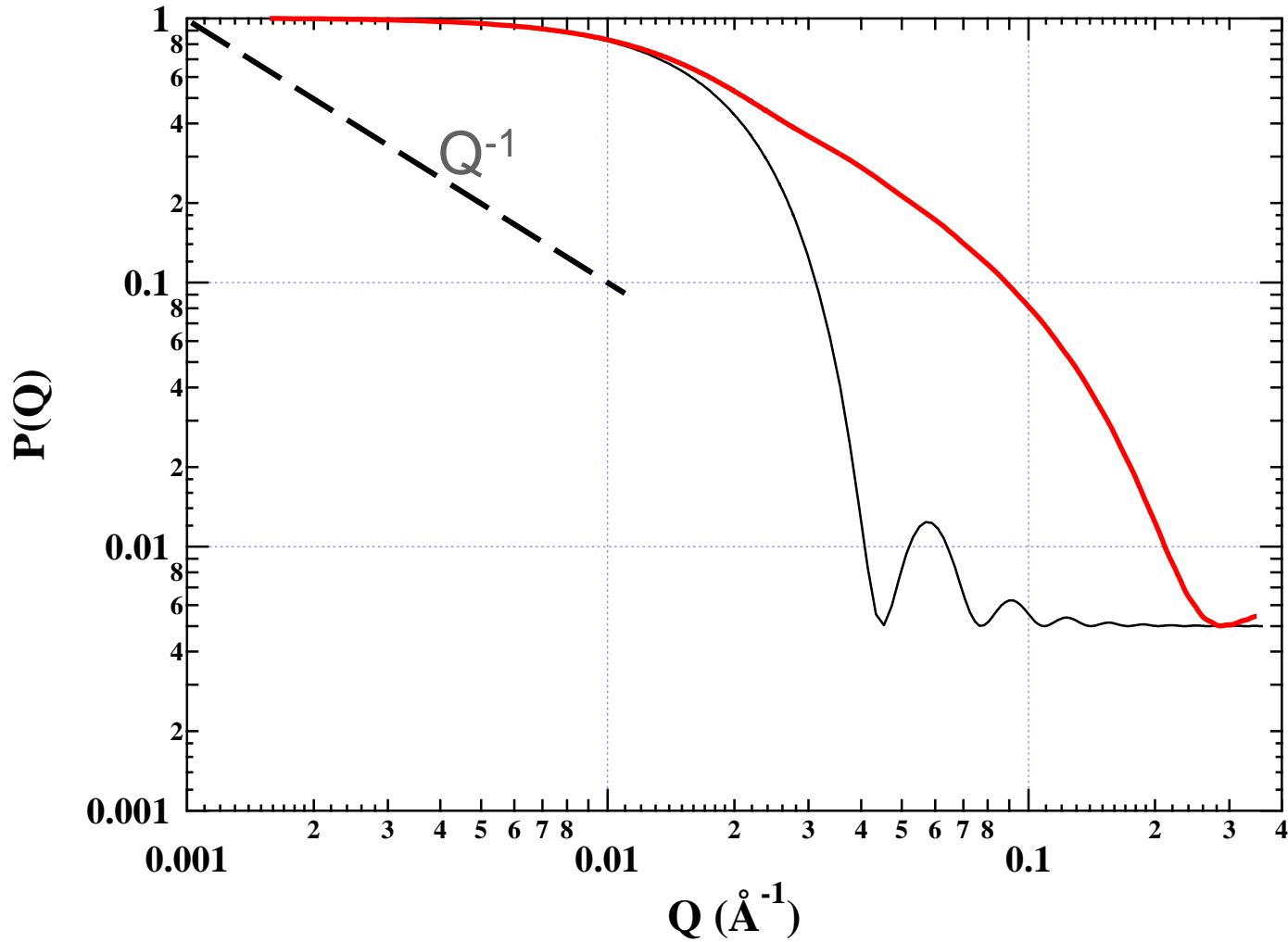


- Size polydispersity smears/dampens fine features in scattering profile.

Spheres of different sizes



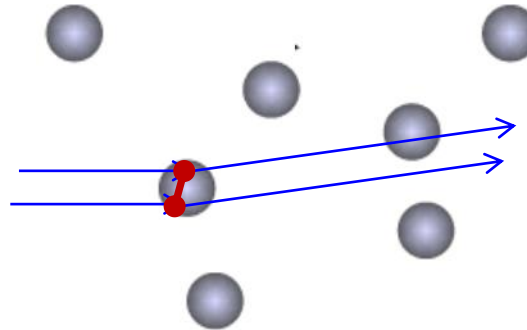
“Long & thin” cylinder



Particle Correlation: Structure Factor

Dilute, randomly distributed particles:

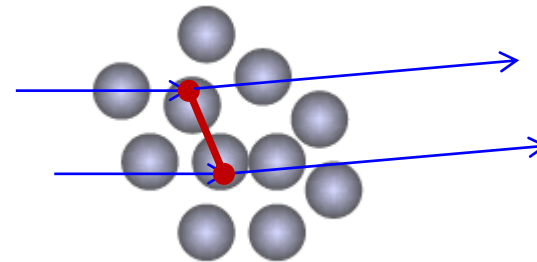
$$I(q) = NP(q)$$



Correlated particles:

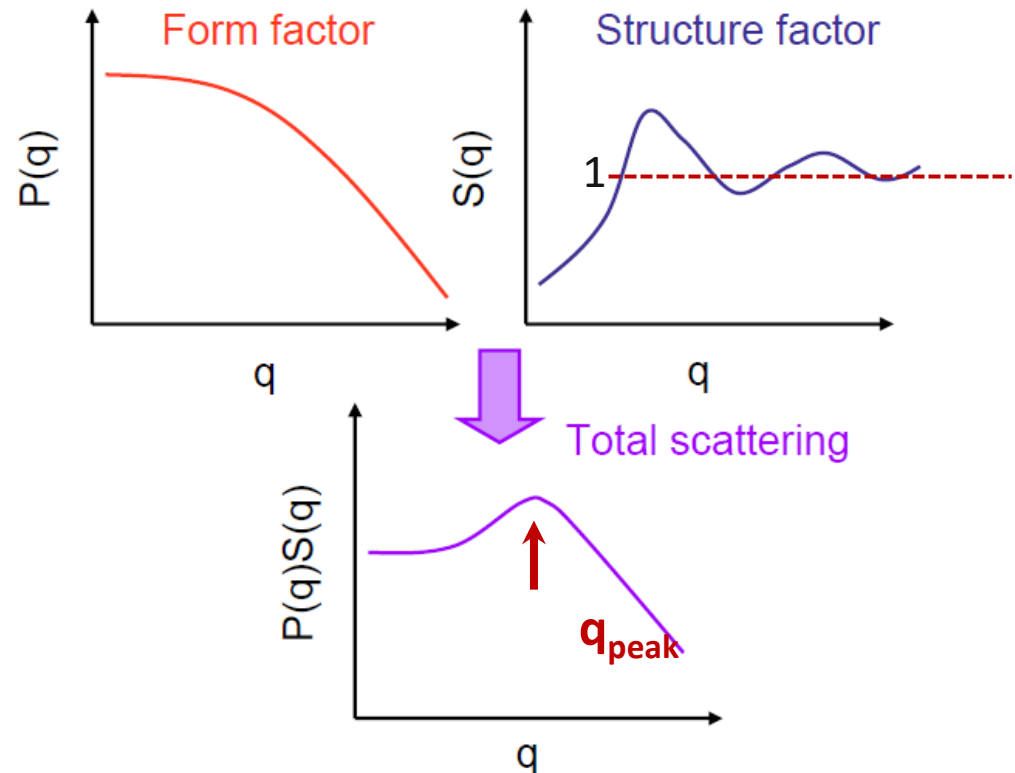
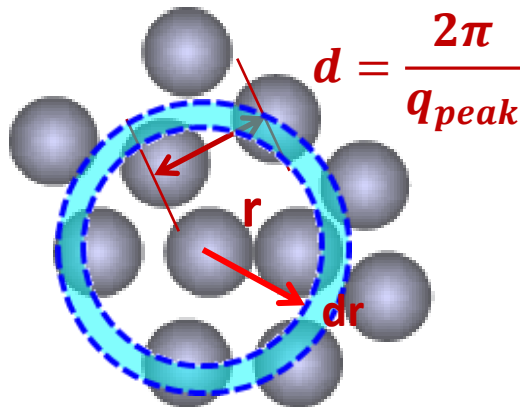
$$I(q) = NP(q)S(q)$$

$S(q)$ structure factor



Structure Factor: Common Spacing(s) between Scatterers

$$S(q) = 1 + 4\pi \frac{N}{V} \int_0^\infty r^2 [g(r) - 1] \frac{\sin qr}{qr} dr$$



- $g(r)$: radial particle distribution function
- Low concentration, $S(q)=1$
- Higher concentration, $S(q)$ oscillates about 1

$$I(q) = NP(q)S(q)$$

Theory and Applications of SAXS

- **Guinier Approximation**
- **Porod Law**
- **Invariant**
- **Hierarchical structural Information**

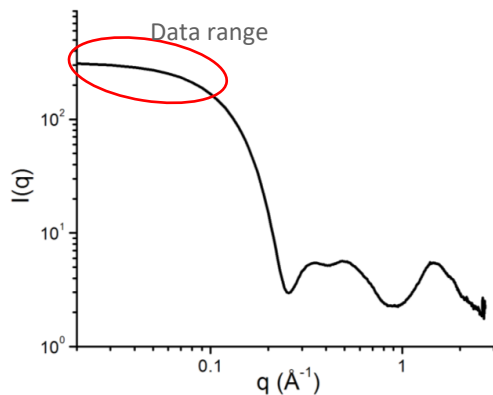
Guinier Equation

When $q \rightarrow 0$,

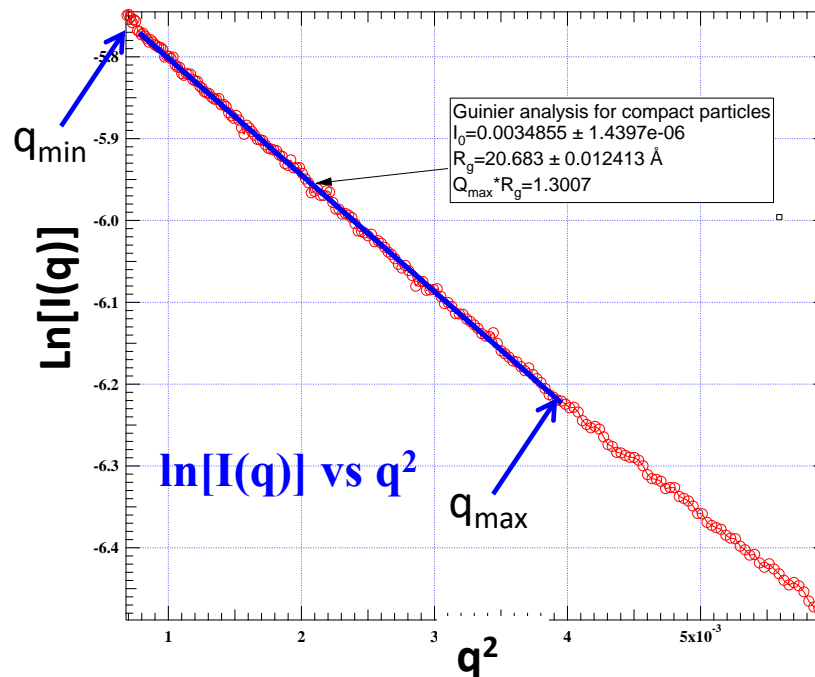
$$I(q) \cong I(0) \exp\left(\frac{-R_g^2 q^2}{3}\right)$$

R_g : radius of gyration

$I(0)$: forward scattering



André Guinier (1911-2000)



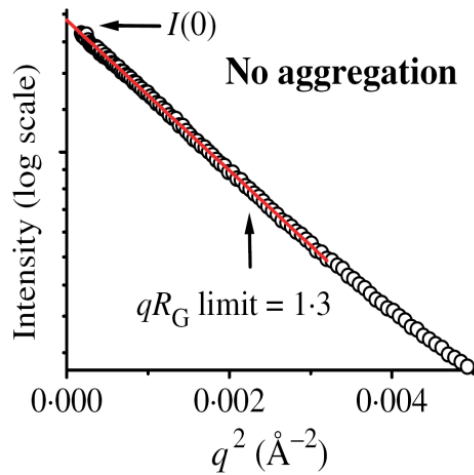
To get reliable Guinier plot / R_g analysis:

- $q_{\max} * R_g < 1.3$ for globular; < 0.8 for elongate
- $q_{\min} \leq \pi / D_{\max}$
- Multiple (≥ 5 ???) data points in linear fashion

Guinier Plot: Data Evaluation & Sample Condition

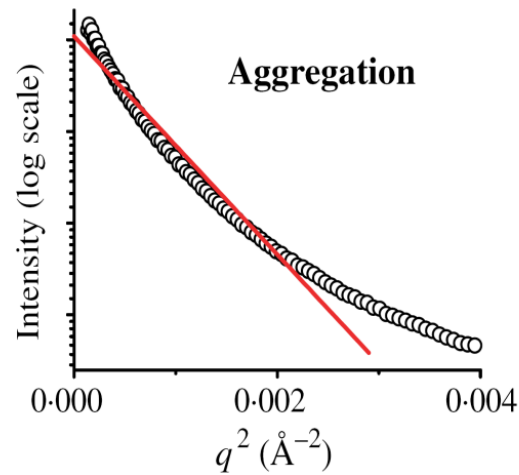
Mono-dispersed

Normal / linear



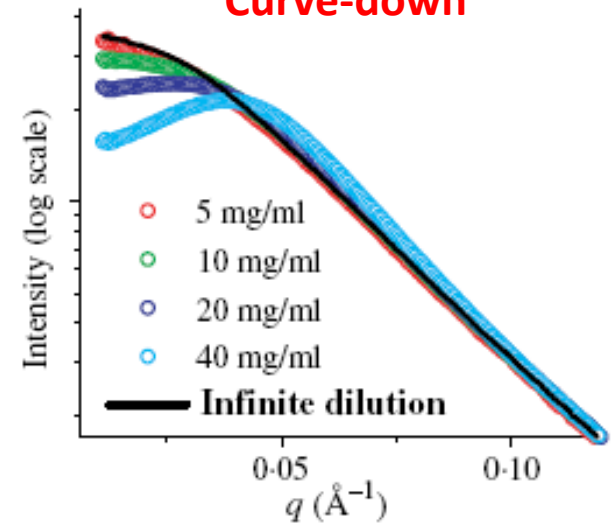
Poly-dispersed
aggregates

Curve-up



Repulsion

Curve-down



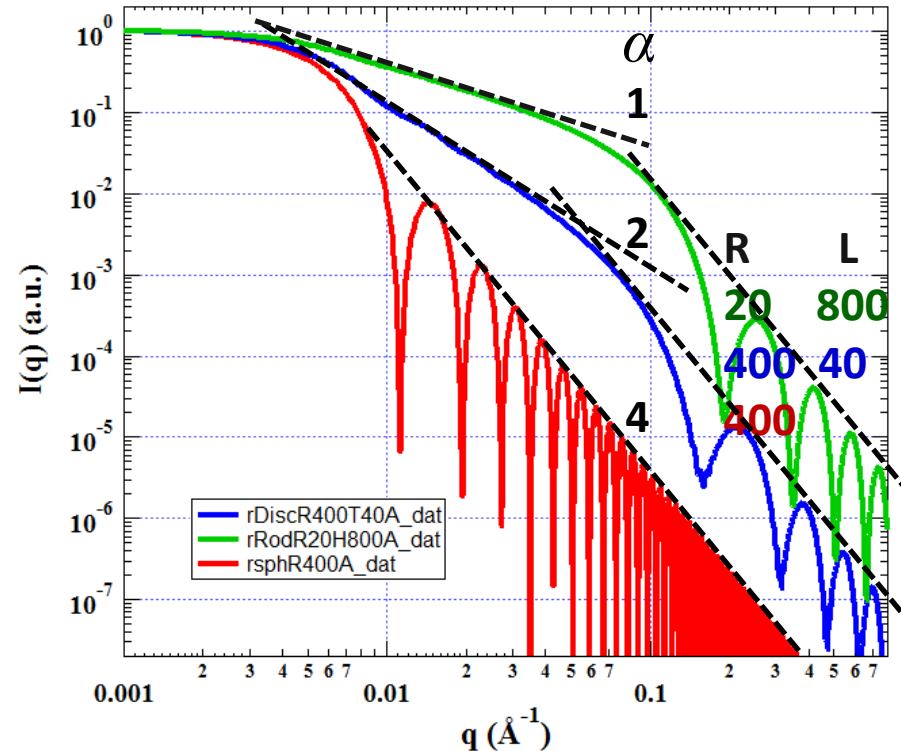
Putnam, D., et al. (2007) Quart. Rev. Biophys. 40, 191-285.

Porod Law

Generalized form:

$$I(q) \propto q^{-\alpha}$$

$\alpha =$ 1 rod-like
2 lamellar/disc
4 sphere
fraction: fractals



- Can provide morphology information
- May not be valid in atomic length region
- Could be misled by inaccurate background subtraction

Porod Invariant and Porod Volume

Porod Invariant Q :

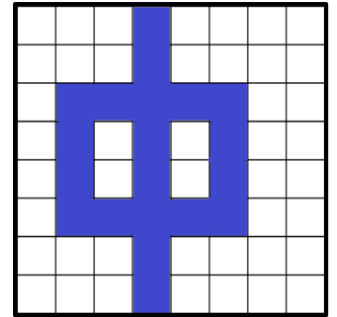
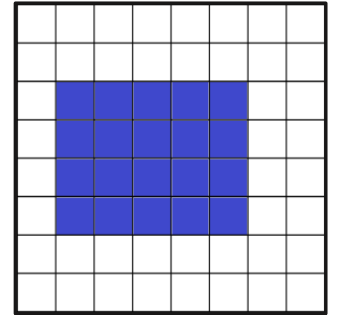
$$Q \equiv \int_0^\infty q^2 I(q) dq = 2\pi^2 (\Delta\rho)^2 V$$

For uniform particles:

$$I(0) = (\Delta\rho V)^2$$

Porod volume:

$$V = \frac{2\pi^2 I(0)}{Q}$$

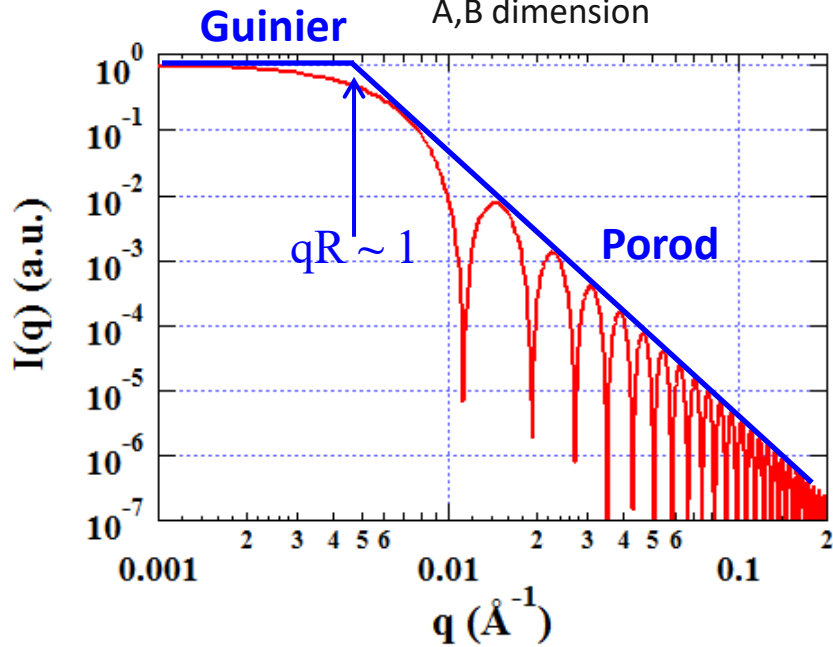


- The invariant measures the total electrons, does not depend on morphology.
- The volume of a molecule can be estimated solely from scattering data.
- Calculation of particle volume does not require absolute data scaling.

Anatomy of SAXS Profile

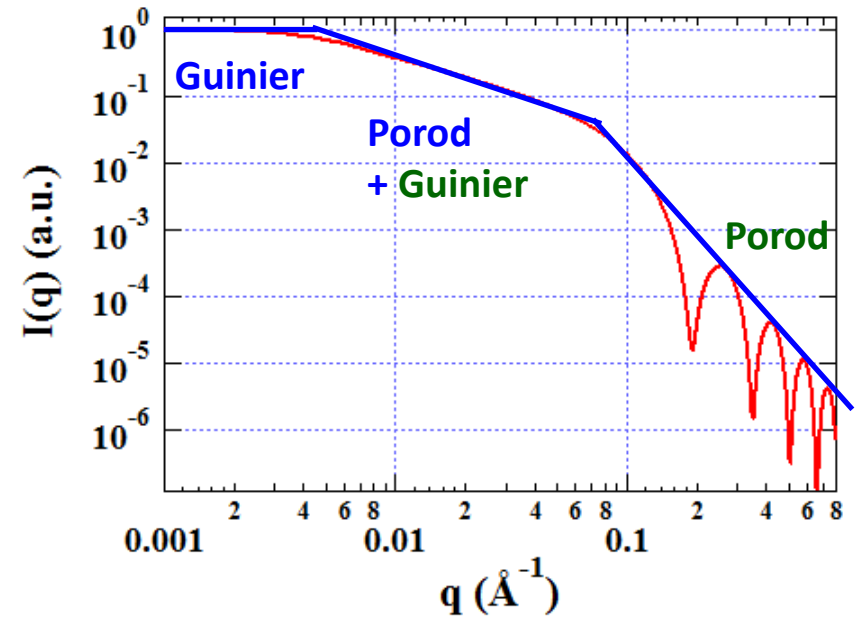


One characteristic length:
 all dimensions similar
 aspect ratio: $d_{\max,A}/d_{\max,B} \sim 1$
 A,B dimension



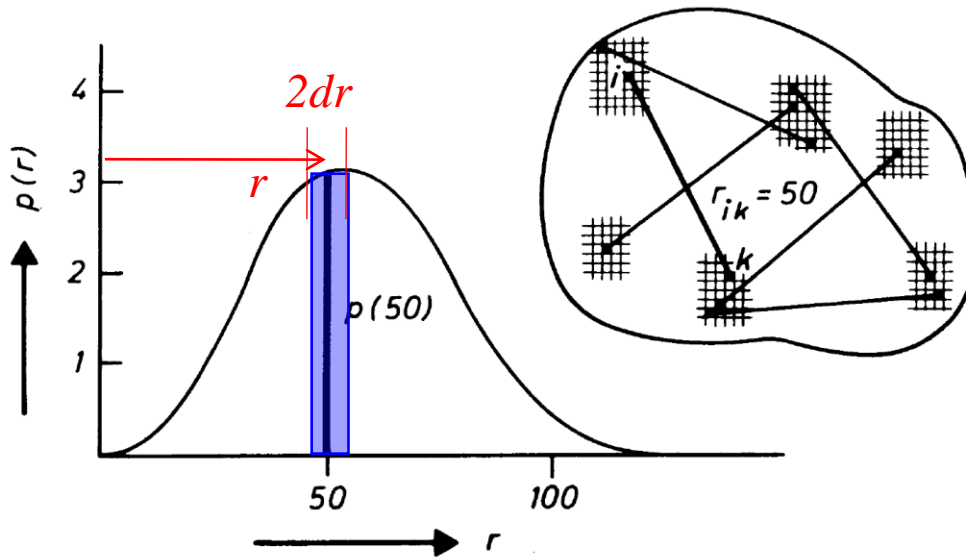
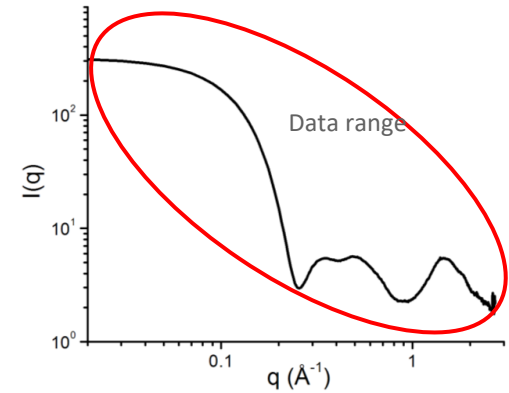
$R=20 \text{ \AA}, L = 800 \text{ \AA}$

Two characteristic lengths:
 $L \gg 2R$



Pair Distance Distribution Function (PDDF)

$$p(r) = \frac{r^2}{2\pi^2} \int_0^\infty q^2 I(q) \frac{\sin qr}{qr} dq \quad \text{--- reverse FT of } I(q)$$

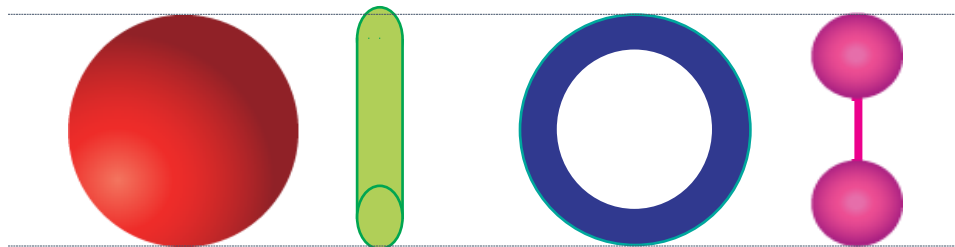
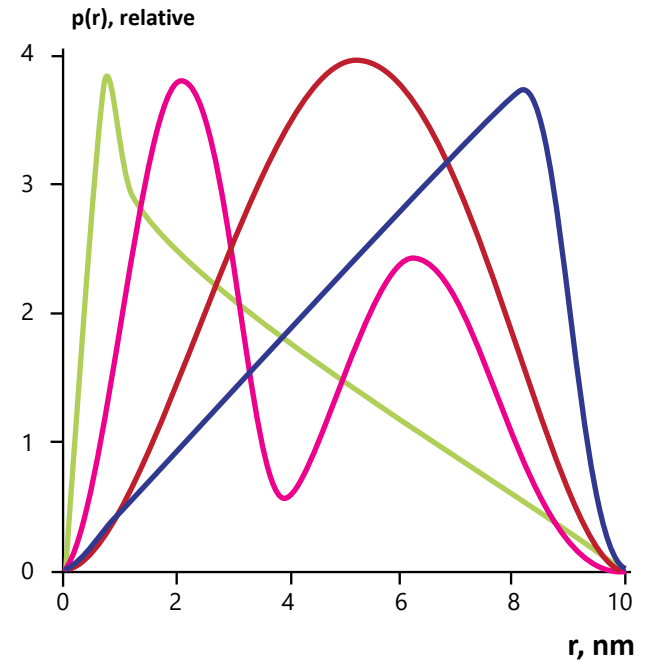
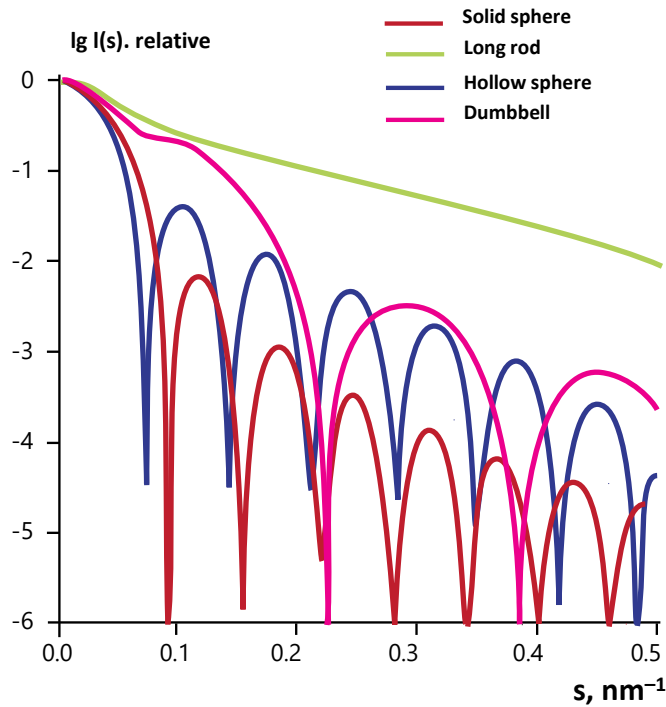


$$p(r) \sim \sum_{|\vec{r}_j - \vec{r}_k| < r + dr} 1 \times \Delta n(\vec{r}_j) \times \Delta n(\vec{r}_k) \times r^2$$

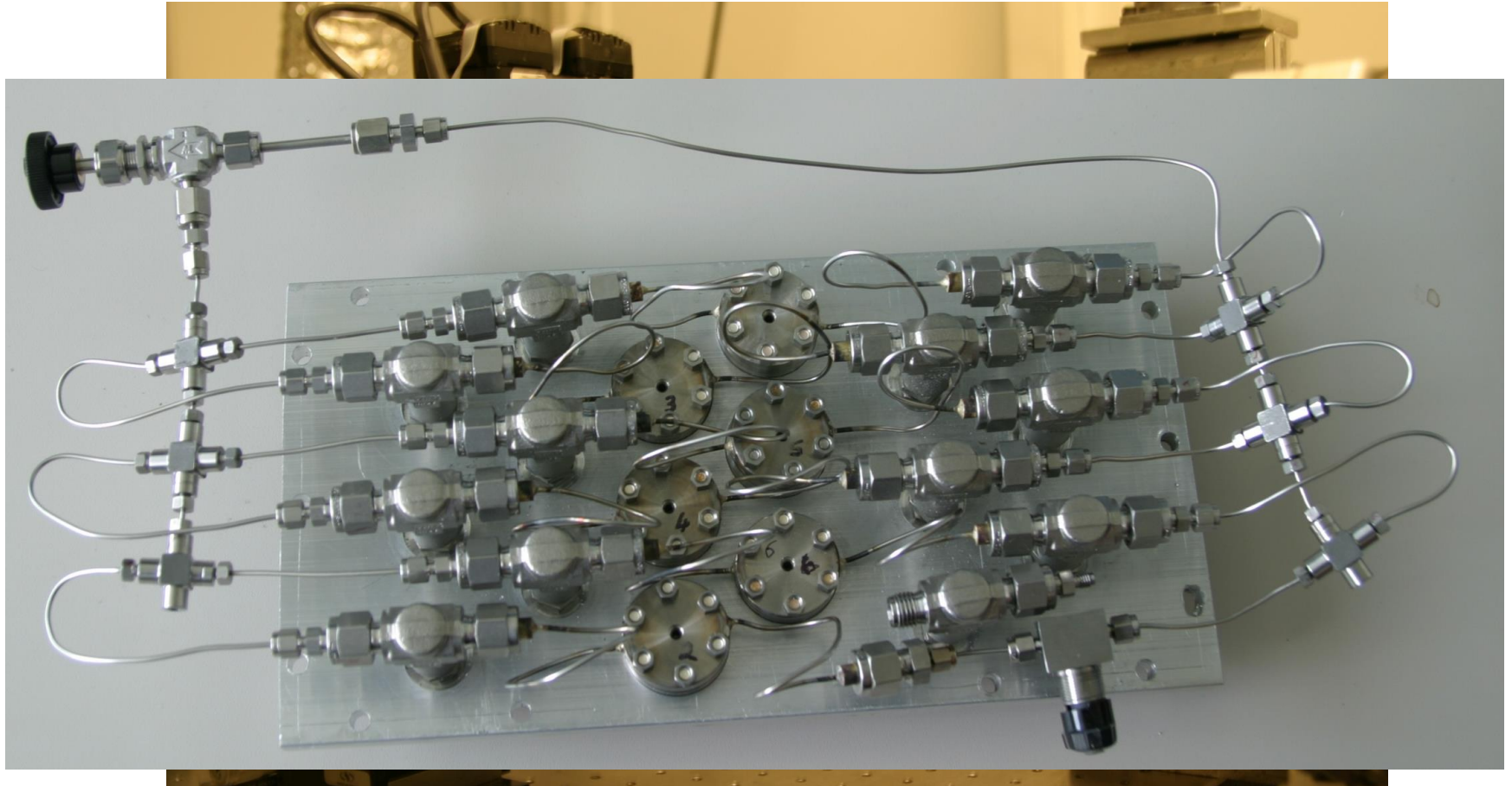
↑
excess electrons of
atom j over solvent

- The PDDF of a molecule is the (net-electrons and distance) weighted atom-pair distance histogram.

Combination of SAXS & PDDF for Shape Determination



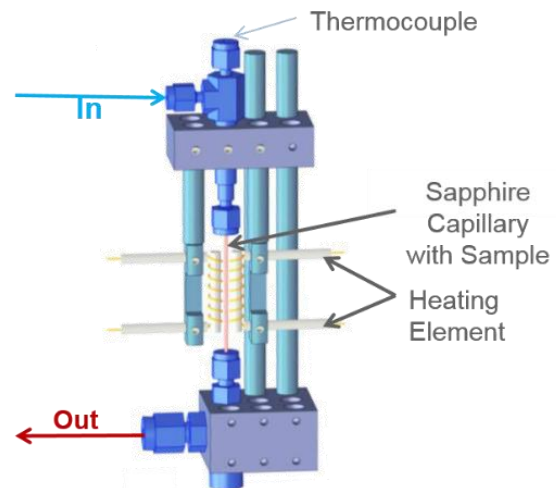
SAXS Instruments at 12ID



Setup



Heating up to 1500 °C

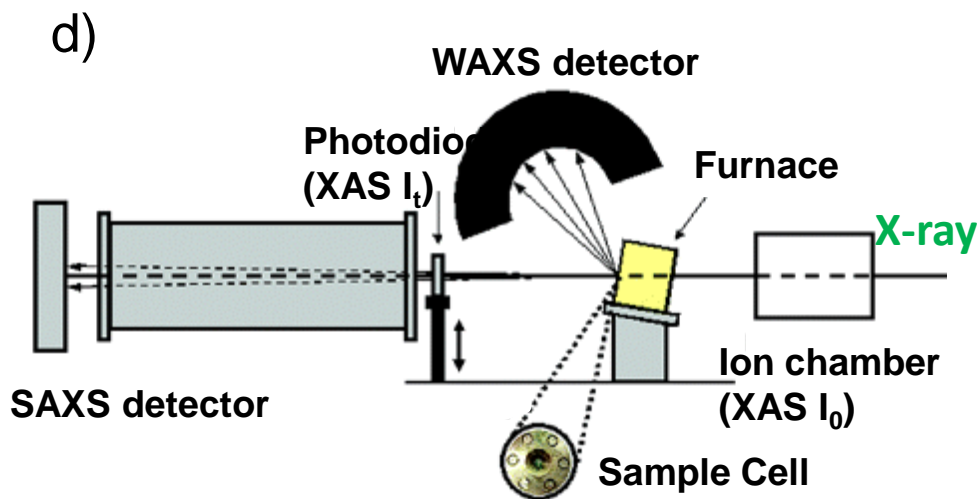
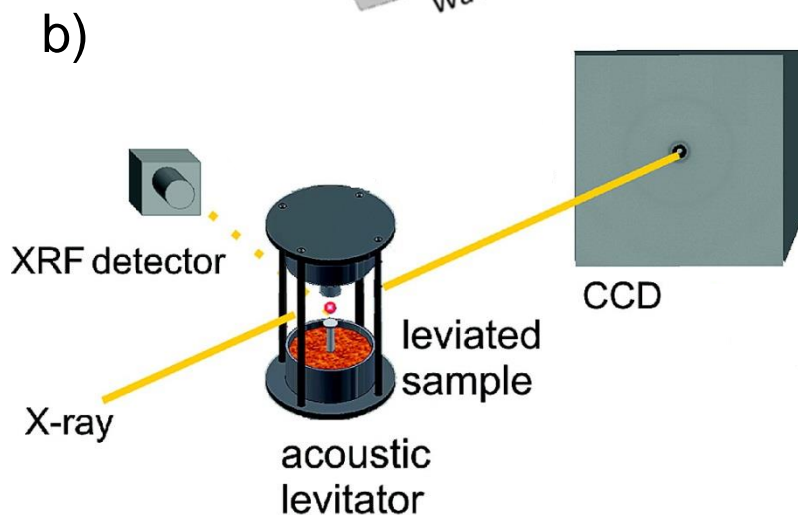
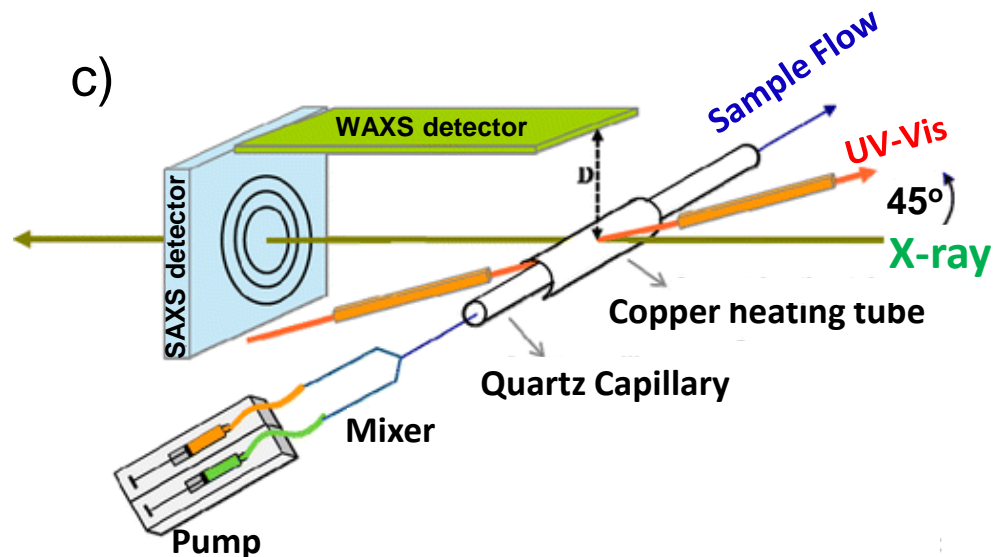
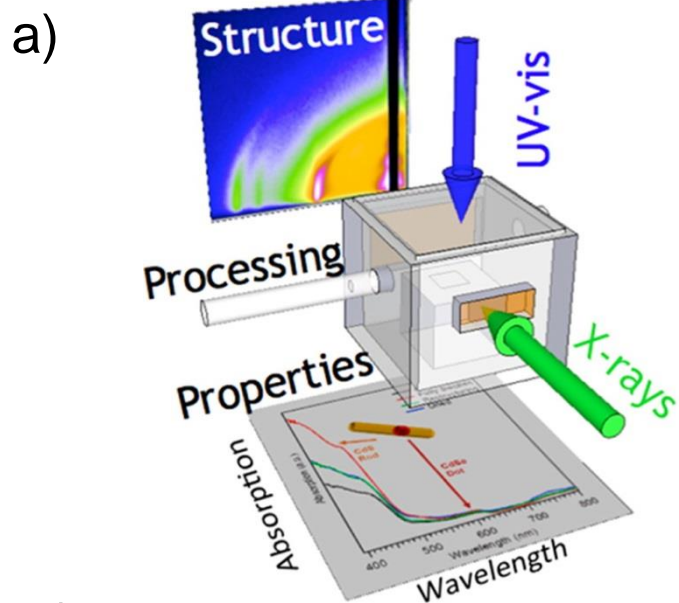


Heated capillary flow cell



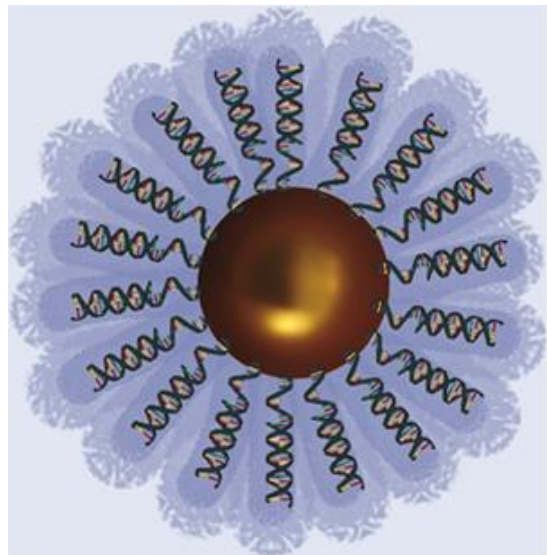
GISAXS cell for high temperature and pressure reactions

Combined SAXS with other technique

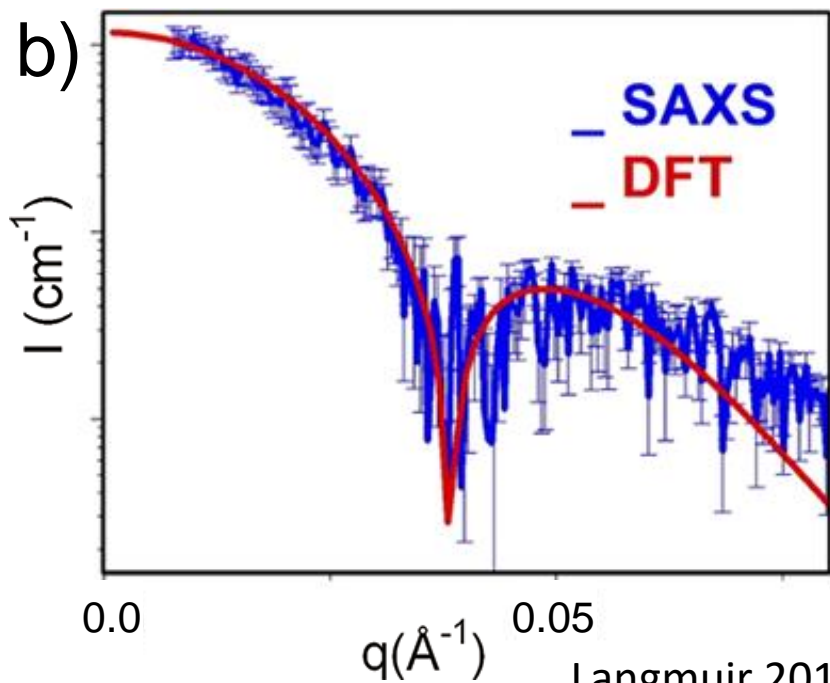


SAXS and SANS

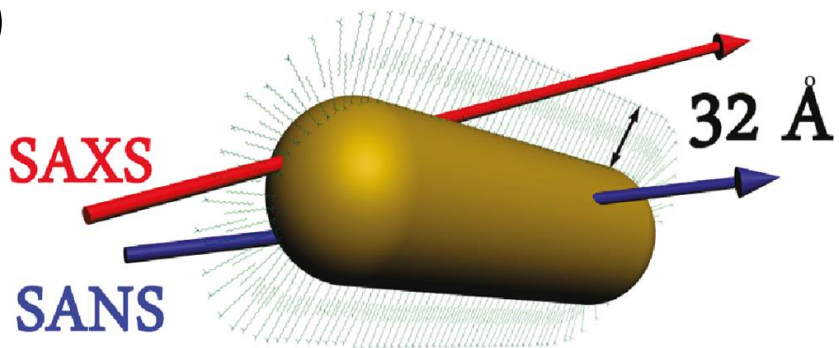
a)



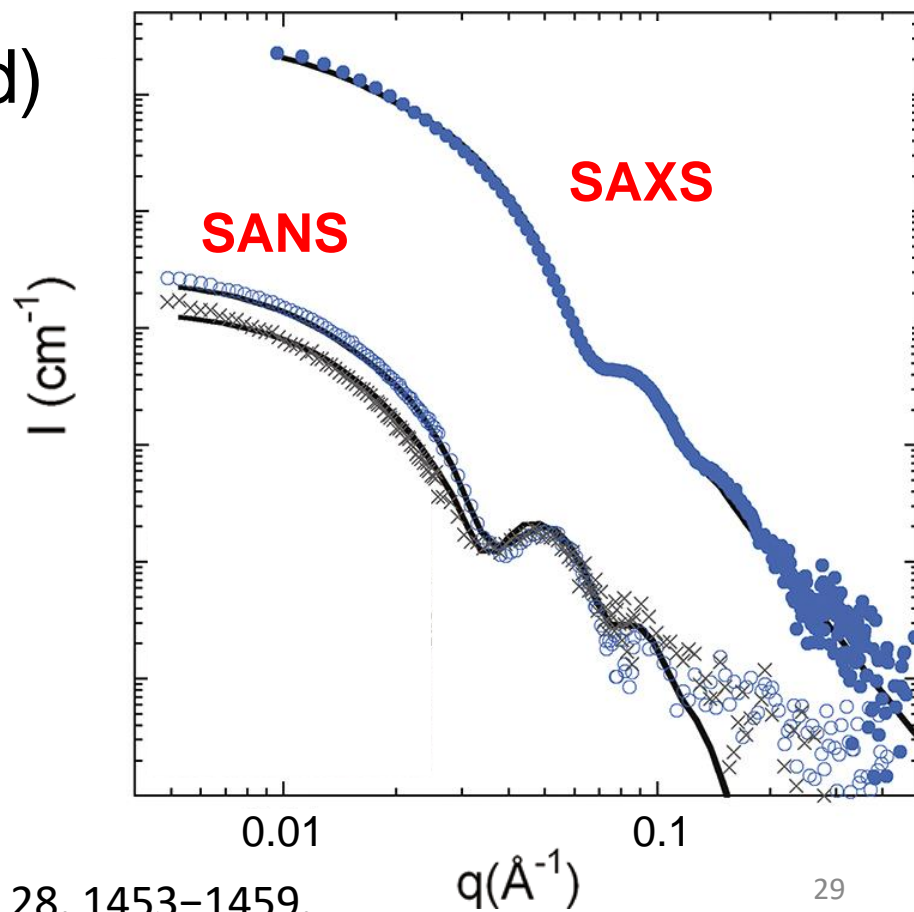
b)



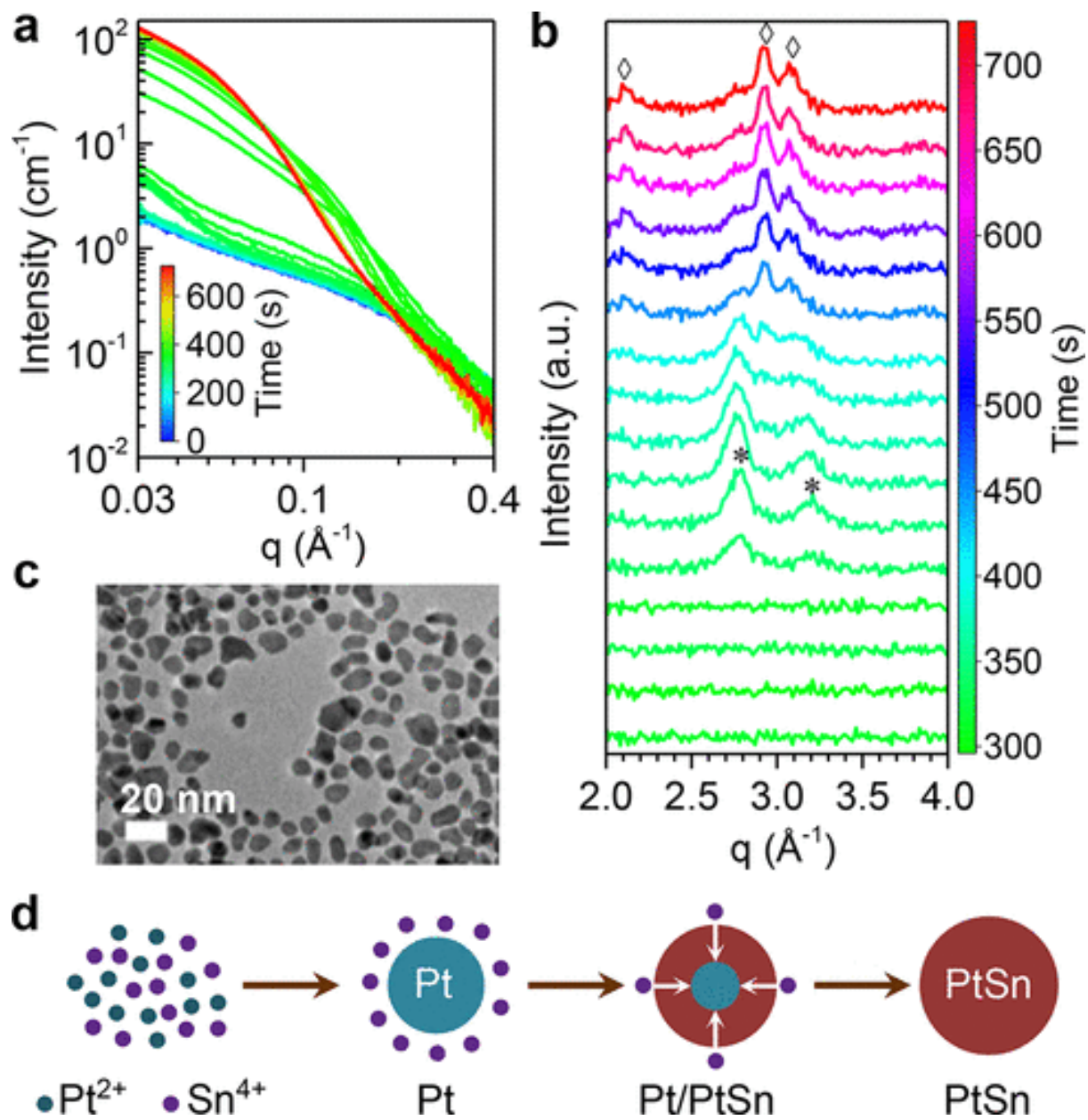
c)



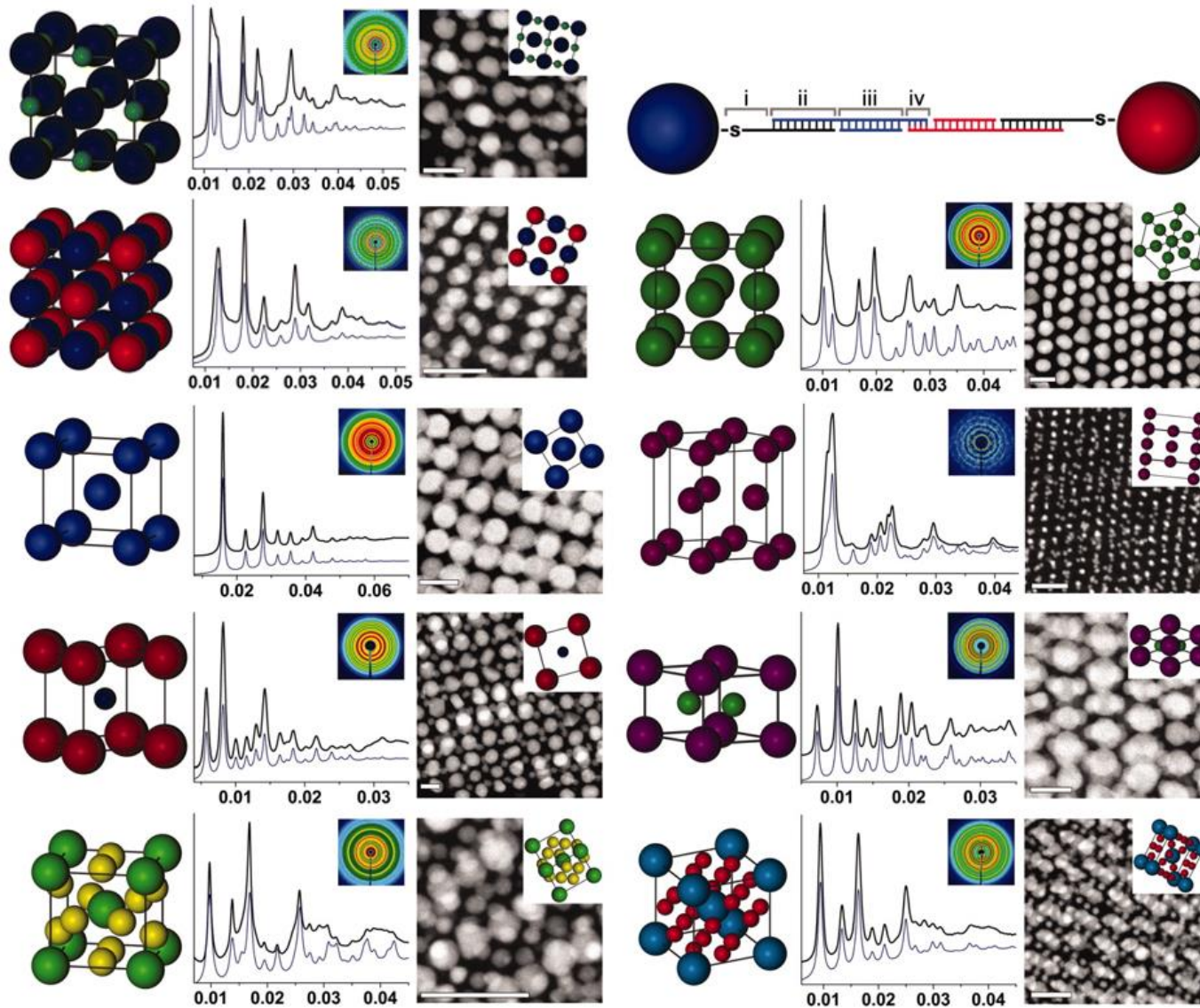
d)



In situ SAXS/WAXS

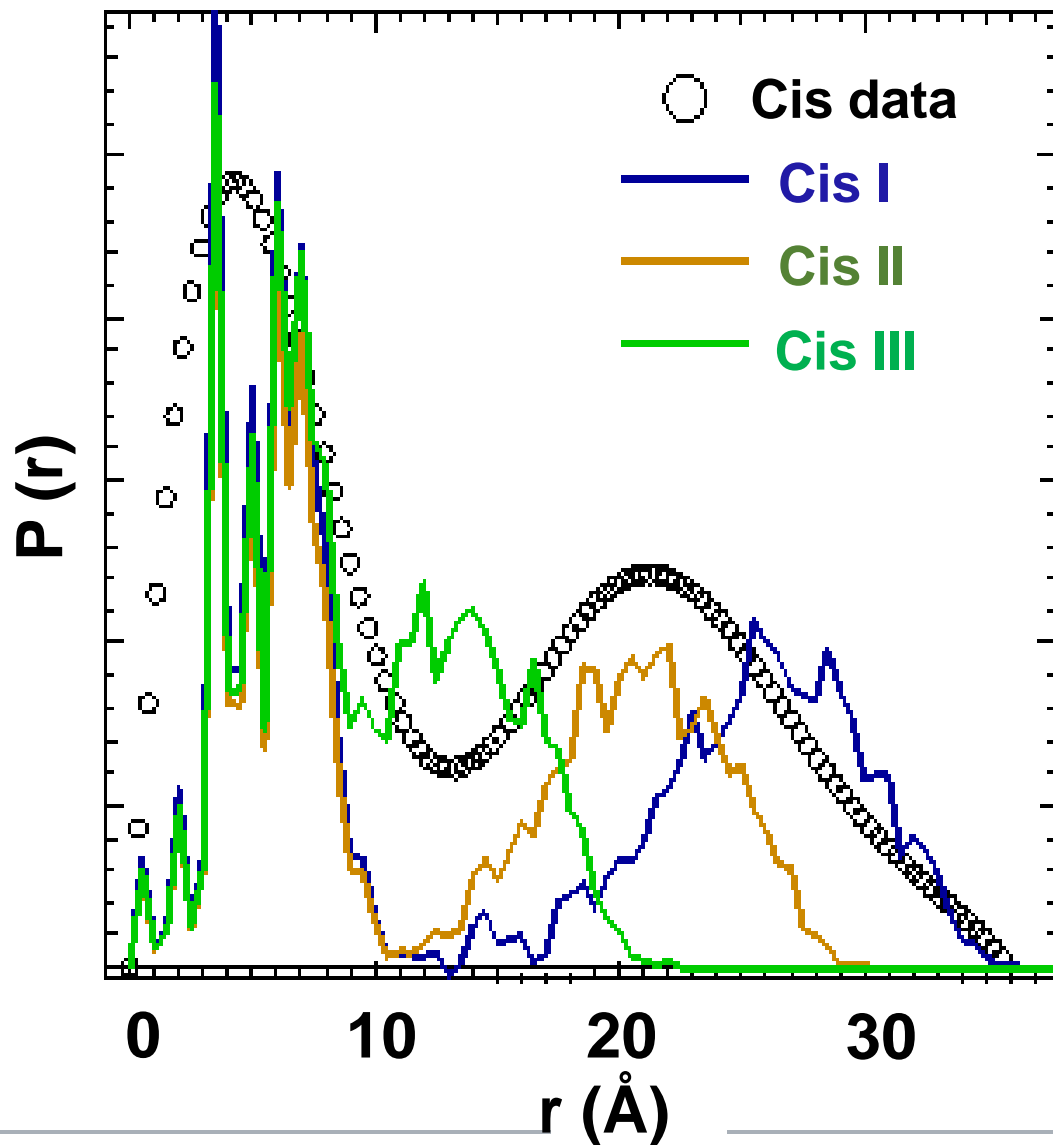
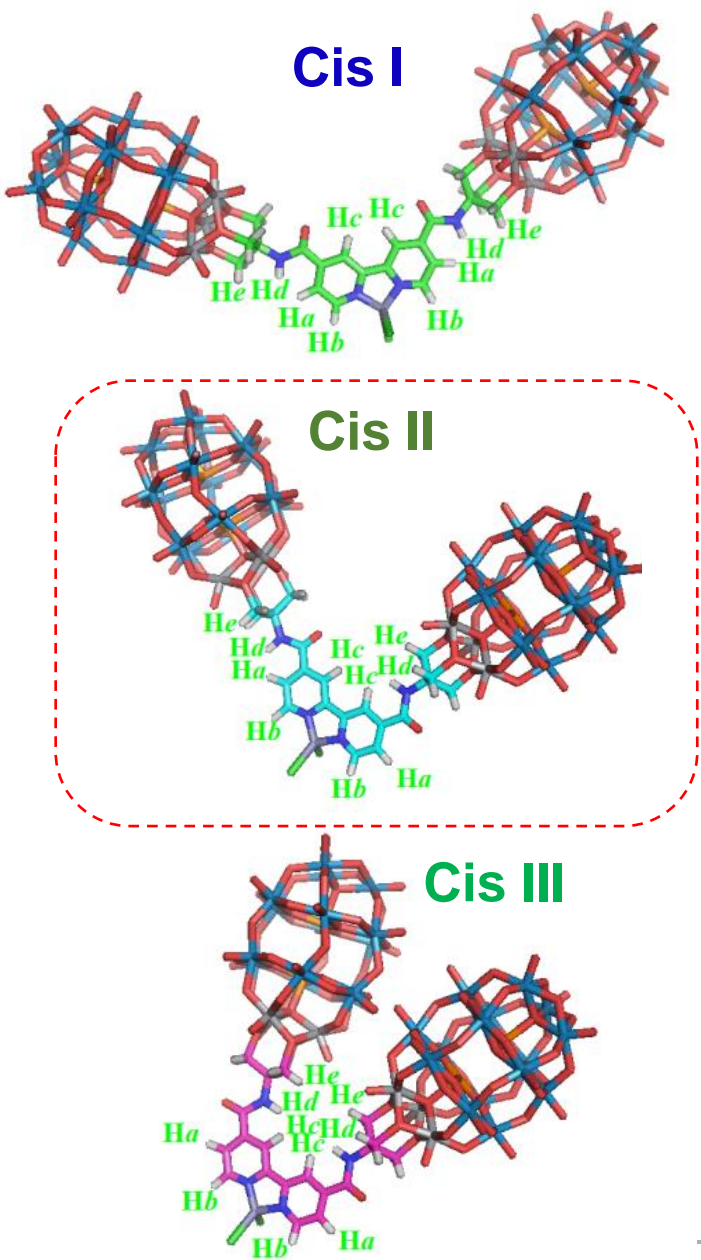


SAXS of Au-DNA Superlattice



P(r) of dimer structure

J. Am. Chem. Soc., 2013, 135 (36),
13425–13432





Lithium Ion Battery

Lithium ion batteries are powering the world.

Consumer electronics



New applications





The Nobel Prize in Chemistry 2019



Ill. Niklas Elmedhed. © Nobel Media.

John B. Goodenough

Prize share: 1/3



Ill. Niklas Elmedhed. © Nobel Media.

M. Stanley Whittingham

Prize share: 1/3

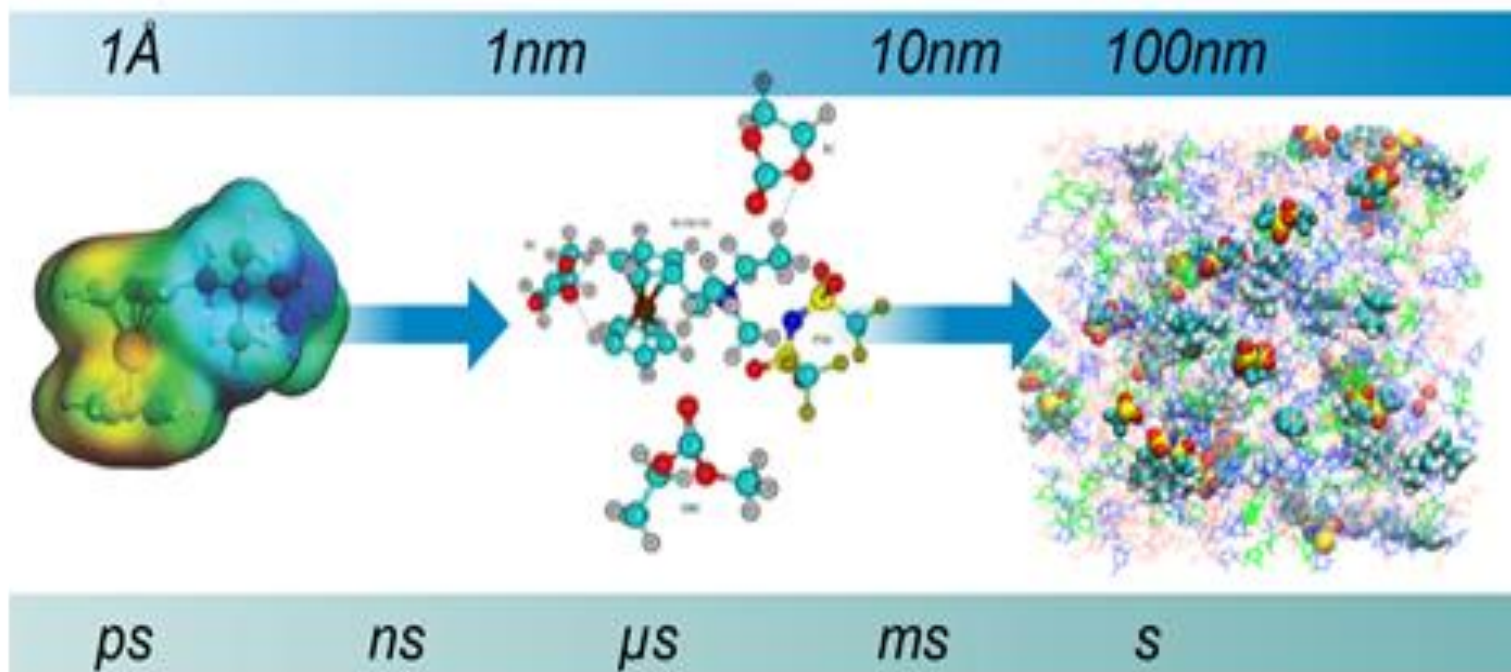


Ill. Niklas Elmedhed. © Nobel Media.

Akira Yoshino

Prize share: 1/3

Electrolytes



How to characterize the electrolyte in the solution?

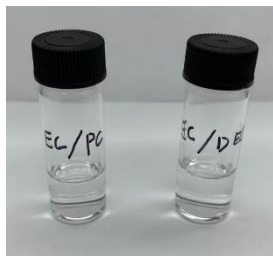
K. Qian, R. Winans, **Tao Li***. *Advanced Energy Materials*, **2021**, 2002821.

K. Qian, Y. Liu, D. Gosztola, H. Nguyen, **Tao Li***. *Energy Storage Materials*, **2021**, 41, 222-229.





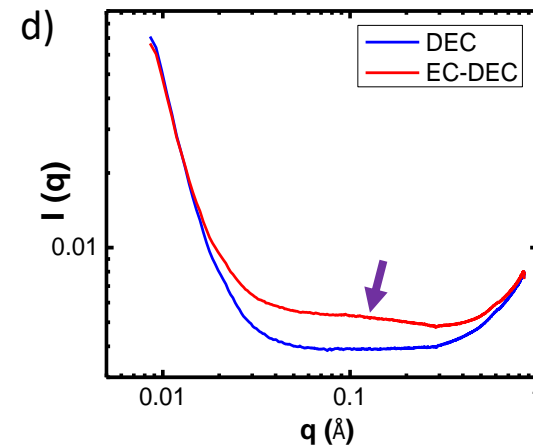
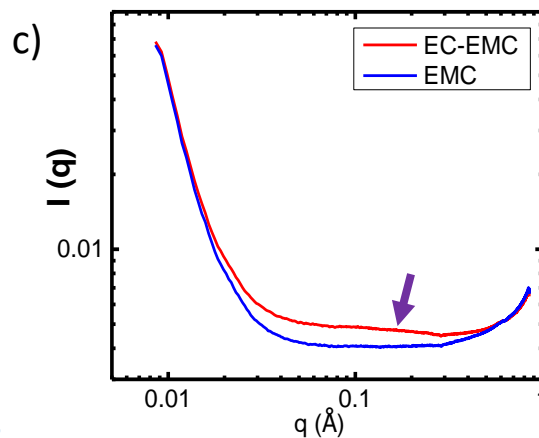
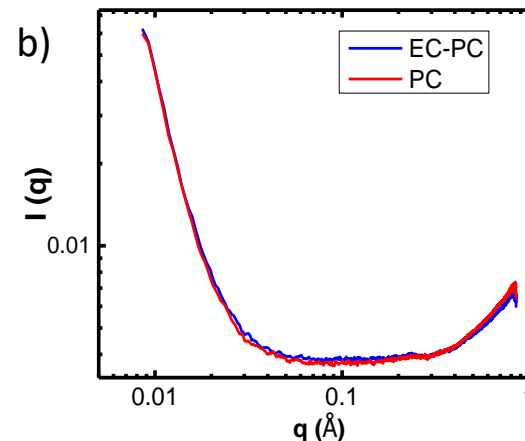
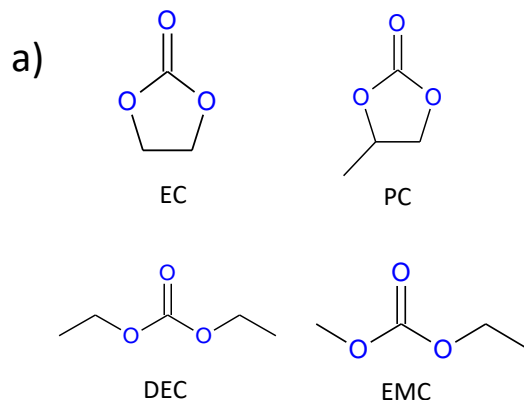
Microscopic View of the Ethylene Carbonate Based Lithium-Ion Battery Electrolyte by SAXS



Co-solvent: aggregates or single molecule?



- EC/PC form no aggregates
- EC/DEC, EC/MC, EC/DMC form ~ 2 nm aggregates
- Cryo-EM, MD simulation: structures



Z. Feng, E. Sarnello, **Tao Li**, * L. Cheng. * *J. Electrochem. Soc.* **2019**, 166, 2, A47-A49.

MD Study of LiTFSI Solvation Structure

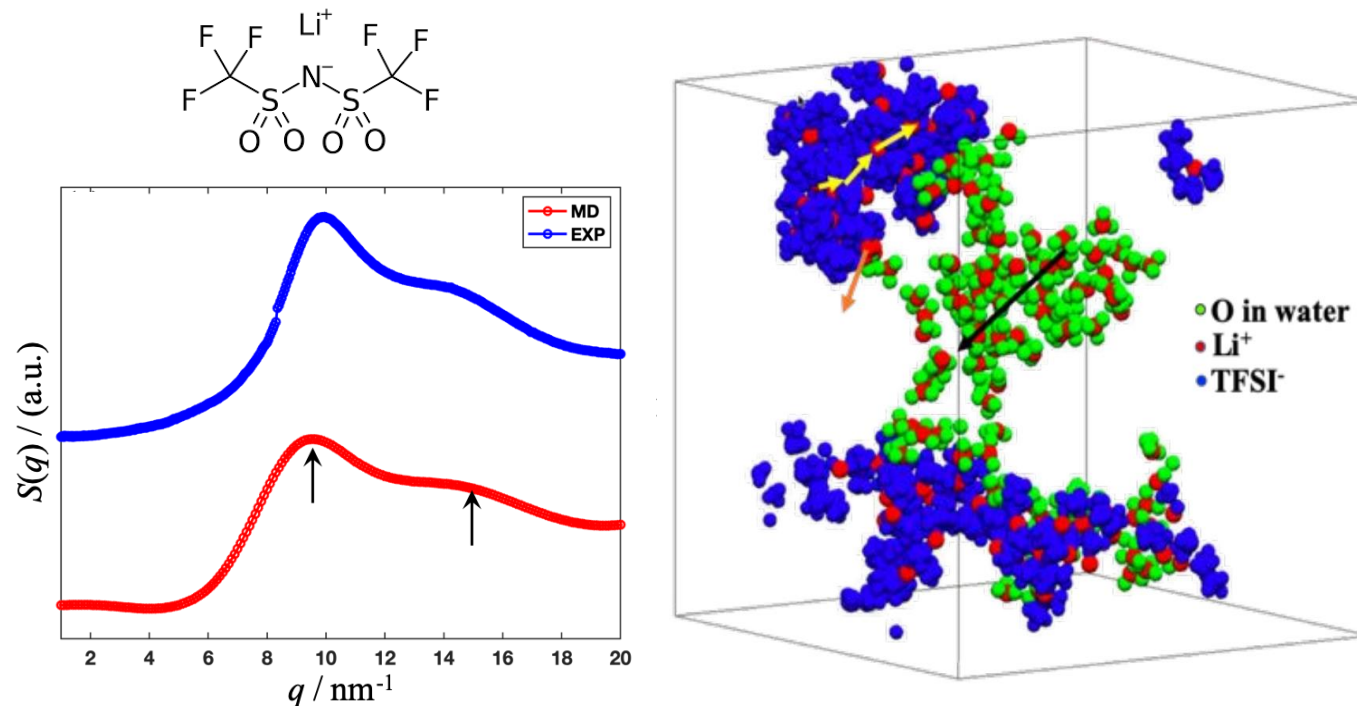
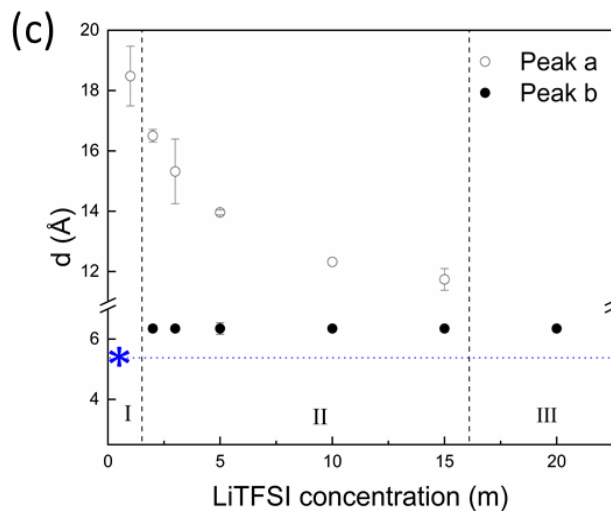
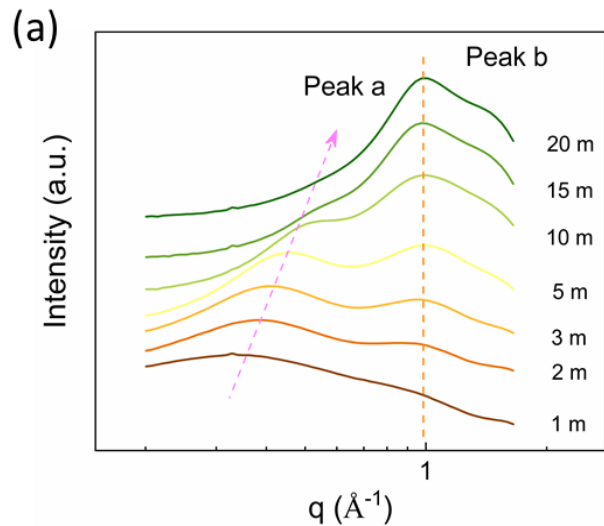


Figure. Experimental and calculated SAXS data for 20 m LiTFSI in water.

(Left) Peaks highlighted with black arrows indicates TFSI⁻ aggregates.

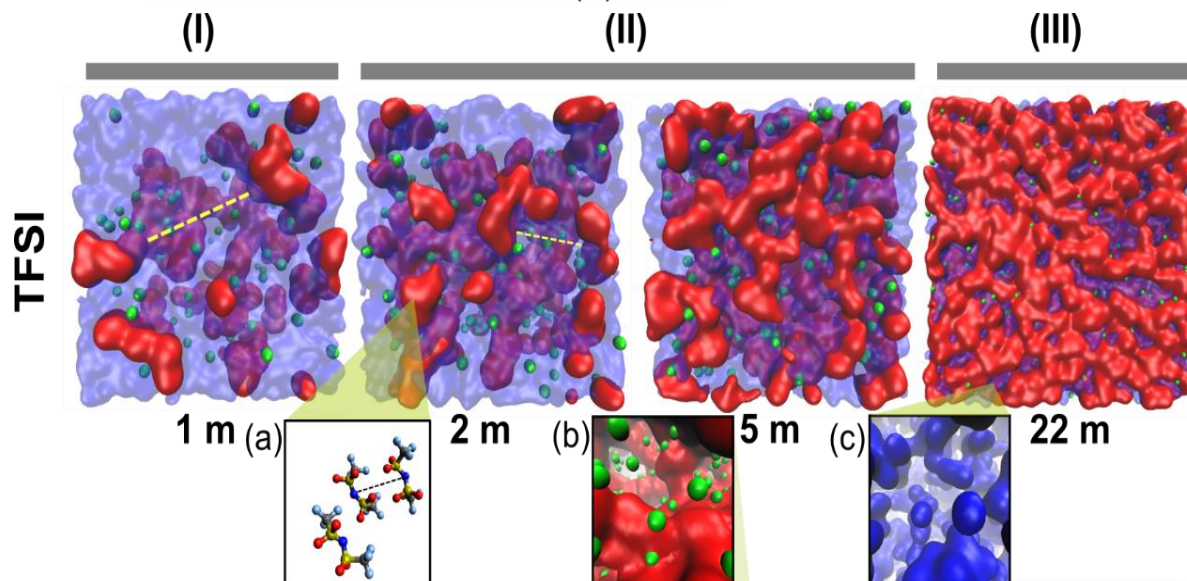
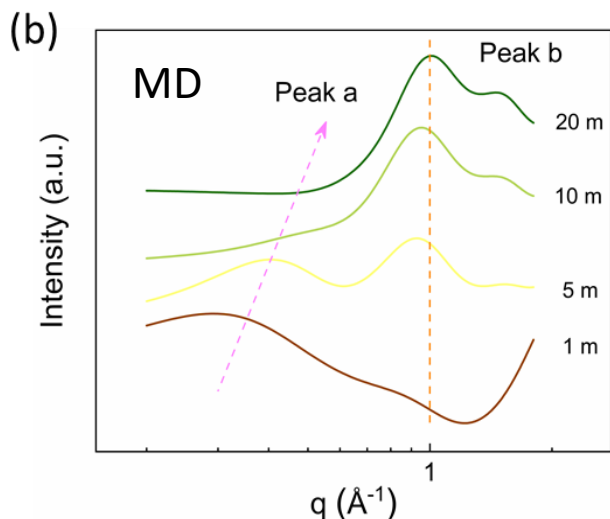
(Right) MD simulations show the heterogeneous structure of the electrolyte comprises percolating networks of ion and water domains consistent with experimental SAXS data.

SAXS/WAXS Study of LiTFSI Solvation Structure



Peak a: TFSI - solvated structure

Peak b: TFSI- network (water molecules act as bridging bond)



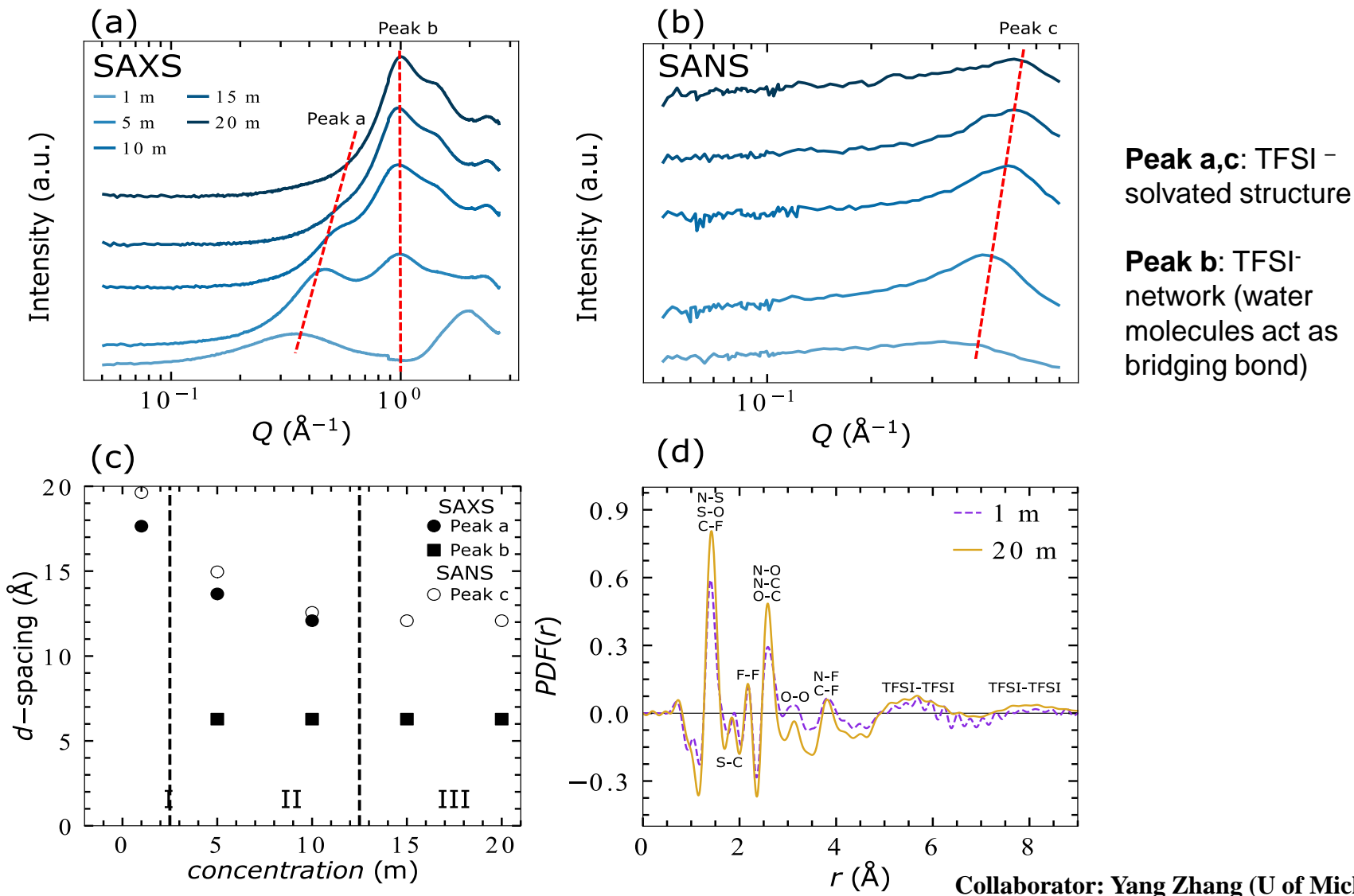
Y. Zhang, et al. *J. Phys. Chem. B* **2021**, 125, 4501–4513

X. Liu, Z. Yu, E. Sarnello, K. Qian, S. Seifert, R. E. Winans, L. Cheng*, **Tao Li.*** *Energy Materials Advances*, **2021**, 7368420.

Liu, # S.-C. Lee, # S. Seifer, R. E. Winans, L. Cheng, Y Z, * **Tao Li.*** *Energy Storage Materials* **2022**, 696-703.



SAXS and SANS Study of LiTFSI Solvation Structure

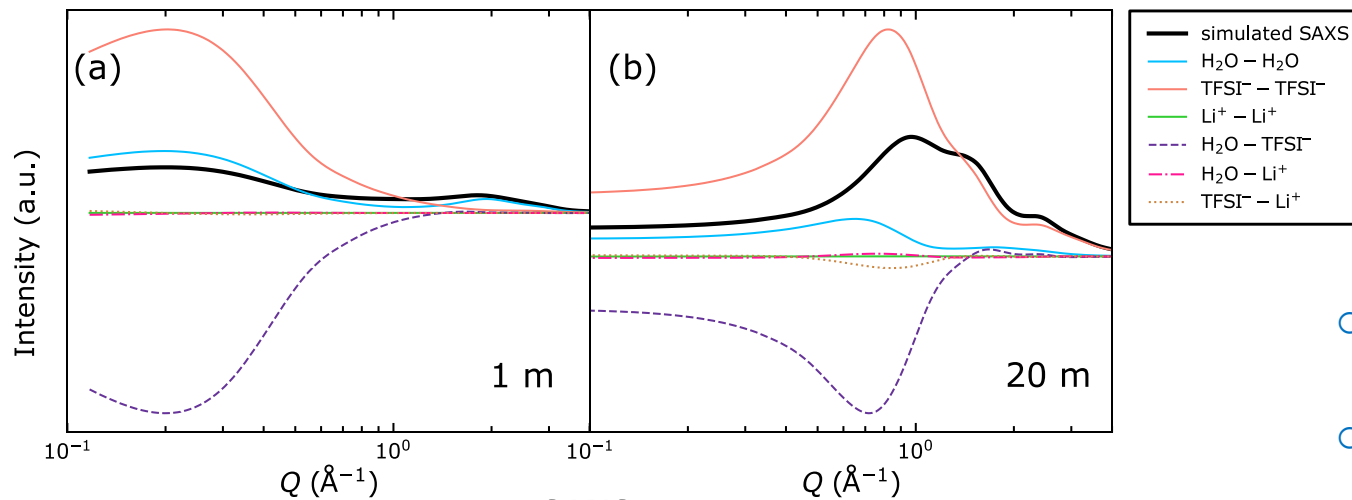


Collaborator: Yang Zhang (U of Michigan)



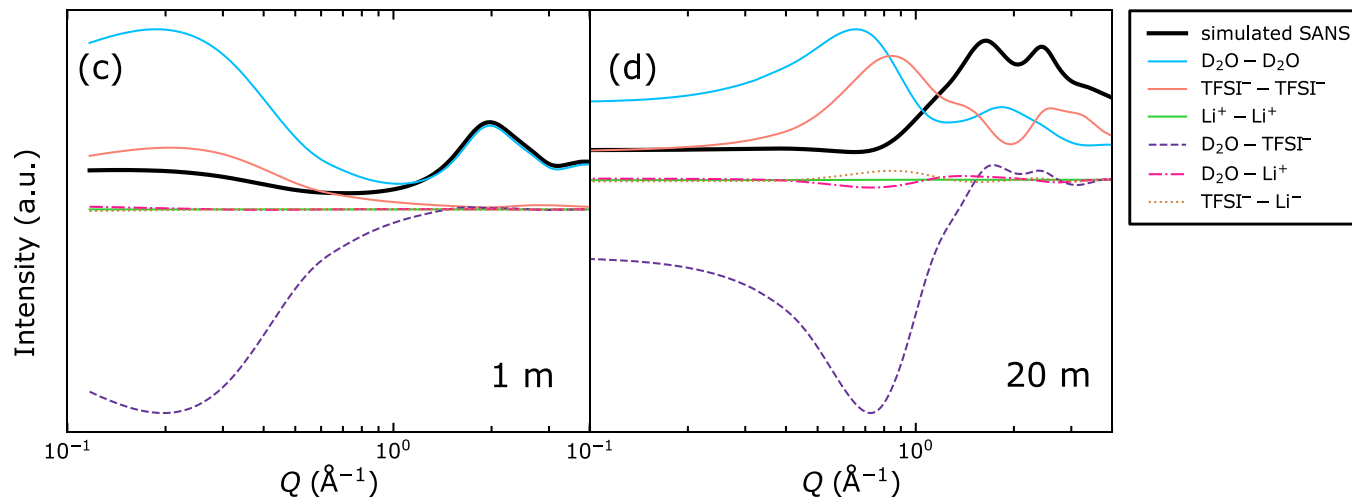
SAXS and SANS: Decomposition

SAXS



- Feature in SAXS is mainly due to TFSI-TFSI
- Feature in SANS is mainly due to $\text{D}_2\text{O} - \text{D}_2\text{O}$

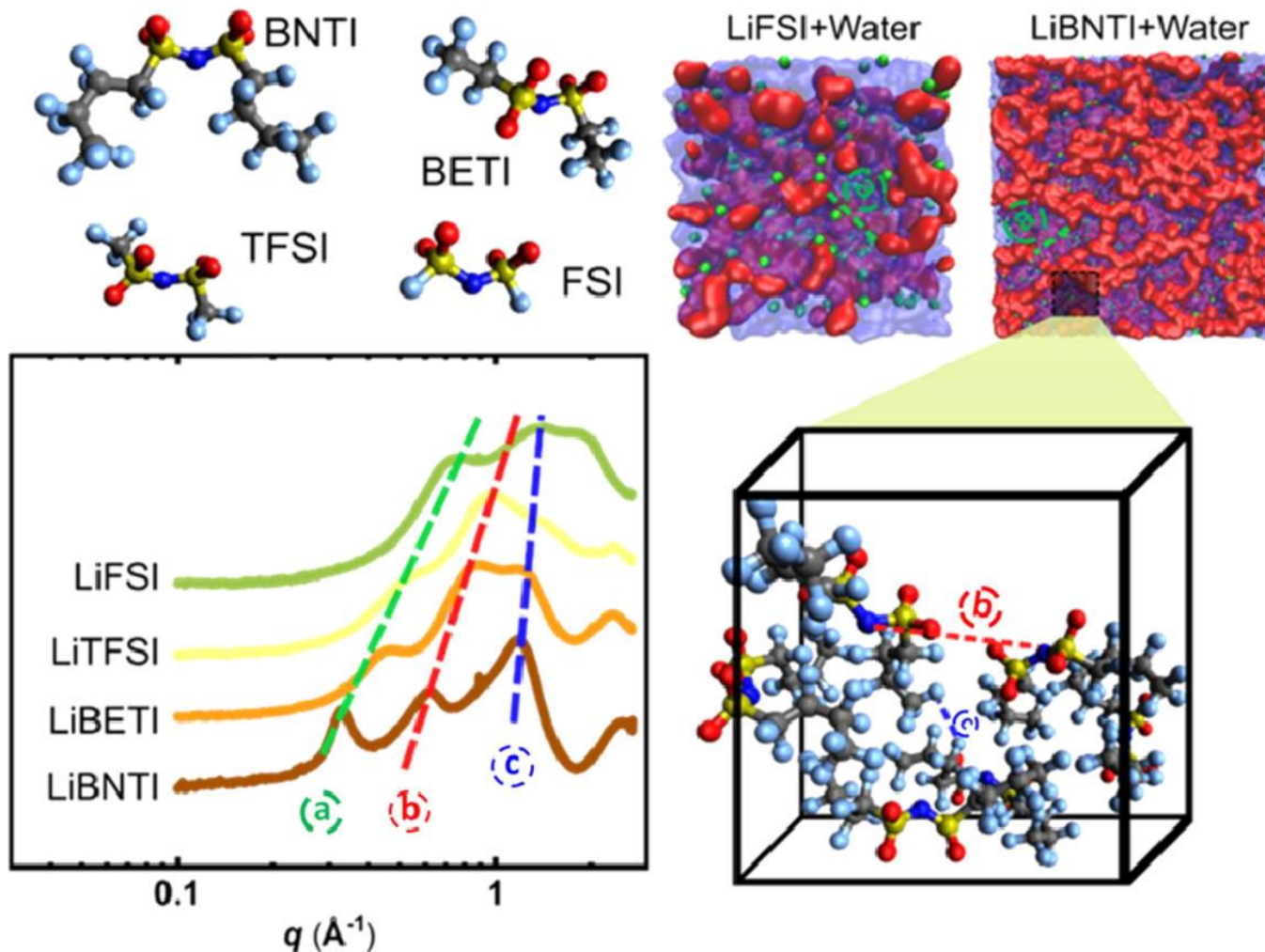
SANS



Collaborator: Yang Zhang (U of Michigan)

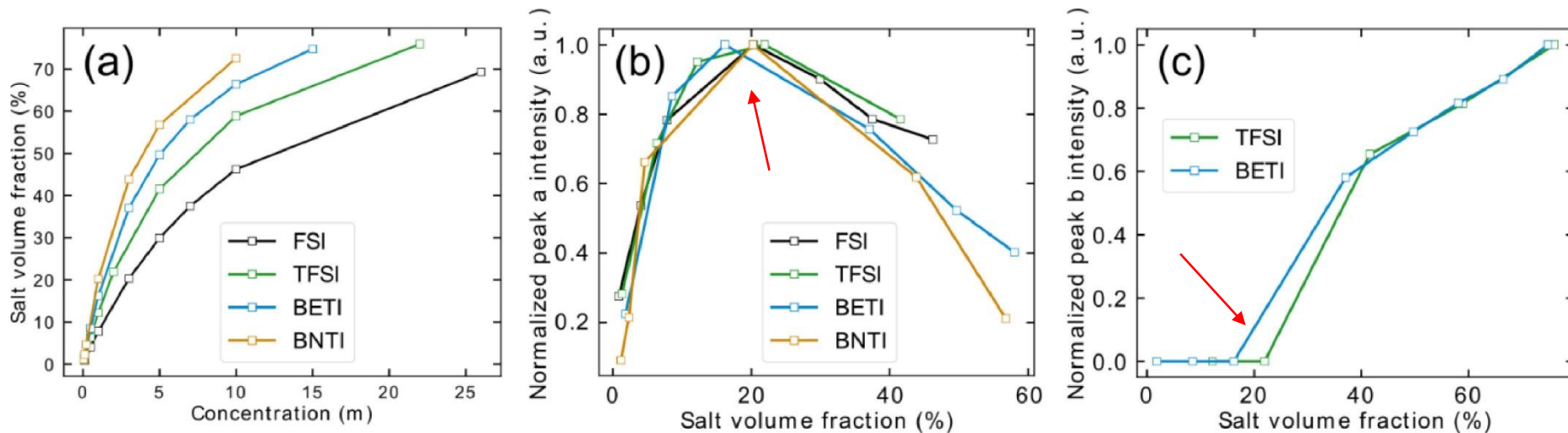
X. Liu, # S.-C. Lee, # S. Seifert, L. He, R. E. Winans, C. Do, Y. Z., * Tao Li. * *Chemistry of Materials*, 2023, 35, 2, 2088-2094.

SAXS/WAXS Study of Different Anions



X. Liu, S.-C. Lee, S. Seifert, R. E. Winans, L. Cheng, Y. Z. Tao Li. *Energy Storage Materials* 2022, 696-703.

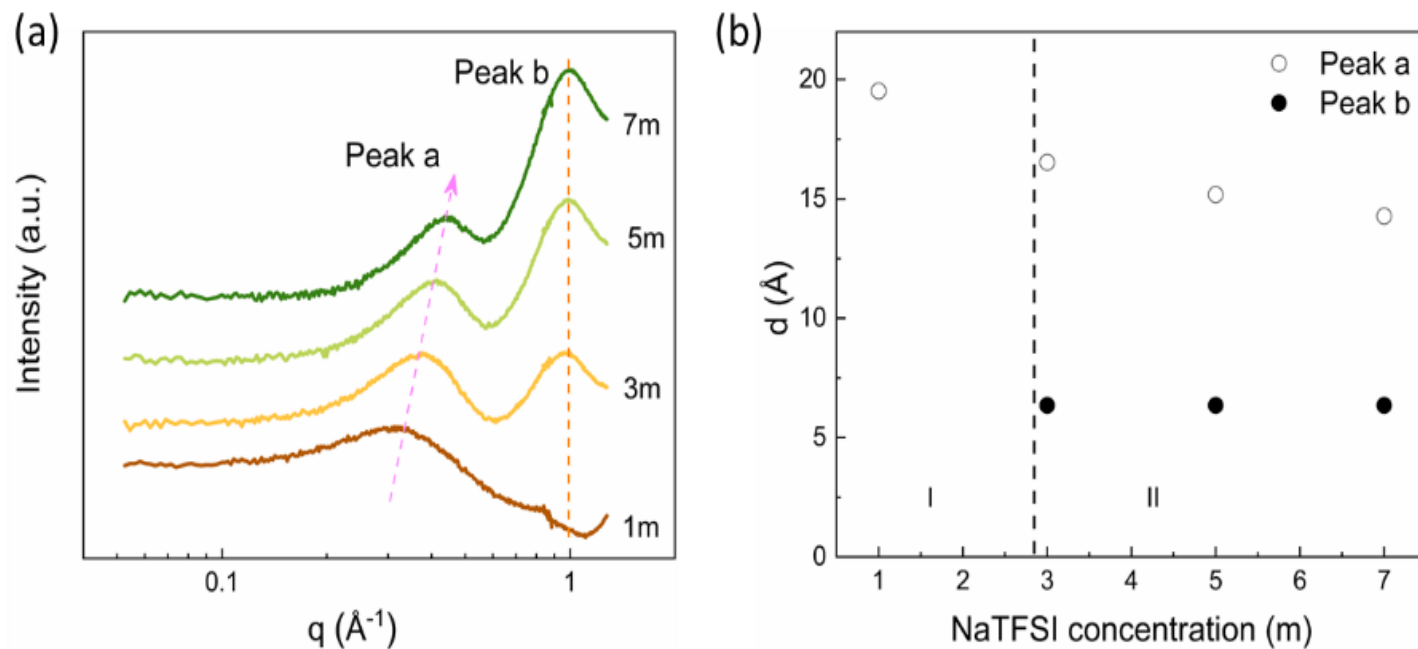
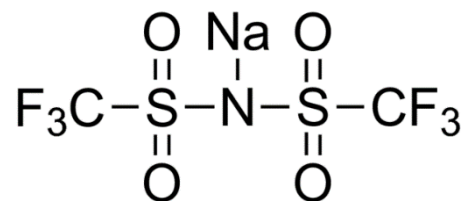
SAXS/WAXS Study of Different Anions



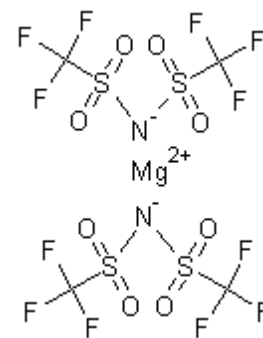
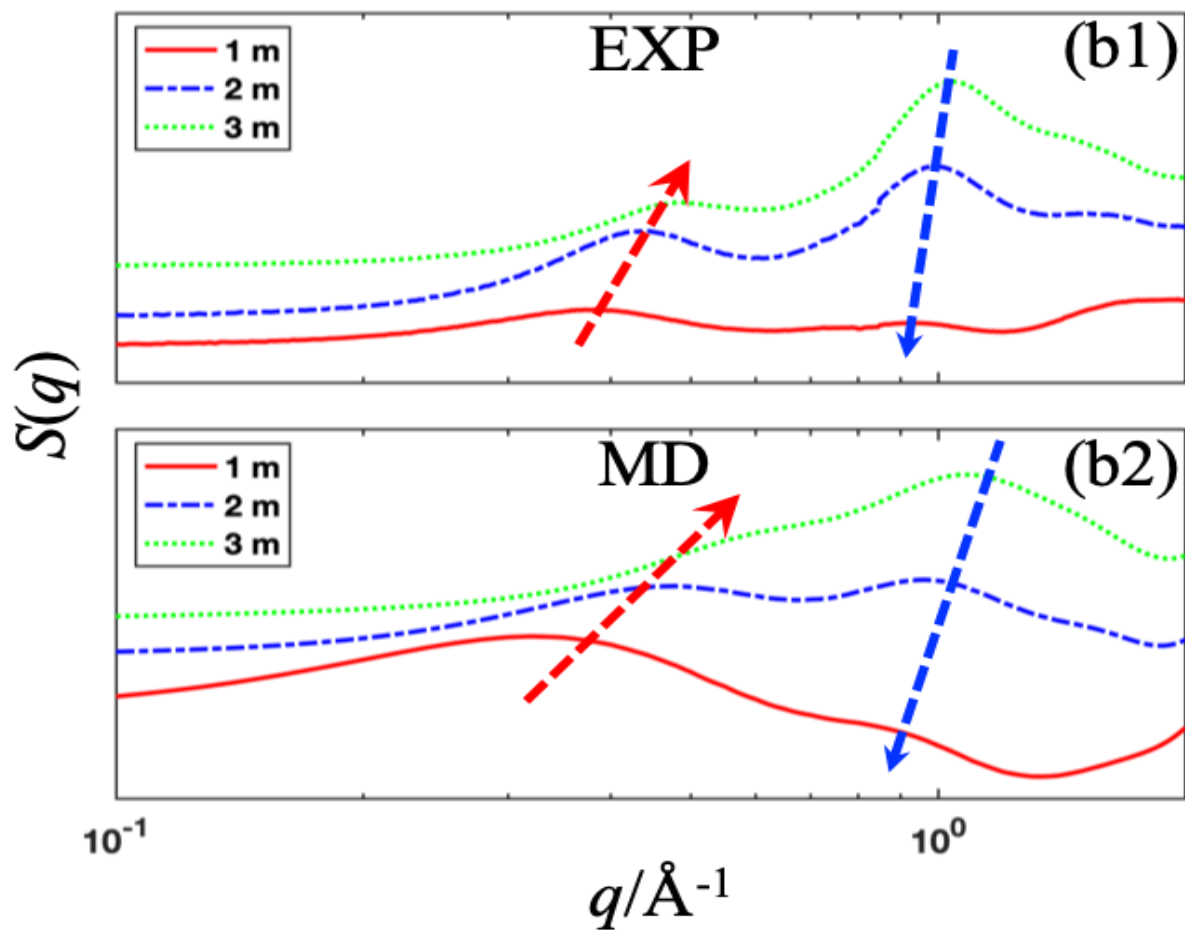
MD simulation results reveal that the formation of anion networks starts at around 20% of the salt volume fraction for all imide-based lithium salts aqueous solutions.



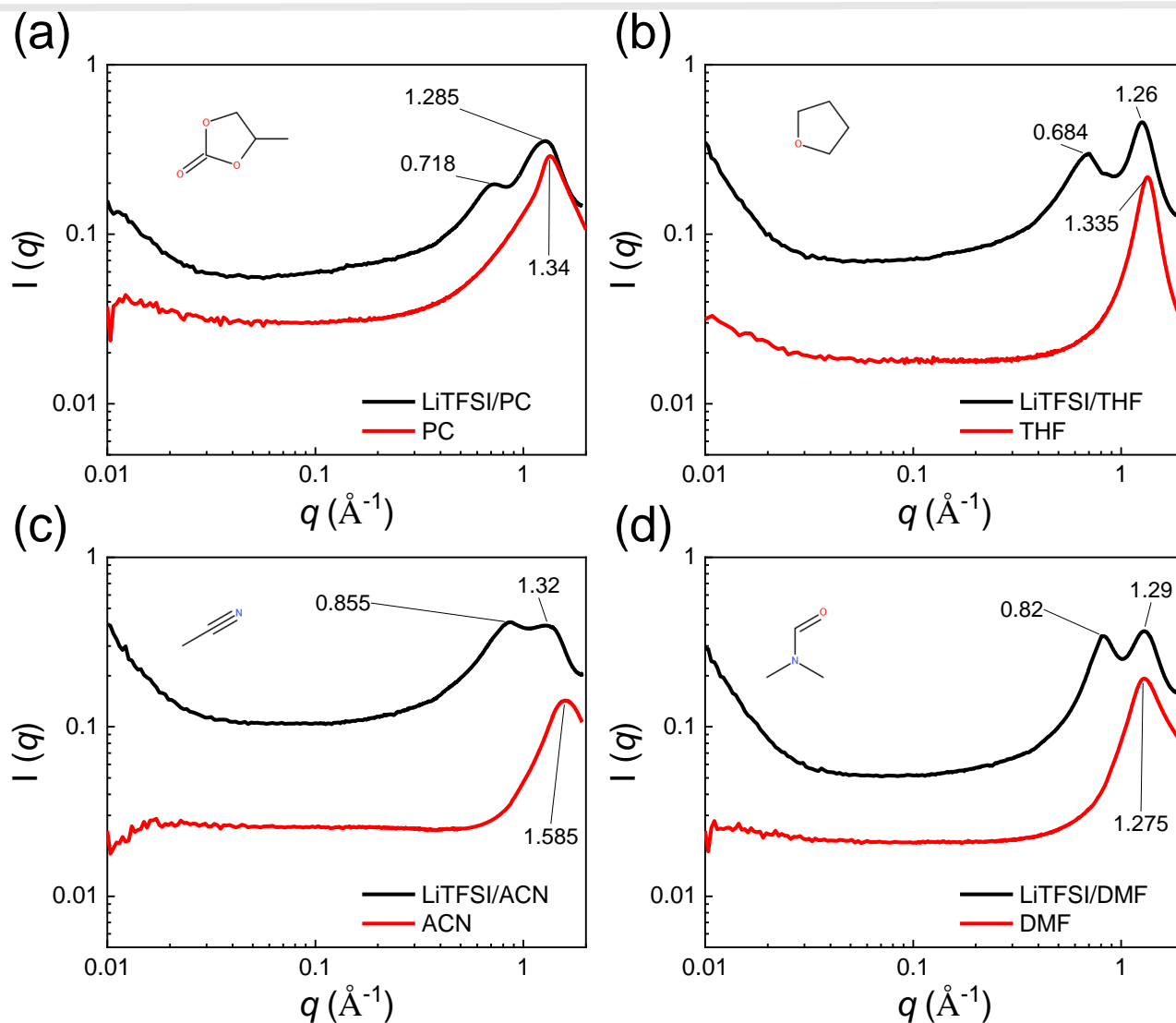
SAXS/WAXS Study of NaTFSI in Water



SAXS/WAXS Study of $\text{Mg}(\text{TFSI})_2$ in Water

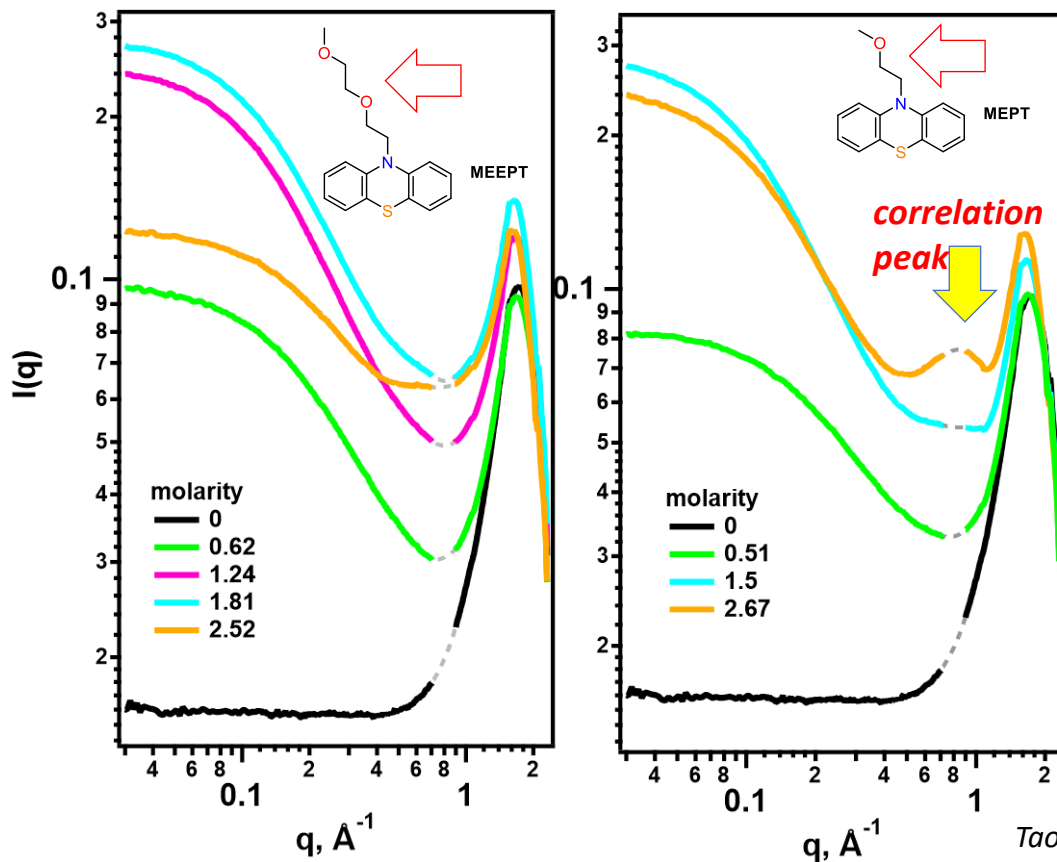


SAXS of LiTFSI in Different Solvents



Molecular Aggregation in Catholyte Redoxmer Solutions

Collaborator: Shkrob (ANL), Odom (Kentucky), Ewoldt (UIUC), Assary (ANL), L. Zhang (UIUC), Carino (ANL)



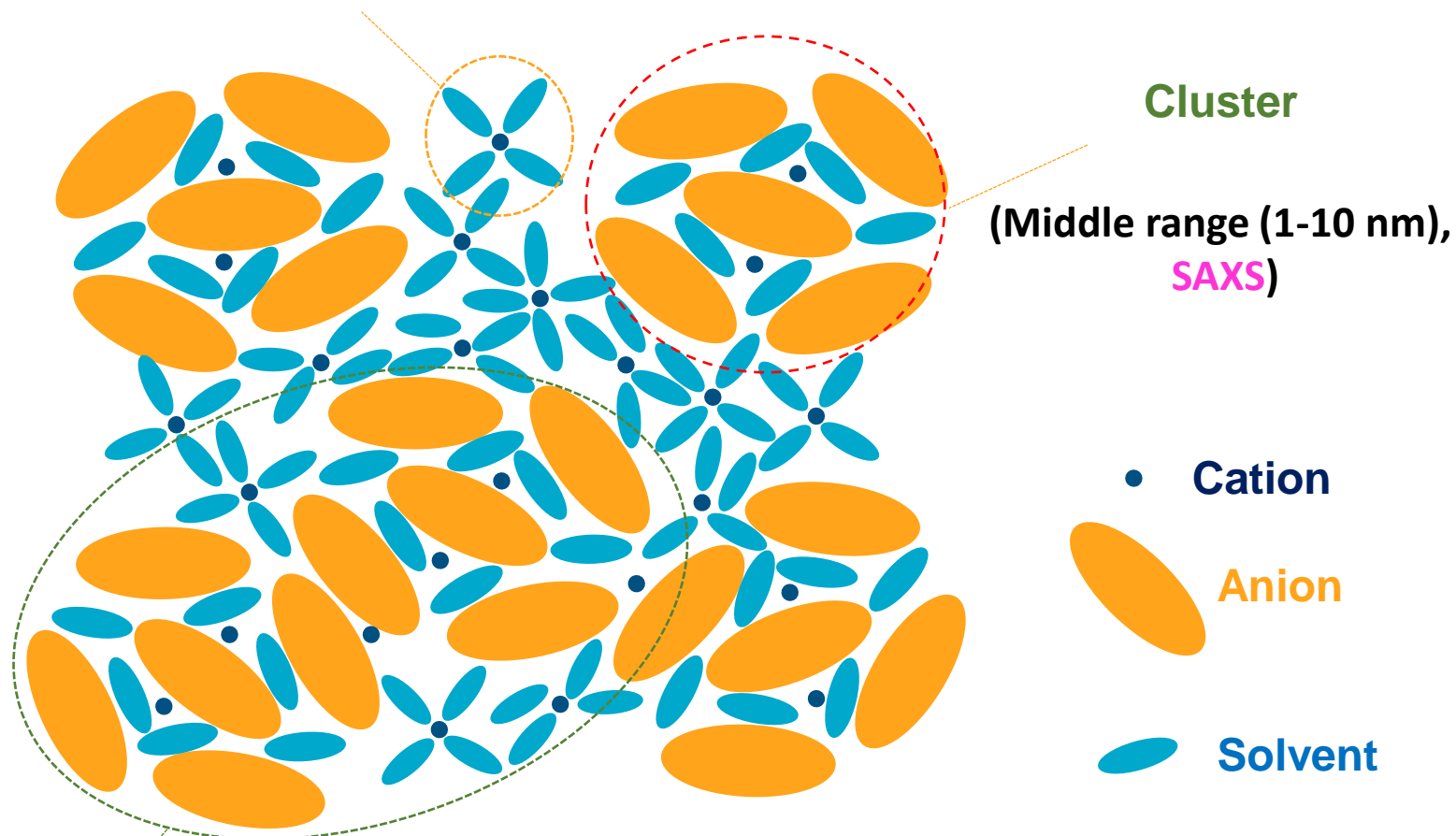
Correlation peaks indicative of redoxmer clustering are observed in concentrated solutions of some redoxmers but not the others, depending on *subtle* structural variations.

- SAXS studies demonstrate aggregation and microscopic phase separation in crowded redoxmer solutions
- Aggregation is shown to affect stability of charged redoxmers
- It strongly affects viscosity and conductivity

Tao and Erik as co-author

1. *Journal of Physical Chemistry B*, **2020**, 124, 45, 10226-10236;
2. *Journal of Physical Chemistry B*, **2020**, 124, 46, 10409-10418;
3. *Journal of Molecular Liquid*, **2021**, 334, 116533.
4. *Journal of Power Sources*, **2021**, 491, 229506.

Solvated Cation (Short-range (less than 10 Å), Raman/**WAXS**)



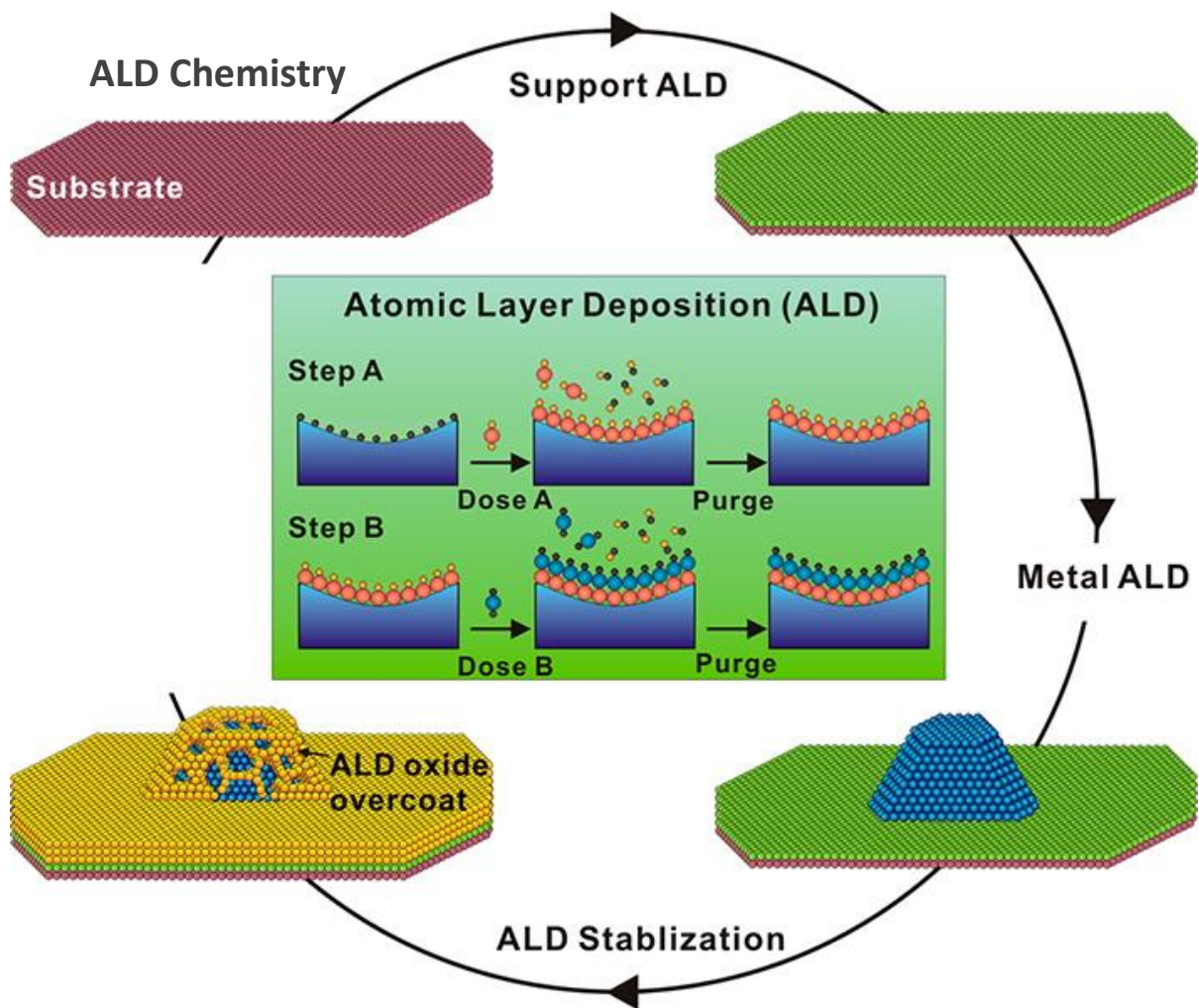
Cluster
(Middle range (1-10 nm), **SAXS**)

Aggregates (Long range (above 10 nm), **USAXS/SAXS**)

- Cation
- Anion
- Solvent

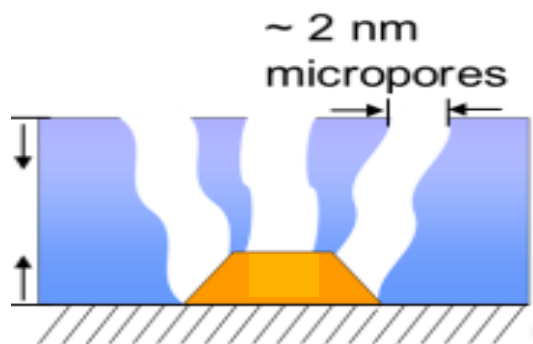
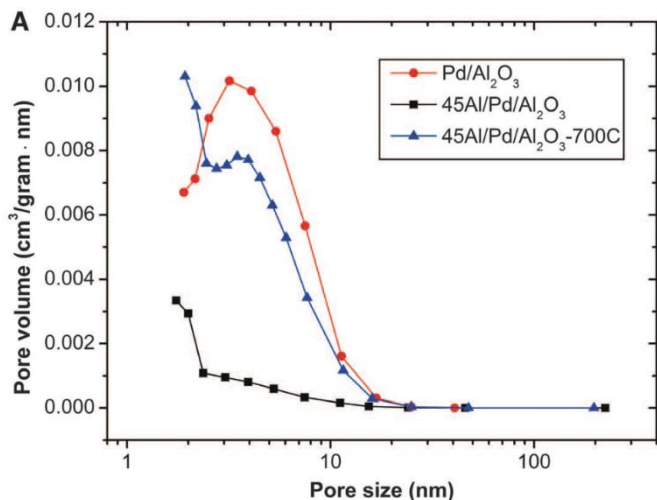


Supported Metal Catalysts by Atomic Layer Deposition



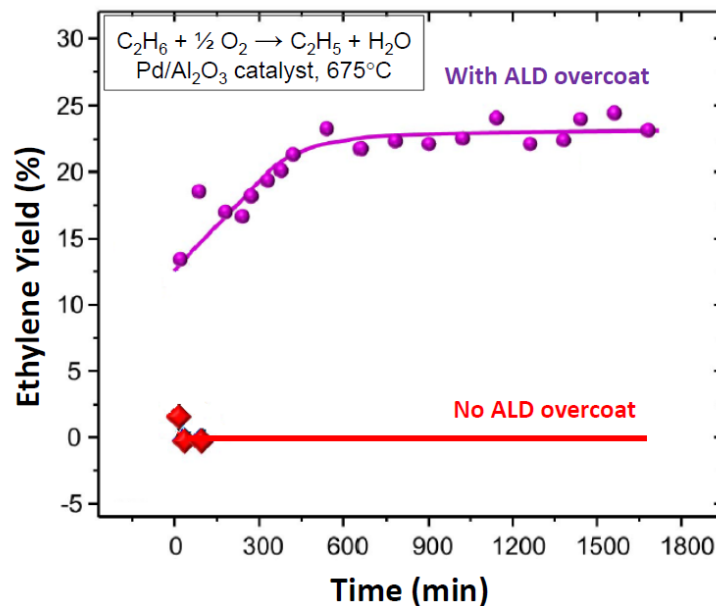
Lu, Junling; Elam, Jeffrey W.; Stair, Peter C. Accounts of Chemical Research (2013), 46(8), 1806-1815

ALD Overcoated Catalyst with Enhanced Stability



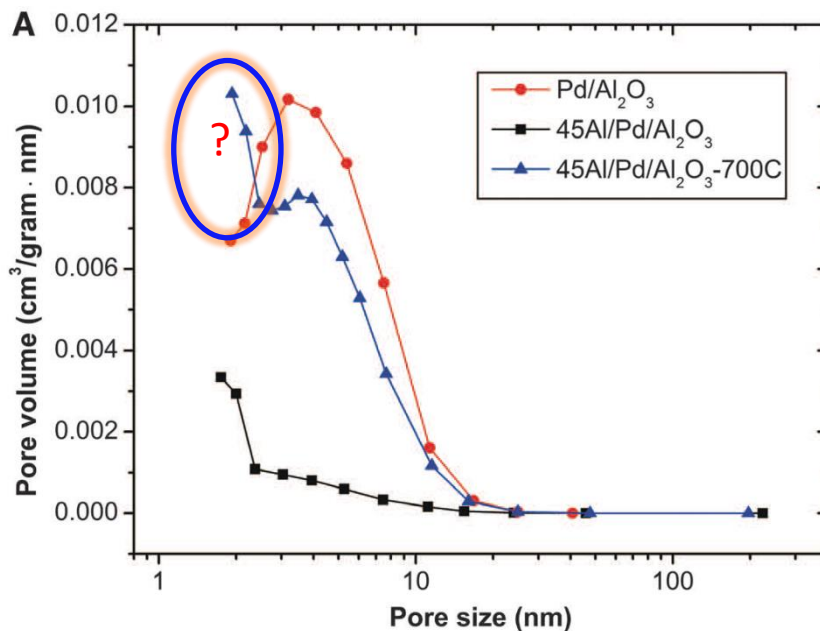
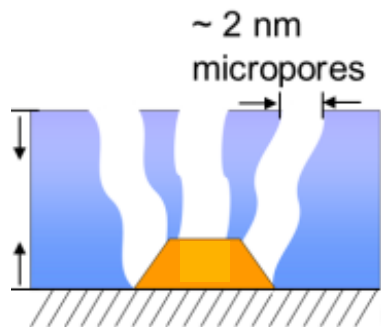
Lu J, et al. Science, 2012, 9, 1205–1208

Lu J, et al. Chem. Mater. 2012, 24, 2047–2055



- Dramatically improved yield and lifetime with ALD overcoat
- Without overcoat, coke formed in <30 min, zero yield
- With overcoat, virtually no coking, >20% yield
- Lifetime enhancement: >100x

45c Al₂O₃ Over-coating on Pd/Al₂O₃ Catalyst

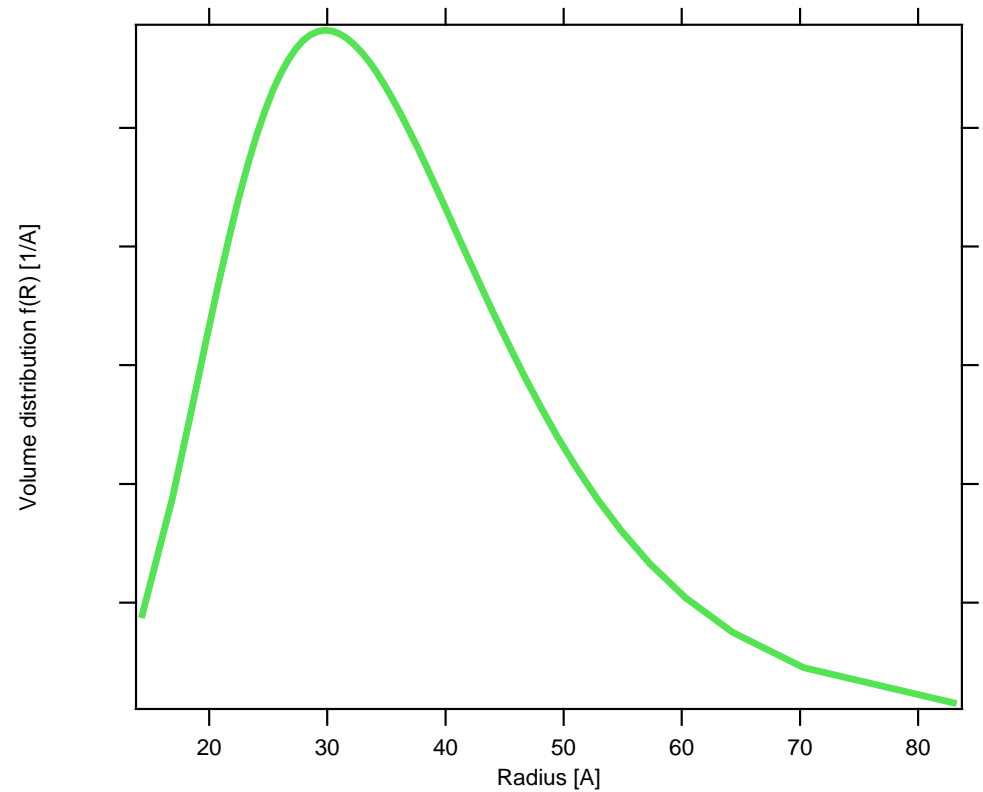
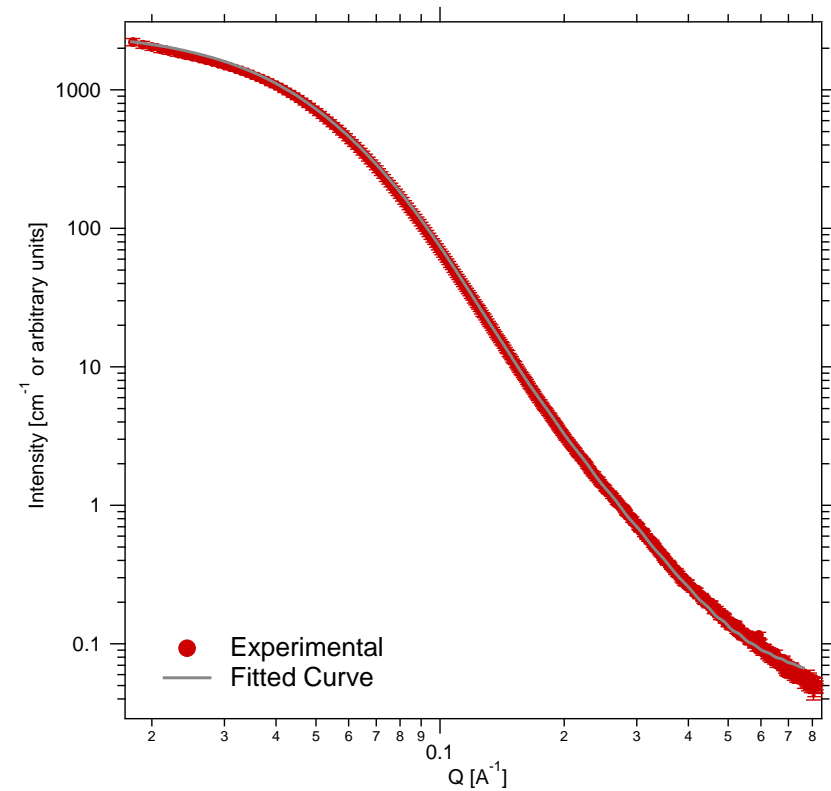


Collaborated with Peter Stair, Northwestern University

Science, 2012, 9, 1205–1208



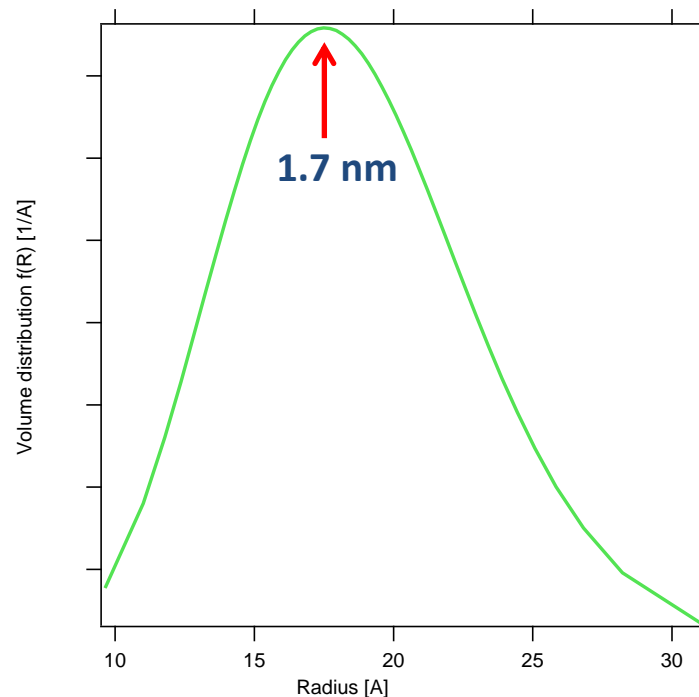
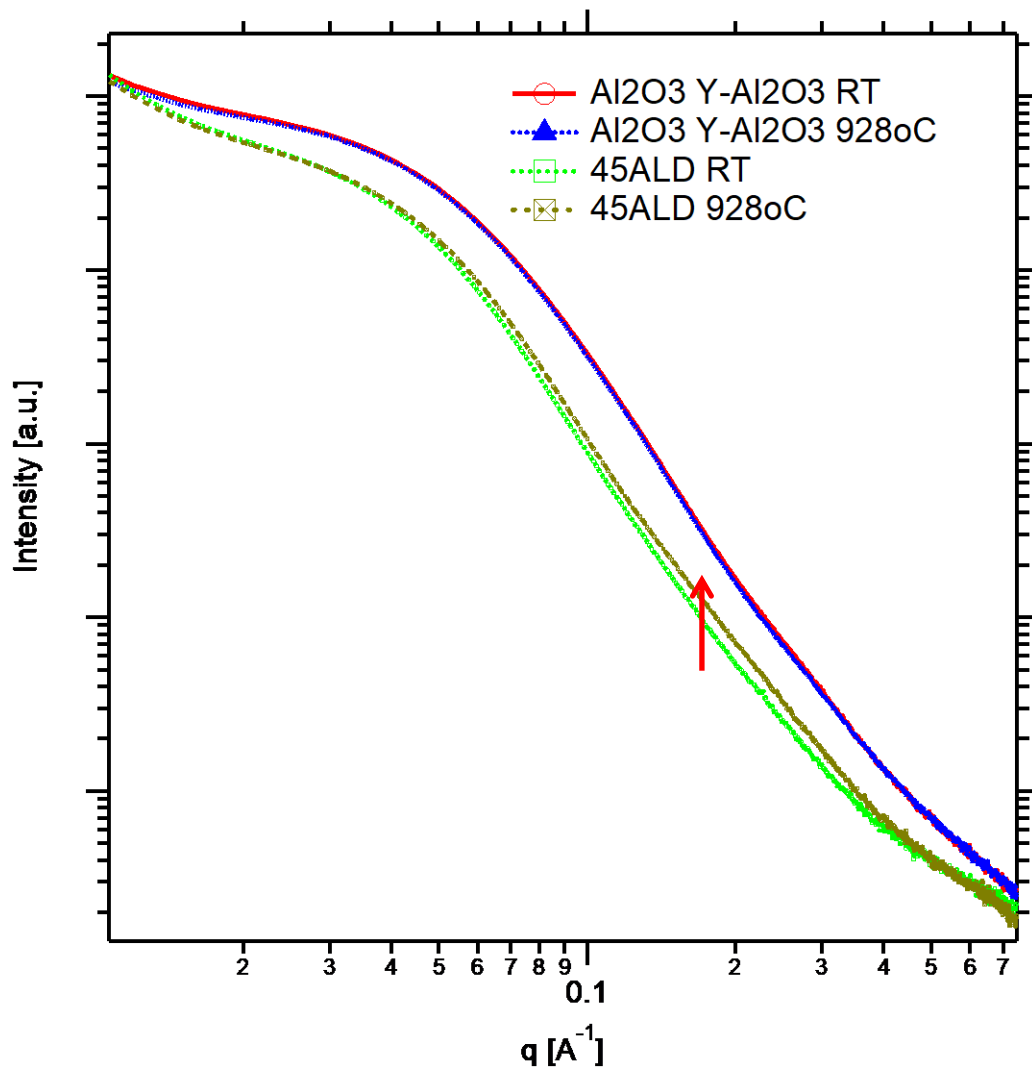
Pore Size of $\gamma\text{-Al}_2\text{O}_3$ Support from SAXS



The fitting shows that the average particle size is 6.7 nm, consistent with BJH Model data 7.1 nm.



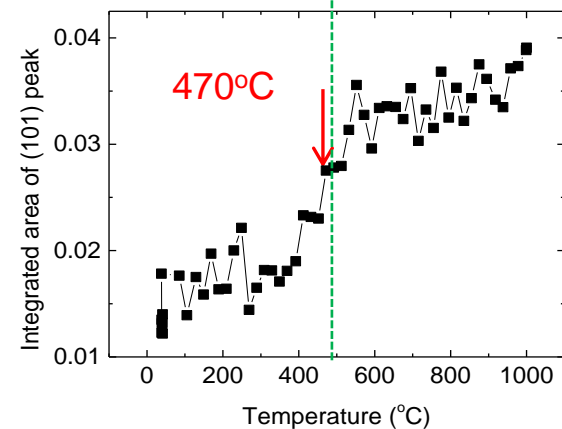
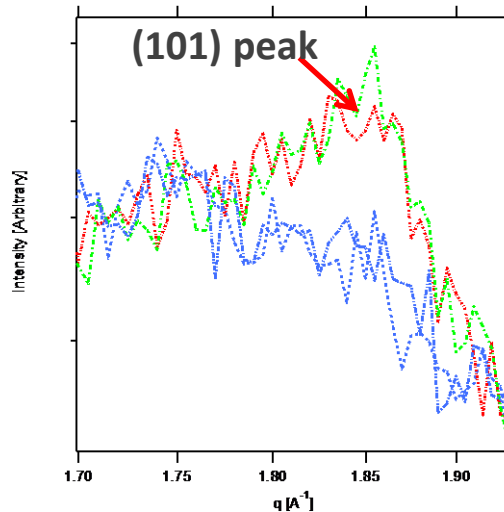
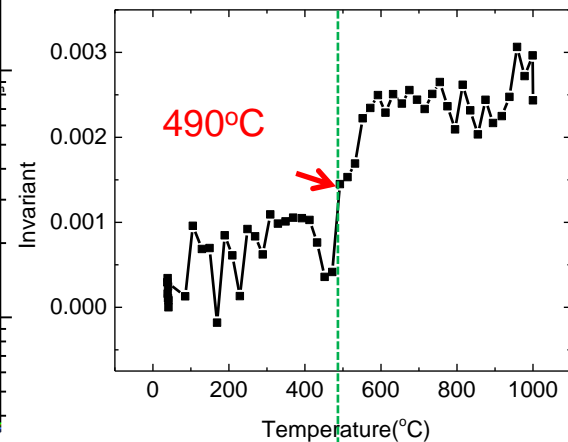
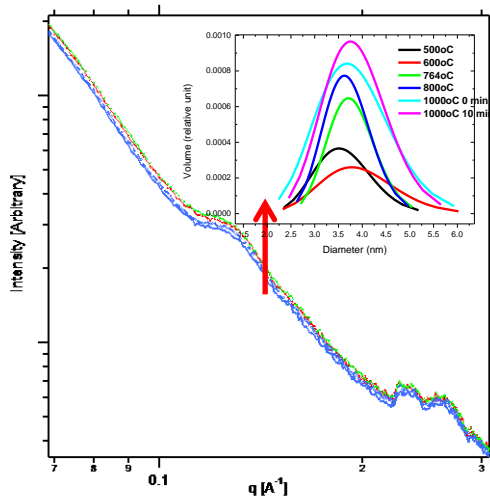
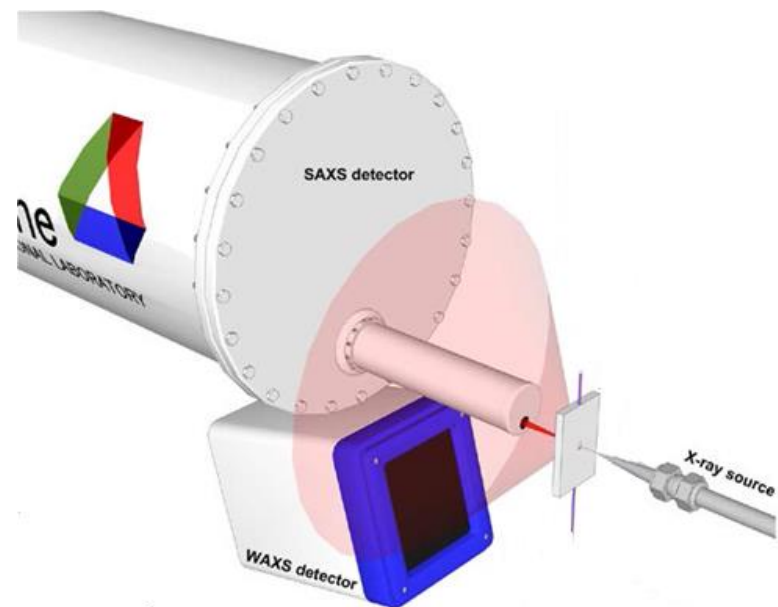
Pore Size of 45ALD/Cu/ γ -Al₂O₃ -700 Support



- **Gamma-Al₂O₃ support has no change before and after heating.**
- **For 45ALD coated samples, the intensity increases, indicating the pore.**
- **Average pore radius 1.7 nm**

Combined SAXS/WAXS of TiO₂ Overcoat

5 nm ALD TiO₂ overcoat on spherical nanodur Heat in air 20°C/min to 1000°C



■ APS (12 ID-B)

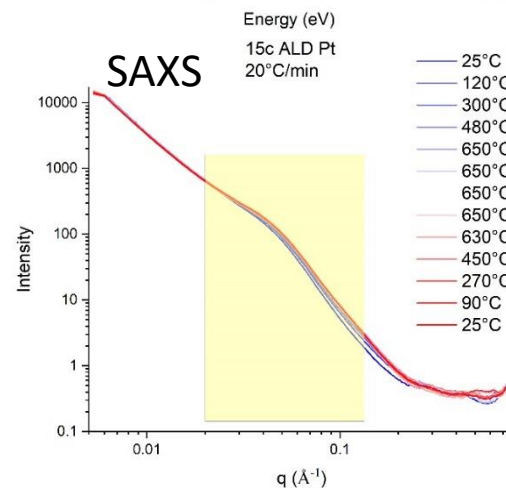
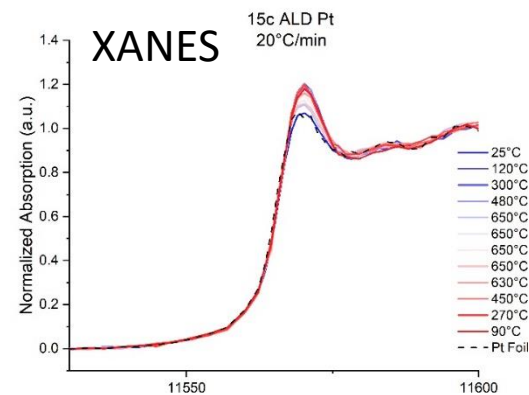
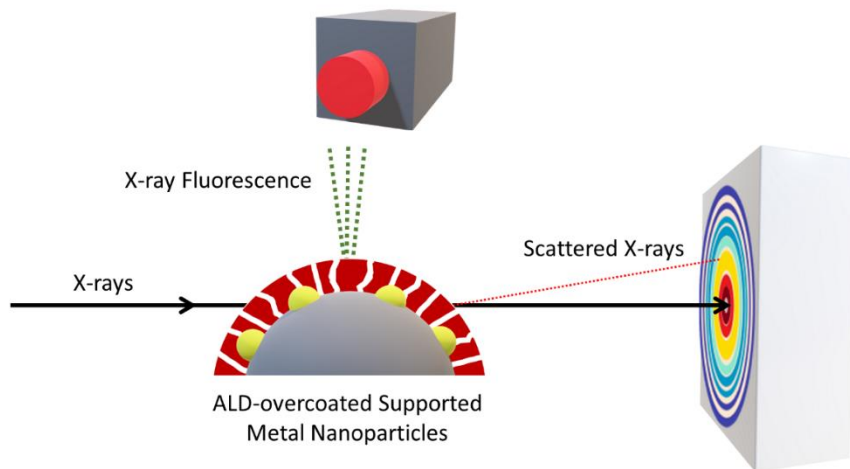
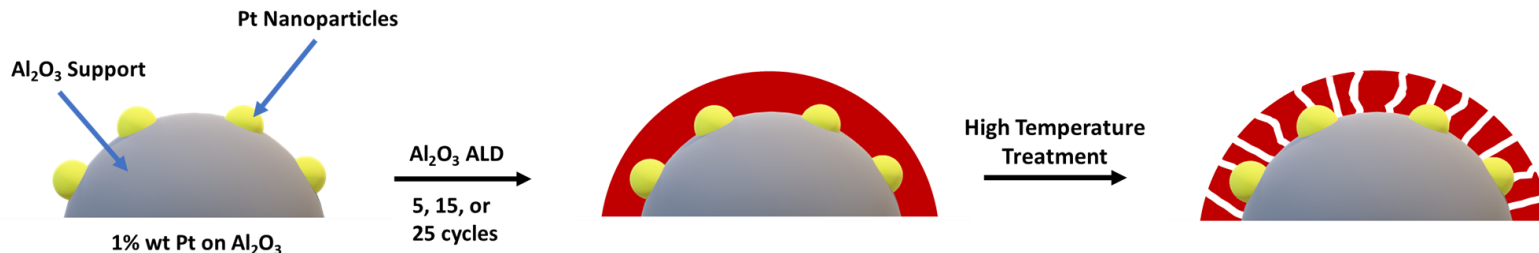
■ Linkam stage (RT to 1500 °C)

○ SAXS: pore size.

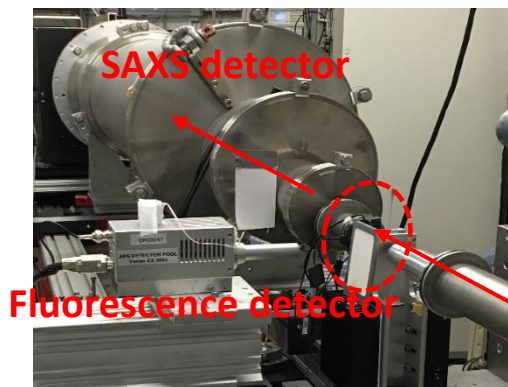
○ WAXS: crystallization.



In Situ SAXS/XAS of Al₂O₃ Overcoat on Nanodour

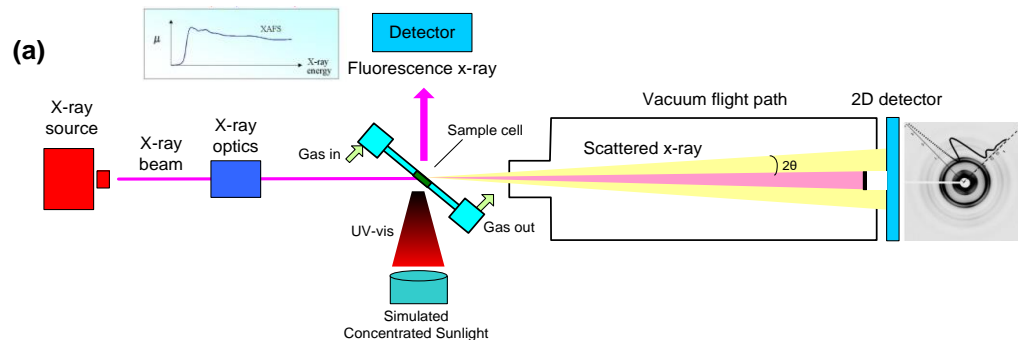
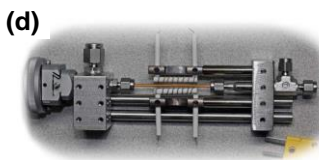
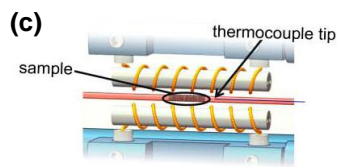
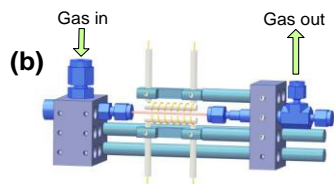
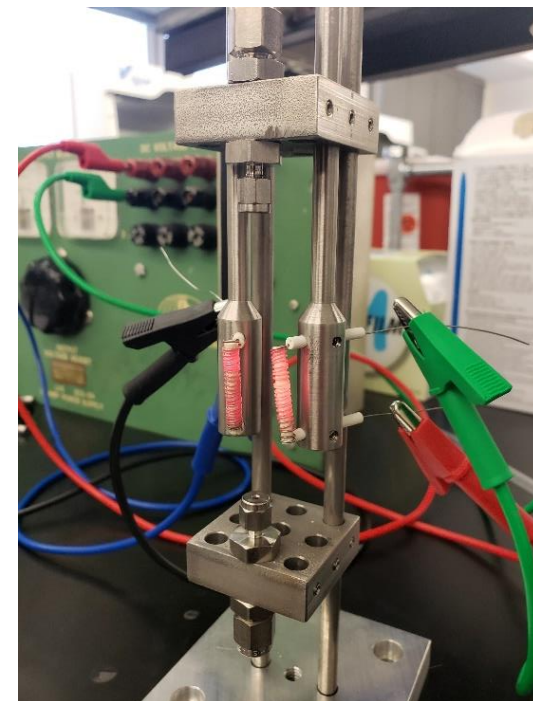
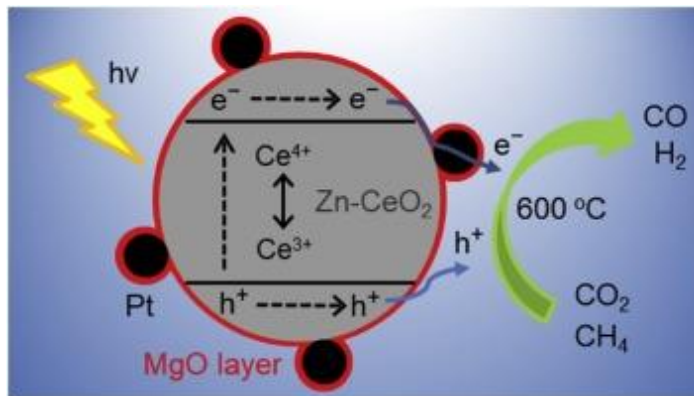


12 ID-C





Integrating Photocatalysis and Thermocatalysis to Enable Efficient Dry Reforming of Methane (DRM)



Collaborator: Ying Li (TAMU)

P. Fu, E. Sarnello, **Tao Li***, Y. Li*. *Appl. Catal. B.*, **2020**, 260, 118189.

L. Fang, Z. Feng, L. Cheng, R. Winans, **Tao Li***. *Small Methods*, **2020**, 2000315.

Z. Du, F. Pan, E. Sarnello, X. Feng, Y. Gang, **Tao Li**, Y. Li*. *Journal of Physical Chemistry C* **2021**, in press



Multiple Techniques to Observe Structure under Real Conditions

(If you can do it in the lab, we can do it on the beamline)

Five sectors provide a suite of in-situ techniques including X-ray scattering and spectroscopy at:

1-ID (high energy SAXS/WAXS (PDF))

9-ID-D(USAXS/SAXS)

9-BM, 20-BM (XAS)

10-ID (XAS)

11-BM (Hi res powder diff)

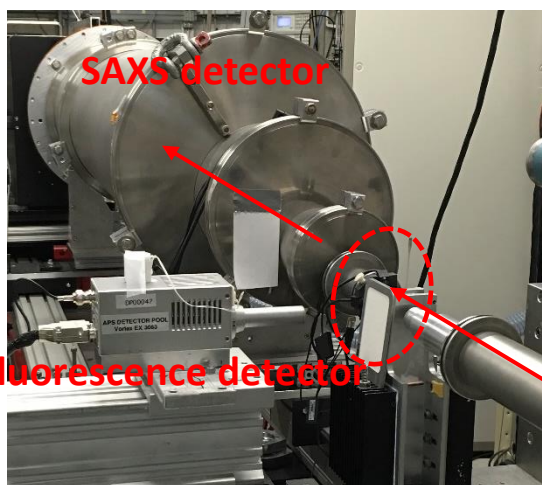
11-ID-B,C (PDF)

12-BM (XAS/SAXS)

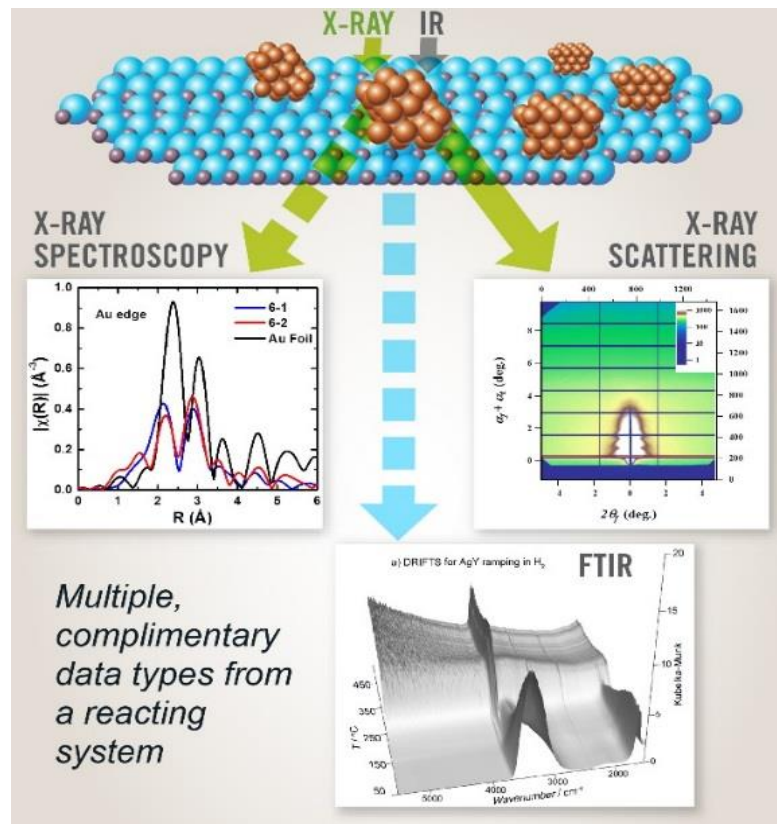
12-ID-B, C (SAXS/WAXS, and GISAXS/GIWAXS)

Also - Imaging and Microscopy (2-BM, 32-ID-BC)

Upgraded
Upgrading



12 ID-C



X-ray scattering and spectroscopy combined with FTIR and Raman to study in situ catalysis on a flat surface.

Software and useful website

<http://smallangle.org/content/Software>

- Fit2D or Nika for data reduction.
- SASfit, Irena, SasView
- Crysol
- GIXSGUI and FitGISAXS

[Irena and Nika software course](#)

[Beyond Rg Materials](#)

[Beyond Rg Bio](#)

[BioSAS: Advanced Applications](#)

THANK YOU